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1	Osmotic ve	ersus conve	entional m	nembrane l	bioreactors	integrated

with reverse osmosis for water reuse: Biological stability,

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# membrane fouling, and contaminant removal

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## **Abstract**

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This study systematically compares the performance of osmotic membrane bioreactor – reverse osmosis (OMBR-RO) and conventional membrane bioreactor - reverse osmosis (MBR-RO) for advanced wastewater treatment and water reuse. Both systems achieved effective removal of bulk organic matter and nutrients, and almost complete removal of all 31 trace organic contaminants investigated. They both could produce high quality water suitable for recycling applications. During OMBR-RO operation, salinity build-up in the bioreactor reduced the water flux and negatively impacted the system biological treatment by altering biomass characteristics and microbial community structure. In addition, the elevated salinity also increased soluble microbial products and extracellular polymeric substances in the mixed liquor, which induced fouling of the forward osmosis (FO) membrane. Nevertheless, microbial analysis indicated that salinity stress resulted in the development of halotolerant bacteria, maintaining the OMBR system biologically active. By contrast, biological performance was relatively stable throughout conventional MBR-RO operation. Compared to conventional MBR-RO, the FO process effectively prevented foulants from permeating into the draw solution, thereby significantly reducing fouling of the downstream RO membrane in OMBR-RO operation. Accumulation of organic matter, including humic- and protein-like substances, as well as inorganic salts in the MBR effluent resulted in severe RO membrane fouling in conventional MBR-RO operation. **Keywords:** Osmotic membrane bioreactor (OMBR); forward osmosis (FO); reverse osmosis

(RO); trace organic contaminants (TrOCs); membrane fouling.

## 1. Introduction

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Water scarcity due to population growth, urbanization, climate change, and environmental pollution is a vexing challenge to the sustainable development of our society (Elimelech and Phillip, 2011). This challenge calls for further efforts to develop and improve technologies that can tap into alternative water sources, such as municipal wastewater, to enhance water supply and mitigate water shortage. The ubiquitous presence of trace organic contaminants (TrOCs) in reclaimed water and wastewater-impacted water bodies remains a major obstacle to water reuse. TrOCs are emerging organic chemicals of significant concerns derived from either anthropogenic or natural activities as they present potential health risks to humans and other living organisms (Luo et al., 2014b). Membrane bioreactor (MBR) is a well-known technology for wastewater treatment and water reuse. MBR combines conventional activated sludge (CAS) treatment and a physical membrane filtration process, typically including microfiltration (MF) and ultrafiltration (UF). As an alternative to CAS treatment, MBR is more robust and versatile and can produce higher standard effluent with smaller sludge production and physical footprint (Hai et al., 2014). Some evidence has emerged that MBR could enhance the removal of TrOCs, particularly moderately biodegradable and hydrophobic compounds compared to CAS treatment (Clara et al., 2005; De Wever et al., 2007). However, some hydrophilic TrOCs are still poorly removed by MBR due to their resistance to biodegradation and low adsorption onto sludge (Tadkaew et al., 2011; Nguyen et al., 2013; Wijekoon et al., 2013). Thus, further treatment by nanofiltration (NF) or reverse osmosis (RO) is usually required to produce high quality water for reuse (Gerrity et al., 2013). The NF/RO process can complement well MBR to achieve effective removal of various TrOCs (Alturki et al., 2010; Nguyen et al., 2013).

Recent progress in water reuse has led to the emergence of a new variation of MBR, namely, osmotic membrane bioreactor (OMBR) (Achilli et al., 2009; Cornelissen et al., 2011; Chen et al., 2014; Nguyen et al., 2016). During OMBR operation, treated water is extracted from the mixed liquor into a highly concentrated draw solution by the forward osmosis (FO) process. By employing a selective, semi-permeable FO membrane, TrOCs can be retained in the bioreactor and thus increase their biodegradation during OMBR operation (Alturki et al., 2012; Holloway et al., 2014). Moreover, FO has a lower fouling propensity, and when fouling occurs, it is readily reversible compared to pressure-driven membrane processes (Mi and Elimelech, 2010; Kim et al., 2014; Luo et al., 2015a; Xie et al., 2015). OMBR can be used as a stand-alone process or coupled with a desalination process, such as RO to form an OMBR-RO hybrid system, for draw solution recovery and recycling water production (Luo et al., 2014a). In the latter configuration, the desalination process may provide an additional barrier to further purify the product water. For instance, Holloway et al. (2014) demonstrated that 15 of 20 TrOCs detected in municipal wastewater were removed to below detection limit by OMBR, and compounds that passed through the FO membrane were effectively retained by the subsequent RO process. An effective contaminant removal by OMBR-RO and its potential for advanced wastewater treatment and water reuse were also subsequently highlighted by Luo et al. (2016b). It is noteworthy that an MF or UF membrane was coupled with OMBR in these two studies to control salinity build-up, which is an inherent issue associated with OMBR due to the high salt rejection by the FO membrane and, more importantly, the reverse draw solute flux. OMBR-RO can offer a range of potential benefits over conventional MBR-RO systems for advanced wastewater treatment and water reuse. Cornelissen et al. (2011) reported that OMBR-RO could reduce the capital cost of wastewater reuse by 5 - 25% compared to conventional MBR-RO using the UF membrane. Cost saving achieved by OMBR-RO

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depends on the assumption that the cost and water permeability of the FO membrane are comparable to those of the UF membrane. Given the low fouling property of FO compared to UF, there can be also a reduction in operational cost related to membrane cleaning and replacement. Cornelissen et al. (2011) also assumed that the two hybrid systems had the same treatment performance, which was probably conservative as the FO membrane can produce higher quality permeate than the UF membrane. The high quality FO permeate would alleviate membrane fouling in the downstream RO process, which is a major issue for costeffective application of conventional OMBR-RO for water reuse (Farias et al., 2014; Al Ashhab et al., 2014). Thus, additional cost saving for OMBR-RO can potentially be derived from a more stable water production from the downstream RO unit with less frequent cleaning and longer service time in comparison with conventional MBR-RO. Of a particular note, to date, no study has directly compared the performance of OMBR-RO and conventional MBR-RO for water reuse. This study aims to compare the performance of OMBR-RO with conventional MBR-RO in terms of biological stability, contaminant removal, and membrane fouling. Similar operating parameters were applied to both bioreactors for a systematic comparison. Water production and salinity build-up during OMBR-RO operation were evaluated. High-throughput sequencing technique was applied to elucidate the effect of salinity build-up on microbial community structure during OMBR-RO operation compared to that in conventional MBR-RO. Fate and transport behaviours of bulk organic matter, nutrients, and TrOCs in these two hybrid systems were systematically examined. In addition, the fouling behaviour of the RO membrane in both systems was also delineated and compared.

