The potential biodiesel production from Cerbera odollam oil (Bintaro) in Aceh

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Abstract. Biodiesel production from non-edible vegetable oils is an effective way to conquer the linked problems with edible oils such as food versus fuel and other environmental impacts. Cerbera odollam oil is one of these possible non-edible feed stocks for future biodiesel production. This study evaluated the potential biodiesel production from cerbera odollam. The seed was collected and extracted from Aceh, Indonesia. Moreover, biodiesel has been produced using degummed (H3PO4) and two step acid catalyst (HCl) and alkaline catalyst (KOH). The results of properties of the cerbera odollam methyl esters show that such as viscosity was about 847.9 mm²/s, density was 3.1578 kg/m³, flash point was 214.0 °C, acid value was 0.4 mg KOH/g, oxidation stability was 6.35 h, FAME content was 97.77 % w/w and heating value was 40.49 MJ/kg. After analysing these properties, it has been found that there is a huge chance to produce biodiesel from this seed which complies with the limits of ASTM 6751 and EN 14214 specifications and therefore it can boost the future production of biodiesel from non-edible sources.

Keywords: Cerbera odollam, esterification, transesterification, biodiesel.

1. Introduction

The diminishing supply of fossil fuels reserves and the growing environmental concerns have made renewable energy an exceptionally attractive alternative energy source for the future [1,2]. Biodiesel is a promising alternative fuel for diesel fuel. It is defined as the mono-alkyl esters of long chain fatty acids derived from vegetable oils or animal fats and alcohol with or without a catalyst. It is renewable, biodegradable non-toxic, portable, readily available and eco-friendly fuel [3]. The Indonesian government has developed alternative fuel to tackle the energy crisis to solve the economic revival of society. Therefore, to the best of the author’s knowledge, limited study on the possibility of cerbera odollam oil as a potential non-edible oil feed stock for biodiesel production has been reported in the literature.

1.1 Objective of this study

The purpose of this paper is to produce biodiesel from cerbera odollam oil using pretreatment degummed oil, homogeneous acid HCl catalyst and alkaline KOH catalyst
followed by an investigation of some physical and chemical properties of the produced methyl ester (COME). The success of this study will promote more investigation of this feed stock for biodiesel production and eventually could yield promising and massive new raw material for biodiesel production on a large scale.

1.2 Description of Cerbera Odollam

*Cerbera odollam* belongs to mangrove family. It is native to the mangrove swamps of Asia, Australia, Madagascar, and the western islands of the Pacific Ocean. It is a medium to large tree with large, glossy, heart-shaped leaves that are conspicuously veined and long-stemmed. The flowers are large and greenish; the sexes are separate. The fruit is oval and about the size of a large husked coconut, brown and rough-surfaced as shown in Fig.1. *Cerbera odollam* trees are abundant due to it is used as a coastal crop, greening and decorating the city park and requires no special maintenance [4]. *Cerbera odollam* seeds contain oil yield is about 43-64% [5].

![Fig. 1 Cerbera odollam tree (a), fruit (b)and seeds (c)](image)

2. Material and Methods

2.1 Materials and Réagents

The *Cerbera odollam* seeds was supplied from Banda Aceh, Indonesia through personal communication in order to obtain Crude *Cerbera Odollam* Oil (CCOO). Reagent chemical such as methanol, H$_3$PO$_4$, HCl, KOH were used. Additionally, the qualitative filter paper (Filtres Fioroni, France) of 150 mm size was used to purify the crude oil from solidified matters.

2.2 Methodology

2.2.1 Degumming process

In this process, 0.7 % (vol) of phosphoric acid (H$_3$PO$_4$, 20% concentration) was added to crude *cerbera odollam* oil at a temperature of 70°C for 30 minutes in a glass beaker. This mixture was separated in separation funnel by density separation in which the phosphate compounds drop down at the bottom. The refined oil produced from this process was reported to be light yellow.

2.2.2 Acid esterification

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In this process, 100 gr of Crude Cerbera Odollam Oil (CCCO) was entered into a two-necked boiling flask equipped with a magnetic stirrer, reflux condenser and thermometer at room temperature. The sample of CCCO was mixed with 1:8 molar ratios of methanol and the mixture was stirring at a speed of 300 rpm. Then, hydrochloric acid (HCl) was added to the sample drop by drop using a connected pipe to the flask. The amount of HCl catalyst added was 1% (v/v). During the reaction, the mixture was stirring at constantly speed of 300 rpm using a magnetic stirrer for 3 hours at 65°C. The esterified oil was then removed from the reactor and entered into a separation funnel left for 24 hours to remove the upper layer which includes extra and methanol from the lower layer of the first esterified CCCO.

2.2.3 Alkaline transesterification

In this process, cerbera odollam esterified oil (COEO) was mixed with 29.76 gr of methanol while stirring at speed of 300 rpm. Then 2 gr of KOH of alkaline catalyst using in this process was diluted which added to the oil and maintained at 60°C for 2 hours. The oil