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## 2. Materials and methods

- 106 2.1 Synthetic wastewater and trace organic contaminants
- 107 A synthetic wastewater was used in this study. The synthetic wastewater was prepared daily
- to obtain 100 mg/L glucose, 100 mg/L peptone, 17.5 mg/L KH<sub>2</sub>PO<sub>4</sub>, 17.5 mg/L MgSO<sub>4</sub>, 10
- mg/L FeSO<sub>4</sub>, 225 mg/L CH<sub>3</sub>COONa, and 35 mg/L urea (Alturki et al., 2010).
- 110 A set of 31 TrOCs were selected to represent four major groups of emerging organic
- chemicals of significant concern endocrine disrupting compounds, pharmaceutical and
- personal care products, industrial chemicals, and pesticides that occur ubiquitously in
- municipal wastewater. Key physicochemical properties of these TrOCs are summarized in
- 114 Table S1, Supplementary Data. Based on their effective octanol water partition coefficient
- 115 (i.e. Log D) at solution pH 8, the 31 TrOCs could be categorized as hydrophobic (i.e. Log D
- > 3.2) and hydrophilic (i.e. Log D < 3.2) (Tadkaew et al., 2011). A stock solution containing
- 117 25 μg/mL of each of TrOCs was prepared in pure methanol and stored at -18 °C in the dark.
- The stock solution was introduced into the synthetic wastewater described above to obtain a
- 119 concentration of 5 µg/L of each compound. The TrOC stock solution was used within a
- month.
- 121 2.2 Experimental systems
- 122 2.2.1 Osmotic membrane bioreactor reverse osmosis
- 123 A lab-scale OMBR-RO system was used (Figure S1a, Supplementary Data). This hybrid
- system was consisted of a feed solution reservoir, a glass bioreactor with a submerged, plate-
- and-frame FO membrane cell, a draw solution reservoir, and a cross-flow RO unit. A water
- level controller was used to regulate a Masterflex peristaltic pump (Cole-Parmer, Vernon
- Hills, IL) to feed the bioreactor. The feed reservoir was positioned on a digital balance

128 (Mettler-Toledo, Hightstown, IL), which was connected to a computer. A decrease in the feed 129 reservoir weight was recorded and then used to calculate the FO water flux. 130 The FO membrane cell was made of acrylic plastic. A flat-sheet, thin-film composite (TFC) 131 FO membrane was mounted on the cell to seal the draw solution flow channel of 20 cm long, 132 15 cm wide, and 0.4 cm high. The membrane active layer was in contact with the mixed liquor (i.e. FO mode) with an effective surface area of 300 cm<sup>2</sup>. The draw solution was 133 134 circulated from a stainless steel reservoir to the membrane cell by a gear pump (Micropump, 135 Vancouver, WA) at a cross-flow velocity of 2.8 cm/s. 136 The TFC FO membrane used in this study was obtained from Hydration Technology 137 Innovations (Albany, OR). Similar to TFC FO membranes from other suppliers (e.g. Oasys 138 Water and Porifera), this membrane comprised a thin, selective polyamide active layer and a 139 porous polysulfone support layer. These TFC FO membranes have higher rejection capacity 140 and much higher water permeability than cellulose triacetate based FO membranes (Cath et 141 al., 2013). In fact, TFC FO membranes with two to three times higher water permeability 142 than the membrane used in this study have been recently reported (Tian et al., 2015; Wei et 143 al., 2015). It is noted that the polyamide active layer of commercial membranes can be 144 slightly modified by proprietary additives. In addition, the support layer structure can also 145 influence the membrane water permeability (Lu et al., 2015). However, this study was specific to the comparison between OMBR and conventional MBR, rather than membrane 146 147 properties. Thus, findings from this study are still valid to OMBR using other FO 148 membranes. 149 The cross-flow RO unit, comprising a Hydra-Cell high pressure pump (Wanner Engineering, 150 Minneapolis, MN) and a membrane cell made of stainless steel, was coupled with OMBR to 151 reconcentrate the draw solution and produce recycling water. A flat-sheet, TFC polyamide 152 RO membrane (LFC3, Hydranautics, Oceanside, CA) was embedded into the membrane cell with a flow channel height of 0.2 cm and an effective membrane surface area of  $40 \text{ cm}^2$  (4 cm  $\times$  10 cm). A bypass valve and a back-pressure regulator (Swagelok, Solon, OH) were used to adjust the hydraulic pressure and cross-flow velocity. A temperature controller (Neslab RTE7, Waltham, MA) installed with a stainless steel heat exchanger coil was used to maintain the RO feed (i.e. OMBR draw solution) temperature at  $21 \pm 1$  °C. Water flux was monitored by a digital flow meter (Optiflow, Palo Alto, CA), which was connected to a computer. Key properties of the FO and RO membranes used in the OMBR-RO hybrid system are provided in Table S2, Supplementary Data.

#### 2.2.2 Conventional membrane bioreactor – reverse osmosis

A lab-scale, conventional MBR-RO system was composed of a hollow fibre MF membrane module (Mitsubishi Rayon Engineering, Tokyo, Japan) in a glass bioreactor and an RO unit (Figure S1b, Supplementary Data). The bioreactor and RO unit were identical to those used in the OMBR-RO system. The MF membrane was made of polyvinylidene fluoride with a nominal pore size and an effective surface area of 0.4 μm and 740 cm², respectively. The MF membrane driven by a Masterflex peristaltic pump (Cole-Parmer, Vernon Hills, IL) was operated in a cycle of 14 min suction and 1 min relaxation. The relaxation time was set to reduce membrane fouling. A high resolution (±0.1 kPa) pressure sensor (Extech Equipment, Australia) was installed to record the trans-membrane pressure (TMP).

### 2.3 Experimental protocol

Activated sludge from the Wollongong Wastewater Treatment Plant (Wollongong, Australia) was used to inoculate the two bioreactors. The bioreactors were acclimatized to the synthetic wastewater described above for over 60 days using MF membranes for effluent extraction under the same conditions. Once acclimatized with regards to bulk organic removal (i.e. over 97% total organic carbon (TOC) removal), the MF membrane was removed from one

177 bioreactor, which was then integrated with the FO and RO components to form the OMBR-178 RO hybrid system. A same RO component was coupled with the other bioreactor to establish 179 the conventional MBR-RO system. 180 Both OMBR-RO and conventional MBR-RO systems were continuously operated for 40 181 days under similar conditions in a constant temperature room (22  $\pm$  1 °C). The bioreactors with working volume of 6 L were continuously aerated to obtain a mixed liquor dissolved 182 183 oxygen (DO) concentration of  $5 \pm 1$  mg/L. The initial mixed liquor suspended solids (MLSS) 184 concentration was adjusted to approximately 5 g/L. The sludge retention time (SRT) was 185 controlled at 20 days by daily wasting 300 mL mixed liquor. The hydraulic retention time 186 (HRT) was in the range of 27 – 60 hours determined by the water flux of OMBR. A 0.5 M 187 NaCl draw solution (with effective volume of 10 L) was used for OMBR. On day 20, 100 g NaCl was added to replenish draw solute loss caused by the reverse salt flux and its passage 188 189 through the downstream RO membrane. This amount was calculated based on a decrease in 190 the electrical conductivity of the draw solution and a NaCl calibration curve. 191 Water flux of the conventional MBR was adjusted daily to be equal to that of OMBR to 192 systematically compare their effects on the downstream RO process. At the same time, the 193 RO water flux was also adjusted accordingly by regulating the applied hydraulic pressure 194 while fixing the cross-flow velocity at 41.7 cm/s. As a result, the working volume of the draw 195 solution and MBR effluent was constant at 10 L over the entire experimental period. No 196 membrane cleaning was conducted for both systems during their operation. A new RO 197 membrane was used once its normalized water permeability decreased to 0.2.

- 198 2.4 Analytical methods
- 199 2.4.1 Measurement of basic water quality parameters
- 200 Basic water quality parameters were analysed every three days. Specifically, TOC and total
- 201 nitrogen (TN) were analysed using a TOC/TN analyser (TOC-V<sub>CSH</sub>, Shimadzu, Kyoto).
- 202 Ammonium (NH<sub>4</sub><sup>+</sup>) and orthophosphate (PO<sub>4</sub><sup>3-</sup>) were determined by a Flow Injection
- 203 Analysis system (QuikChem 8500, Lachat, CO). An Orion 4-Star Plus pH/conductivity meter
- 204 (Thermo Scientific, Waltham, MA) was used to monitor the solution pH and electrical
- 205 conductivity on a daily basis.
- 206 2.4.2 Measurement of trace organic contaminants
- 207 Aqueous samples were taken from the OMBR-RO and MBR-RO systems every ten days for
- 208 TrOC analysis using a method previously described by Hai et al. (2011). Briefly, the method
- involved solid phase extraction, derivatisation, and quantification by a gas chromatography –
- 210 mass spectrometry system (QP5000 GC-MS, Shimadzu, Kyoto).
- In OMBR-RO, TrOC removal rates by the bioreactor ( $R_{Bio}$ ), OMBR ( $R_{OMBR}$ ), and OMBR-RO
- $(R_{Overall})$  are defined as follows:

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$$R_{Bio} = (1 - \frac{C_{Sup}V_{Bio} + C^*_{Draw}\Delta V_{FO}}{C_{Food}\Delta V}) \times 100\%$$
 (1)

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$$R_{OMBR} = (1 - \frac{C^*_{Draw}}{C_{Feed}}) \times 100\%$$
 (2)

$$215 R_{Overall} = \left(1 - \frac{C_{Permeate}}{C_{Feed}}\right) \times 100\% (3)$$

- where  $C_{Feed}$ ,  $C_{Sup}$ , and  $C_{Permeate}$  is the measured TrOC concentration (ng/L) in the feed, mixed
- liquor supernatant, and RO permeate, respectively;  $C^*_{Draw}$  is the TrOC concentration in the
- FO permeate;  $V_{Bio}$  is the effective bioreactor volume (i.e. 6 L); and  $\Delta V_{FO}$  is the volume of

- water passed through the FO membrane between time t and  $t+\Delta t$ . TrOCs accumulate in the draw solution when they pass through the FO membrane but are retained by the subsequent RO membrane (D'Haese et al., 2013). Thus,  $C^*_{Draw}$  is determined from a mass balance:
- $222 C^*_{Draw} = \frac{M_{FO}}{Q_{FO}} (4)$

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$$M_{FO} = \frac{V_{Draw}(C_{Draw(t+\Delta t)} - C_{Draw(t)})}{\Delta t} + \frac{\frac{(C_{RO(t+\Delta t)} + C_{RO(t)})}{2} \Delta V}{\Delta t}$$
 (5)

$$224 \Delta V = Q_{RO} \Delta t (6)$$

where  $M_{FO}$  is the mass flow rate of TrOCs crossed through the FO membrane;  $C_{Draw(t)}$  and  $C_{Draw(t+\Delta t)}$  is the measured TrOC concentration in the draw solution at time t and  $t+\Delta t$ , respectively;  $C_{RO(t)}$  and  $C_{RO(t+\Delta t)}$  is the measured TrOC concentration in the RO permeate at time t and  $t+\Delta t$ , respectively; and  $Q_{FO}$  and  $Q_{RO}$  is the water flux of the FO and RO membranes, respectively. As noted in Section 2.3, the RO water flux ( $Q_{RO}$ ) was adjusted to be equal to that of the FO membrane ( $Q_{FO}$ ). Based on eqs. (4) – (6),  $C_{Draw}^*$  is calculated from

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$$C^*_{Draw} = \frac{V_{Draw}(C_{Draw(\#\Delta t)} - C_{Draw(t)})}{\Delta V_{FO}} + \frac{(C_{RO(\#\Delta t)} + C_{RO(t)})}{2}$$
 (7)

- According to eqs. (1) (3), the observed TrOC rejection rate by the FO  $(R_{Ob\ FO})$  and RO  $(R_{Ob\ FO})$
- 233 <sub>RO</sub>) membranes is calculated as:

$$234 R_{ObFO} = R_{OMBR} - R_{Bio} (8)$$

$$235 R_{ObRO} = R_{Overall} - R_{OMBR} (9)$$

- The observed TrOC rejection rate does not reflect the real separation capacity of the FO and
- 237 RO membranes, but can be used to infer their contributions to TrOC removal in OMBR-RO.
- 238 Similar to OMBR-RO operation, the RO water flux was adjusted daily to match the MBR

effluent flow rate, maintaining the effluent reservoir with a working volume of 10 L during conventional MBR-RO operation (Section 2.3). Therefore, the calculation process listed above was also applicable to evaluate TrOC removal by different compartments of conventional MBR-RO.

#### 2.4.3 Fluorescence excitation – emission matrix spectroscopy

Fluorescence intensities of the OMBR and MBR mixed liquor supernatant, draw solution, and MBR effluent samples at the beginning and conclusion of OMBR-RO and conventional MBR-RO operation were measured to determine organic substances likely responsible for fouling of the RO membrane using a two-dimensional fluorescence spectrophotometer (Perkin-Elmer LS-55) with excitation wavelengths between 240 and 450 nm and emission wavelengths between 290 and 580 nm (in 5 nm increments). Samples were prepared and analysed based on the method reported by Cory and McKnight (2005). Fluorophores detected in certain areas of optical space in an excitation-emission-intensity matrix (EEM) correspond to specific fractions of dissolved organic matter (Henderson et al., 2009; Xie and Gray, 2016). All samples were diluted to a same TOC concentration for resolving and comparing EEM spectra.

#### 2.4.4 Microbial community analysis

Sludge samples were taken at the beginning and conclusion of OMBR and conventional MBR operation for microbial community analysis according to a method reported previously by Luo et al. (2016c). In brief, the method included DNA extraction using the FastDNA® SPIN Kit for soil (MP Biomedicals, Santa Ana, CA), PCR amplification of V1 – V3 16S rRNA gene, and amplicon sequencing on a Illumina MiSeq platform (Australian Genome Research Facility, Queensland, Australia).

Paired-end reads were assembled using PEAR (version 0.9.8) (Zhang et al., 2014) and then processed with Quantitative Insights into Microbial Ecology (QIIME 1.9.1) (Caporaso et al., 2010), USEARCH (version 8.0.1623) (Edgar, 2013), and UPARSE pipeline. All sequencing data here are available at the Sequence Read Archive (accession number: SRP072961) in the

National Centre for Biotechnology Information (Bethesda, MD).

#### 2.4.5 Biomass characterisation

MLSS and mixed liquor volatile suspended solids (MLVSS) concentrations in the bioreactor were analysed based on Standard Method 2540 (APHA, 2005). Biomass activity was evaluated by determining the specific oxygen uptake rate (SOUR) of activated sludge using Standard Method 1683 (APHA, 2005). Extracellular polymeric substance (EPS) in the sludge were extracted using a method from Zhang et al. (1999). EPS and soluble microbial products (SMP) in the mixed liquor were measured by analysing their protein and polysaccharide concentrations. Proteins and polysaccharides were determined by the Folin method with bovine serum albumin as the standard and the phenol-sulfuric acid method with glucose as the standard, respectively (Semblante et al., 2015).

## 277 2.4.6 Membrane autopsy

At the conclusion of OMBR-RO and conventional MBR-RO operation, a scanning electron microscopy (SEM) coupled with energy dispersive spectroscopy (EDS) (JCM-6000, JEOL, Tokyo, Japan) was used to identify the morphology and composition of the fouling layer on the membrane surface. Membrane samples were air-dried in a desiccator before being coated with an ultra-thin gold layer with a sputter coater (SPI Module, West Chester, PA) for SEM imaging. Attenuated Total Reflection – Fourier Transform Infrared (ATR-FTIR) spectroscopy (IRAffinity-1, Shimadzu, Kyoto, Japan) was also used to probe the chemical composition of the fouling layer. The measured spectrum ranged between 600 and 4000 cm<sup>-1</sup>

with 2 cm<sup>-1</sup> resolution. Each scan was performed 20 times. A background correction was conducted before each measurement.

### 3. Results and discussion

289 3.1 Process performance

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290 3.1.1 Mixed liquor salinity and water flux

Salinity build-up in the bioreactor is an inherent issue associated with OMBR due to the effective salt rejection by the FO membrane and the reverse draw solute flux. Thus, the mixed liquor conductivity increased significantly within the first 10 days of OMBR operation (Figure 1). Less significant conductivity increase was observed thereafter, which could be attributed to a decrease in the reverse draw solute flux associated with the water flux decline. At the same time, daily sludge wastage to control the SRT could also remove some dissolved inorganic salts, contributing to a more gradual conductivity increase from day 10 onward (Figure 1). High salinity could negatively affect the system biological stability and membrane performance (Lay et al., 2010). Since salinity build-up continued to occur, a long term study is necessary to determine the steady state level of salinity in the bioreactor. It is also noted that several strategies to mitigate salinity build-up in OMBR have been proposed, for example, by using organic draw solutions (Luo et al., 2016a) and integrating with the MF/UF membrane for salt bleeding (Holloway et al., 2015; Luo et al., 2016b). In contrast to salinity build-up in OMBR, the mixed liquor conductivity was constant at approximately 0.38 mS/cm (corresponding to 0.19 g/L NaCl) throughout conventional MBR operation (Figure 1). This is because the MF membrane does not retain any dissolved salts. Overall, different sludge characteristics, microbial community structure, and biological treatment performance between OMBR and conventional MBR were observed as discussed in the following sections.

**[Figure 1]** 

A continuous decrease in the water flux was observed for OMBR (Figure 1). The observed
flux decline could be ascribed to salinity build-up in the bioreactor, a decrease in the draw
solution concentration, and membrane fouling. Although the RO membrane effectively
rejected NaCl solute (> 98%), the draw solution concentration decreased over time (Figure S2
Supplementary Data), due to the reverse solute transport and, to a lesser extent, its passage
through the RO membrane. Both salinity increase in the bioreactor and concentration
decrease of the draw solution could reduce the net driving force (i.e. effective trans-
membrane osmotic pressure) for water permeation. On day 20 of the experiment, 100 g NaCl
was added to replenish the draw solute loss, which slightly enhanced the water flux (by
approximately 1.5 L/m²h). Despite the low fouling propensity of the FO membrane, a cake
layer was observed on the membrane surface at the conclusion of OMBR operation,
predominately consisting of carbon, oxygen, phosphorus, sodium, magnesium, and calcium
(Figure S3, Supplementary Data).
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Water flux of conventional MBR was adjusted daily to match that of OMBR and thus maintain a comparable effluent flux toward the downstream RO process. As a result, the MF membrane was operated at a relatively low water flux, which in turn resulted in negligible membrane fouling as indicated by a small TMP increase throughout conventional MBR operation (Figure S4, Supplementary Data). In practice, the water flux of conventional MBR is usually above $10 \text{ L/m}^2\text{h}$ (Hai et al., 2014), which is considerably higher than the water flux $(4-8 \text{ L/m}^2\text{h})$ used in this study. Although FO is more resistant to fouling compared to
Water flux of conventional MBR was adjusted daily to match that of OMBR and thus maintain a comparable effluent flux toward the downstream RO process. As a result, the MF membrane was operated at a relatively low water flux, which in turn resulted in negligible membrane fouling as indicated by a small TMP increase throughout conventional MBR operation (Figure S4, Supplementary Data). In practice, the water flux of conventional MBR is usually above 10 L/m²h (Hai et al., 2014), which is considerably higher than the water flux (4 – 8 L/m²h) used in this study. Although FO is more resistant to fouling compared to UF/MF given the different mechanisms of water transport (i.e. osmotically driven and

et al., 2016), fouling resistant, high flux and high separation performance FO membranes can be available in a near future. Indeed, several different research groups have reported new FO membranes with water permeability two to three times higher than that of the commercial membrane used in this study (Tian et al., 2015; Wei et al., 2015). Such progress in membrane fabrication may provide more opportunities in the deployment of OMBR with better contaminant removal and less membrane fouling.

### 3.1.2 Biomass characteristics

Salinity build-up in the bioreactor altered biomass characteristics during OMBR operation (Figure 2). A small but discernible decrease in biomass concentration (i.e. MLSS and MLVSS) and SOUR was observed within the first two weeks (Figure 2a-c). This observation is in good agreement with previous studies (Wang et al., 2014; Luo et al., 2015b), and could be ascribed to the inhibition on biomass growth and activity with salinity increase. In addition, the high salinity also increased SMP and EPS concentrations in the mixed liquor (Figure 2d, e), which might be responsible for the FO membrane fouling as discussed above. It has been reported that the elevated salinity could increase the endogenous respiration of microorganisms in activated sludge and thus enhance the secretion of organic cellular substances (Lay et al., 2010; Chen et al., 2014). Nevertheless, biomass concentration and activity recovered gradually from day 14 onward, possibly due to microbial acclimatization to the saline condition, which consequently resulted in the dominance of halotolerant bacteria in the bioreactor (Figures 3 and 4). Meanwhile, SMP and EPS concentrations in the mixed liquor decreased and then stabilized at approximately 20 and 55 mg/g MLVSS, respectively (Figure 2d, e).

357 [Figure 2]

Biomass growth (indicated by the MLSS and MLVSS concentrations) and activity (indicated by the sludge SOUR) were relatively stable during conventional MBR operation (Figure 2). However, both SMP and EPS concentrations in the mixed liquor decreased significantly, likely due to a reduction in the organic loading rate caused by the decreasing water flux (to match that of OMBR). Given a stable sludge concentration, a decrease in the organic loading rate could lower the ratio of food to microorganism (i.e. F/M), thereby increasing the SMP and EPS biodegradation (Wu et al., 2013).

### 3.1.3 Microbial community structure

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Sludge samples collected from OMBR and conventional MBR were clustered based on the unweighted Unifrac distance by applying hierarchical clustering (Figure 3). The unweighted Unifrac distance among samples represents the dissimilarity in their microbial communities in a phylogenetic tree. Results show that microbial community structure varied differently in OMBR and conventional MBR (Figures 3 and 4). Sludge samples taken at the beginning of OMBR and conventional MBR operation (i.e. on day 0) formed one cluster, confirming similar microbial communities in these two systems at the initial phase (Figure 3). However, the sludge sample collected at the end of OMBR operation (i.e. on day 40) created a branch distinct from the cluster of other samples. This result indicates that salinity build-up in OMBR significantly impacted the development of the microbial community, rendering it different from that in conventional MBR. In addition, distant clusters were observed for the two sludge samples taken at the beginning and conclusion of conventional MBR operation. This observation could be attributed to microbial variation in response to the prolonging HRT due to continuous water flux decline (Figure 1). It is noteworthy that natural and transient changes in microbial community during MBR operation could also occur (Luo et al., 2016c; Phan et al., 2016).

382 [Figure 3]

383 [Figure 4]

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Further taxonomic analysis revealed significant difference in the microbial community between OMBR and conventional MBR (Figure 3). For instance, the phylum *Planctomycetes* in conventional MBR was much more abundant than that in OMBR (Figure 4a). This result is consistent with our previous study that showed a decrease in the abundance of the phylum Planctomycetes in a conventional MBR when its influent salinity increased (Luo et al., 2016c). Microbial species of the phylum *Bacteroidetes* are usually detected in both marine and freshwater environments (Zhang et al., 2013). Thus, the phylum Bacteroidetes was identified in all sludge samples with noticeable abundance (Figure 4a). Nevertheless, the abundance of the phylum Bacteroidetes increased during OMBR operation, which could be further attributed to an increase in the dominance of the family Cytophagaceae (Figure 4b). Abundance of the phylum Proteobacteria varied differently during OMBR and conventional MBR operation, although it was the most abundant phylum in both systems (Figure 4a). As conventional MBR operated, the abundance of the phylum Proteobacteria increased significantly, which was mainly contributed by the dominance of the class  $\beta$ -proteobacteria. Members of the class  $\beta$ -proteobacteria are typically dominant in freshwater environment (Zhang et al., 2013). Detailed analysis attributed this class dominance to the predominance of the families Oxalobacteraceae and Comamonadaceae (Figure 4b). By contrast, a small increase in the abundance of the class  $\beta$ -proteobacteria was observed during OMBR operation, which was only contributed by the dominance of the family Comamonadaceae. Zhang et al. (2013) also reported an increase in the abundance of the class  $\beta$ -proteobacteria along a salinity gradient of 0.34 – 6.86 g/L NaCl in a Chinese wetland. These results indicate that some microbes affiliated to the class  $\beta$ -proteobacteria, such as Comamonadaceae, are salt-tolerant and could proliferate under saline conditions. On the other hand, the abundance of the class y-proteobacteria decreased considerably in both systems mainly due to the

decaying of the family *Xanthomonadaceae*. A similar decrease in both systems also occurred for the family *Ellin 6075* belonged to the phylum *Acidobacteria*.

#### 3.2 Contaminant removal

### 3.2.1 Removal of bulk organic matter and nutrients

No significant difference between OMBR-RO and conventional MBR-RO was observed regarding overall removal of bulk organic matter (i.e. TOC and TN) (Figure 5) and nutrients (i.e. NH<sub>4</sub><sup>+</sup> and PO<sub>4</sub><sup>3-</sup>) (Figure 6). Nevertheless, these contaminants exhibited considerably different fates and transport behaviours in the two hybrid systems. A small increase in TOC concentration in the bioreactor was observed at the beginning of OMBR operation (Figure 5a). This observation could be attributed to negative effects of salinity build-up on biomass activity as discussed above and the high rejection of biologically persistent substances by the FO membrane. The high rejection FO membrane resulted in negligible TOC concentration in the draw solution and thus ensured a complete overall removal by OMBR-RO. Given the stable biological treatment and the permeation of non-biodegradable organic substances through the MF membrane, TOC concentration in the bioreactor during conventional MBR operation was less than one-tenth of that in OMBR (Figure 5b). Organic substances that were resistant to conventional MBR treatment were effectively retained by the RO membrane, causing notable TOC accumulation in the MBR effluent reservoir and near complete removal by the whole system.

427 [Figure 5]

Without a denitrification step, TN removal by activated sludge is limited and depends mainly on microbial assimilation (Hai et al., 2014). In OMBR-RO operation, the high rejection FO and RO membranes induced a considerable TN accumulation in the bioreactor and the draw solution, respectively (Figure 5c). It has been reported that contaminants that permeated

through the FO membrane but were rejected by the RO membrane could accumulate in the draw solution in closed-loop FO-RO systems (e.g. OMBR-RO), and eventually deteriorated the product water quality (Shaffer et al., 2012; D'Haese et al., 2013). As a result, the overall TN removal by OMBR-RO decreased from nearly 100 to 50% within 40 days (Figure 5c). By contrast, TN concentration in the bioreactor was stable at approximately 30 mg/L during conventional MBR operation (Figure 5d). Nevertheless, the high rejection RO membrane caused a significant TN build-up in the MBR effluent reservoir, which consequently reduced its overall removal by MBR-RO (from approximately 98 to 40%).

Effective nitrification occurred in both OMBR and conventional MBR systems as manifested by the removal of NH<sub>4</sub><sup>+</sup> in their bioreactors (Figure 6a, b). Nevertheless, ammonia oxidizing bacteria (AOB), which oxidize ammonia to nitrite, were not efficiently detected in all sludge samples (Figure 4). Only nitrite oxidizing bacteria (which oxidize nitrite to nitrate) affiliated to the phylum *Nitrospirae* were identified at a small abundance (Figure 4a). Similar results were also reported previously and could be attributed to the presence of AOB species that were unidentifiable by 16S rRNA-gene sequencing (Luo et al., 2016c; Phan et al., 2016). Additionally, the effective NH<sub>4</sub><sup>+</sup> removal could also be contributed by ammonia oxidizing archaea, which however, were not targeted by the primers designed in this study.

**[Figure 6]** 

Similar to bulk organic matter (i.e. TOC), a small and transient increase in  $NH_4^+$  concentration was observed in the bioreactor at the beginning of OMBR operation (Figure 6a). Nevertheless, the high rejection FO membrane resulted in negligible  $NH_4^+$  concentration in the draw solution. On the other hand,  $NH_4^+$  concentration in the bioreactor was constantly low during conventional MBR operation. However, the high rejection RO membrane induced a small but discernible  $NH_4^+$  build-up in the MBR effluent reservoir from day 20 onward.

Without chemical precipitation, phosphate removal in activated sludge treatment relies mainly on microbial assimilation, especially by polyphosphate accumulating organisms (PAOs). PAOs are vulnerable to saline stress and a small osmotic pressure increase within their cells caused by salinity build-up may severely reduce their phosphate accumulating ability (Lay et al., 2010; Yogalakshmi, 2010). Nevertheless, the FO membrane can almost completely retain phosphate ions as they are negatively charged and have large hydrated radius (Holloway et al., 2007). As a result, PO<sub>4</sub><sup>3-</sup> accumulated considerably in the bioreactor while its presence in the draw solution was negligible during OMBR-RO operation (Figure 6c). On the other hand, phosphate could permeate through the MF but not the RO membrane. Thus, phosphate build-up in the MBR effluent reservoir was observed during conventional MBR-RO operation (Figure 6d). It is noteworthy that PO<sub>4</sub><sup>3-</sup> concentration in the bioreactor was slightly higher than that in influent during conventional MBR operation, possibly owing to its retention by the dynamic fouling layer formed on the MF membrane surface and/or phosphate release from unmetabolized substrates (Yogalakshmi, 2010).

470 3.2.2 Removal of trace organic contaminants

All 31 TrOCs investigated were almost completely removed by both OMBR-RO and conventional MBR-RO (Figure 6), due to the synergy of biological treatment and physical membrane rejection. Nevertheless, removal behaviours of these TrOCs were significantly different in these two hybrid systems, depending on their physiochemical properties, including hydrophobicity and molecular structure (Table S1, Supplementary Data). Of the 31 TrOCs investigated, 18 compounds were hydrophobic (i.e. Log D < 3.2) and 13 compounds were hydrophobic (i.e. Log D > 3.2) (Section 2.1).

478 [Figure 7]

All 13 hydrophobic TrOCs were biologically removed by over 90% in both systems (Figure 7). Compared to conventional MBR, salinity build-up in the bioreactor did not significantly affect the biological removal of these hydrophobic compounds during OMBR operation. Their high removal by activated sludge has also been demonstrated in several previous studies (Tadkaew et al., 2011; Wijekoon et al., 2013) and can be ascribed to their adsorption onto sludge, which facilitated their biodegradation. Results reported here are also consistent with a previous study by Luo et al. (2015b) who reported insignificant variation in the removal of hydrophobic TrOCs by conventional MBR as the mixed liquor salinity increased (up to 16.5 g/L NaCl). The effective removal of hydrophobic TrOCs by activated sludge could consequently reduce their permeation through the FO and subsequent RO membranes, leading to near complete removal by both hybrid systems (Figure 7). It has been reported that an initial adsorption but subsequent partition and diffusion of hydrophobic TrOCs (particularly, non-ionic compounds) through membranes could reduce their rejection in a stand-alone FO or RO process (Nghiem and Coleman, 2008; Xie et al., 2014). Varying removal rates of hydrophilic TrOCs were observed in both bioreactors (Figure 7). Effective removal (> 90%) was observed for several hydrophilic compounds, including salicylic acid, ketoprofen, naproxen, metronidazole, ibuprofen, gemfibrozil, enterolactone, pentachlorophenol, DEET, and estriol. This result can be attributed to the high biodegradability of these compounds, whose molecular structures have strong electron donating functional groups (e.g. amine and hydroxyl) (Tadkaew et al., 2011). On the other hand, some hydrophilic TrOCs were poorly removed in both bioreactors, with removal rates only in the range of 20 - 70%. They included clofibric acid, fenoprop, primidone, diclofenac, carbamazepine, atrazine, and ametryn, which are well-known biologically resistant substrates. Their resistance to biological treatment mainly resulted from the presence of strong electron withdrawing functional groups (such as chloride, amide, and nitro) or the lack

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of strong electron donating functional groups in their molecular structures (Tadkaew et al.,

505 2011; Wijekoon et al., 2013).

Despite the low removal of biologically persistent TrOCs by activated sludge, the high rejection FO membrane prevented their permeation into the draw solution and allowed almost complete removal by OMBR (Figure 7a). A small but nevertheless discernible contribution by the downstream RO membrane was only observed for atrazine and ametryn, which were slightly permeable through the FO membrane. Our results are consistent with previous studies which demonstrated excellent removal of TrOCs by FO or FO-RO (Hancock et al., 2011; D'Haese et al., 2013; Alturki et al., 2013). On the other hand, conventional MBR could not effectively remove these biologically persistent TrOCs although the dynamic fouling layer formed on the MF membrane surface rejected them to some extent (Figure 7b). Nevertheless, the high rejection RO membrane complemented well to MBR for the high overall removal of these compounds.

### 517 3.3 Reverse osmosis membrane fouling

Water flux of the RO process subsequent to conventional MBR was adjusted daily to match that of the RO process subsequent to OMBR (Section 2.3). Changes in the applied hydraulic pressures to the RO membrane in these two hybrid systems are shown in Figure 8a. To compare fouling development on the RO membrane surface, the normalized water permeability was also determined (Figure 8b), which is the ratio of the effective membrane water permeability to the initial value ( $P/P_0$ ).

524 [Figure 8]

The normalized water permeability of the RO membrane decreased less significantly than in conventional MBR-RO (Figure 8). This result indicates that the RO membrane fouling was more severe when treating conventional MBR effluent compared to reconcentrating the

OMBR draw solution due to their different water qualities and foulant contents (Figures 5 and 6). Thus, although the RO membrane in OMBR-RO was operated at a higher initial hydraulic pressure to overcome the osmotic pressure of the draw solution (i.e. 0.5 M NaCl), the hydraulic pressure applied to the RO membrane in conventional MBR-RO increased much more rapidly and frequent RO membrane replacement was needed to match the water flux of OMBR-RO (Figure 8a). Severe RO membrane fouling observed in conventional MBR-RO can be attributed to foulant accumulation in the MBR effluent reservoir. Indeed, EEM analysis revealed foulant build-up, such as humic-like ( $\lambda_{ex/em}$ =300-370/400-500 nm) and protein-like substances ( $\lambda_{ex/em}$ =275-290/330-370 nm), in the MBR effluent (Figure S5, Supplementary Data). The FO process effectively prevented foulants from permeating into the draw solution (Figures 6 and 7), thereby reducing membrane fouling in the downstream RO process. For instance, the humic- and protein-like substances accumulated considerably in the bioreactor, but their presence in the draw solution was negligible (Figure S5, Supplementary Data). However, the RO normalized water permeability decreased gradually and stabilized at approximately 0.35 from day 20 onward during OMBR-RO operation (Figure 8b). The observed permeability decline could be attributed to membrane compaction (particularly within the first week of operation) and fouling under the high hydraulic pressure (Figure 8a). The fouling layer on the RO membrane surface exhibited different morphologies in OMBR-RO and conventional MBR-RO (Figure 9a, b). Foulant clusters were sparsely distributed without forming a dense fouling layer on the RO membrane surface subsequent to OMBR (Figure 9a). Elementary analysis by EDS revealed that these clusters comprised carbon, oxygen, sodium, and chloride (Figure 9c). By contrast, a compact and homogenous cake layer formed on the RO membrane surface in conventional MBR-RO (Figure 9b), predominantly containing carbon, oxygen, magnesium, calcium, and phosphate (Figure 9d).

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This result indicates the formation of both organic and inorganic membrane fouling. However, regularly shaped or needle-like crystals typically formed with inorganic scaling were not visualized on the RO membrane surface subsequent to conventional MBR although magnesium, calcium, and phosphate were detected (Figure 9b). This result was possibly due to the formation of inorganic precipitates in the organic fouling layer or the complexation between these divalent cations and organic molecules (e.g. protein-like substances) on the membrane surface (Zhao et al., 2010).

560 [Figure 9]

Organic fouling layer on the RO membrane surface was characterized by ATR-FTIR measurement (Figure 10). The pristine RO membrane showed typical absorbance peaks at wavenumbers of 3345 cm<sup>-1</sup> (N-H stretching), 3300 cm<sup>-1</sup> (O-H stretching), 1671 cm<sup>-1</sup> (strong amide C=O), 2946 and 1487cm<sup>-1</sup> (C-H stretching), and 1168 cm<sup>-1</sup> (amide ring). Similar ATR-FTIR spectra were also observed for the RO membrane coupled with OMBR, confirming the formation of slight and scattered organic fouling layer on the membrane surface. By contrast, the RO membrane subsequent to conventional MBR exhibited distinctive adsorption peaks at 1653 cm<sup>-1</sup>, which usually associates with alkene (C=C) in aliphatic structures and/or amide I (C=O) bonds, and at 1543 cm<sup>-1</sup>, representing amide II (C-N-H) bonds. In addition, the fouled RO membrane also showed a sharp peak at 1032 cm<sup>-1</sup>, indicating carbonyl (C=O) bonds of polysaccharides. These results suggest that humic- and protein-like substances accumulated in the MBR effluent were likely responsible for the severe organic fouling of the RO membrane in conventional MBR-RO.

574 [Figure 10]

#### 575 3.4 Implications

High product water quality and low membrane fouling imply robustness of OMBR-RO in advanced wastewater treatment and water reuse. Results reported here highlight the benefits of OMBR-RO over conventional MBR-RO. Compared to conventional MBR-RO, the high rejection FO membrane prevents the downstream RO process from severe membrane fouling, thereby reducing membrane cleaning and maintenance during OMBR-RO operation. Moreover, OMBR-RO has the potential to simultaneously achieve seawater desalination and wastewater recycling when seawater is used as the draw solution in coastal regions. By virtue of osmotic dilution, low pressure RO systems can be coupled with OMBR to remove the need for concentrate disposal, thereby reducing energy consumption for seawater desalination and wastewater recovery (Valladares Linares et al., 2016). several strategies to mitigate salinity build-up in OMBR have been proposed, for example, by using organic draw solutions (Luo et al., 2016a) and integrating with the MF/UF membrane for salt bleeding (Holloway et al., 2015; Luo et al., 2016b).

Sludge produced by OMBR is expected to be saline. Thus, further study is necessary to quantify the impact of salinity on subsequent sludge treatment and available sludge reuse

### 4. Conclusion

options.

Results reported here show that both OMBR-RO and conventional MBR-RO systems can effectively remove bulk organic matter, nutrients, and all 31 TrOCs investigated. Nevertheless, salinity build-up in the bioreactor reduced the water flux and adversely impacted biological stability by altering biomass characteristics and microbial community structure during OMBR operation. Salinity increase also resulted in more SMP and EPS in the mixed liquor, inducing the FO membrane fouling. With the succession of halophobic

- bacteria by halotolerant ones, the OMBR system remained biologically active. Moreover, the high rejection of foulants by the FO membrane prevented the downstream RO process from severe membrane fouling. In contrast to biological variation in OMBR, biological performance was relatively stable during conventional MBR operation. However, foulants (e.g. humic- and protein-like matters and inorganic salts) accumulated considerably in the
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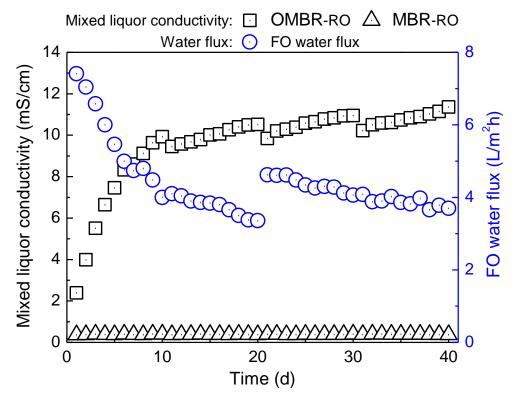
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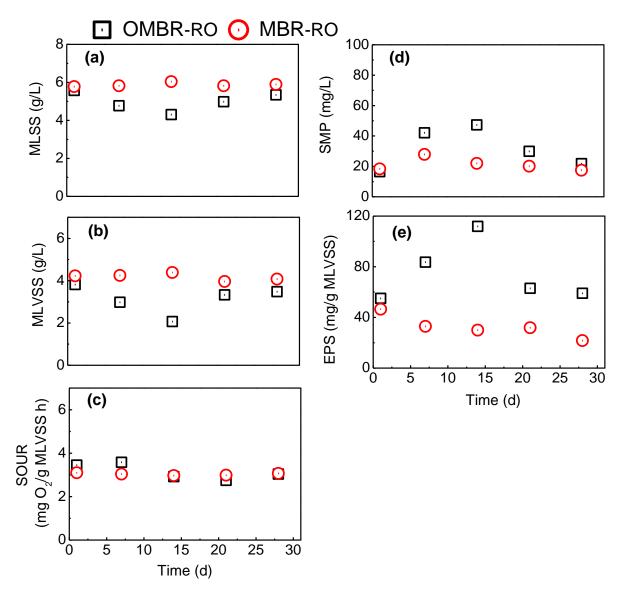
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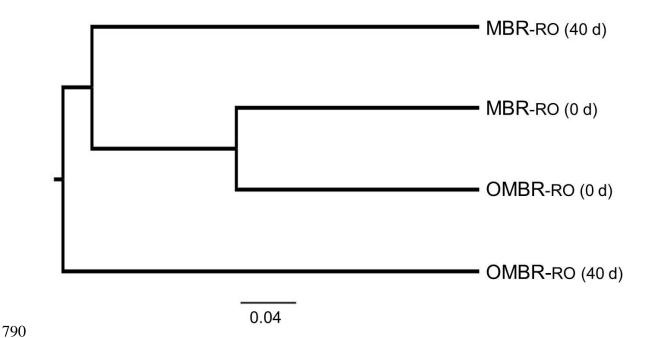
## **LIST OF FIGURES**



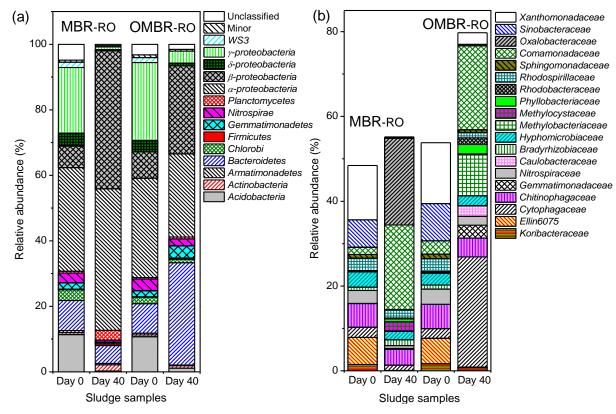
**Figure 1:** Mixed liquor electrical conductivity and FO water flux during OMBR-RO and conventional MBR-RO operation. MF (used in conventional MBR) and RO water fluxes were adjusted daily to match that of FO. The MF membrane was operated in a cycle of 14 min on and 1 min off. Experimental conditions: DO = 5 mg/L; initial MLSS = 5.5 g/L; HRT = 27 - 60 h; SRT = 20 d; temperature =  $22 \pm 1$  °C; initial FO draw solution = 0.5 M NaCl; draw cross-flow velocity = 2.8 cm/s; RO cross-flow velocity = 41.5 cm/s. On day 20, 100 g NaCl was added to OMBR draw solution (with constant working volume of 10 L) to replenish draw solute loss.



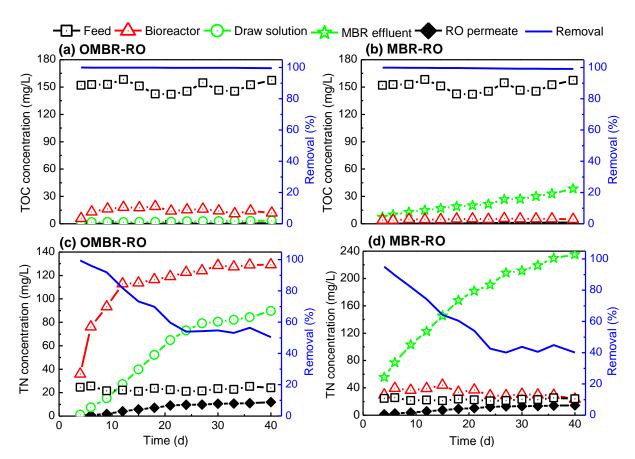
**Figure 2:** Key biomass characteristics during OMBR-RO and conventional MBR-RO operation.



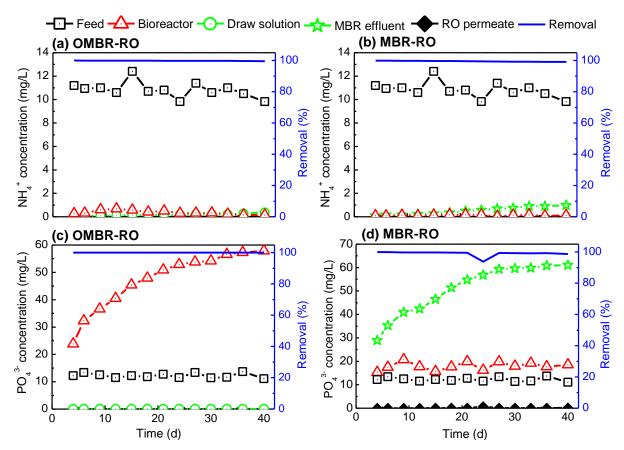
**Figure 3:** Hierarchical clustering based on the unweighted UniFrac metric. The branch length represents the distance (indicated by scale bar) among bacterial communities of sludge samples in UniFrac units. Labels on the branch indicate sludge samples collected from bioreactors at the beginning (0 day) and conclusion (40 day) of OMBR-RO and conventional MBR-RO operation. Experimental conditions are as described in the caption of Figure 1.



**Figure 4:** Relative abundance of dominant (a) phyla and (b) families (>1%) in sludge samples collected from bioreactors at the beginning (day 0) and conclusion (day 40) of OMBR-RO and conventional MBR-RO operation. The phylum *Proteobacteria* comprised the classes  $\alpha$ -,  $\beta$ -,  $\delta$ - and  $\gamma$ -proteobacteria. Phyla with relative abundance < 0.5% were grouped as "Minor".



**Figure 5:** TOC and TN concentrations and removal rates during OMBR-RO and conventional MBR-RO operation. The two systems were operated under the same conditions as described in the caption of Figure 1.



**Figure 6:**  $NH_4^+$  and  $PO_4^{3-}$  concentrations and removal rates during OMBR-RO and conventional MBR-RO operation. Experimental conditions are as described in the caption of Figure 1.

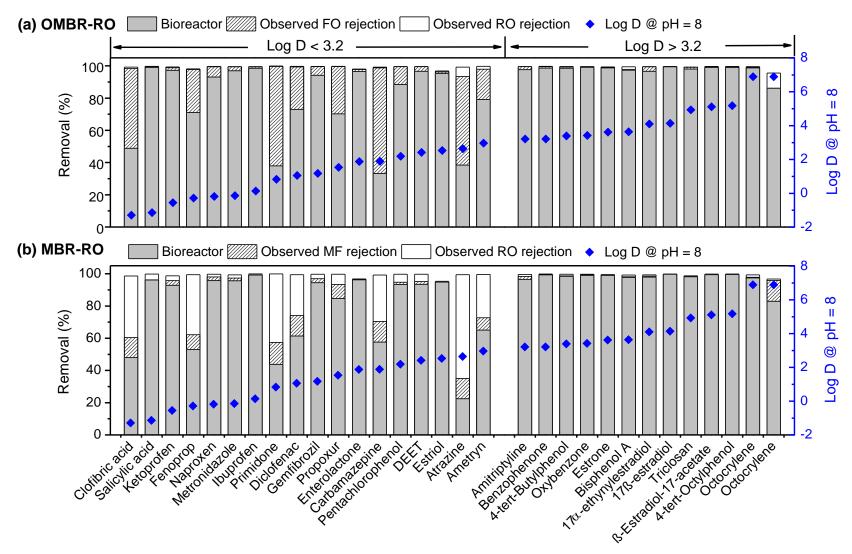
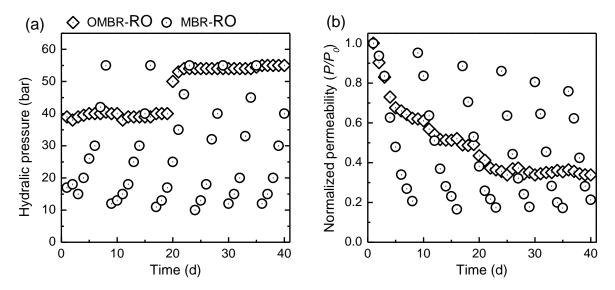
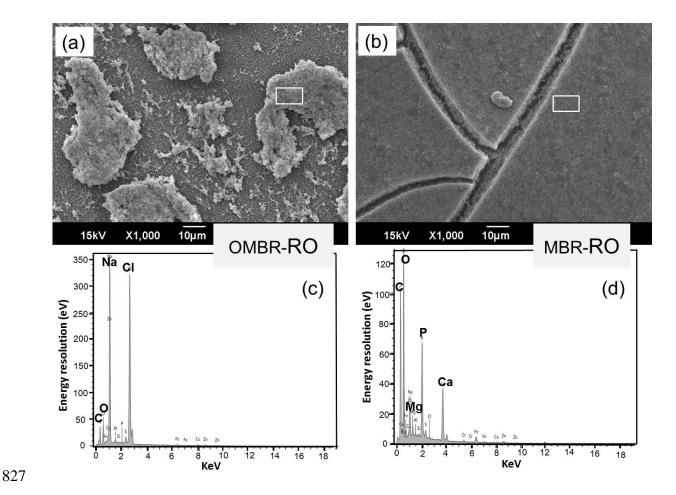


Figure 7: TrOC removal by different compartments of (a) OMBR-RO and (b) conventional MBR-RO. Average removal data obtained from four measurements (once every ten days) were demonstrated with standard deviation in the range of 0 - 20% (not shown in the Figure). Based on

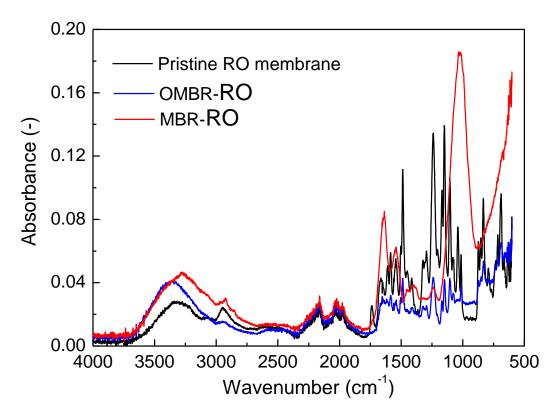
their effective octanol-water partition coefficient (Log D) at solution pH 8, the 30 TrOCs investigated were classified as hydrophobic (i.e. Log D > 3.2) and hydrophilic (i.e. Log D < 3.2). Observed TrOC rejection rates do not reflect the real separation capacity of the membranes, but can be used to infer their contributions to TrOC removal in the two hybrid systems. Experimental conditions are as described in the caption of Figure 1.



**Figure 8:** Hydraulic pressure (a) applied to the RO membrane and its normalized permeability (b) during OMBR-RO and conventional MBR-RO operation. The normalized water permeability was the ratio of the effective membrane water permeability to the initial value ( $P/P_0$ ). Water flux of the RO membranes was adjusted daily to match that of OMBR. On day 20, 100 g NaCl was added to OMBR draw solution (with constant working volume of 10 L) to replenish draw solute loss. A new RO membrane was used once the membrane normalized permeability decreased to 0.2. Experimental conditions are as described in the caption of Figure 1.



**Figure 9:** (a, b) SEM and (c, d) EDS analyses of the RO membrane surfaces at the conclusion of OMBR-RO and conventional MBR-RO operation. Experimental conditions were as described in the caption of Figure 1.



**Figure 10:** ATR-FTIR absorption spectra of the RO membrane surfaces before and after OMBR-RO and conventional MBR-RO operation. Experimental conditions are as described in the caption of Figure 1.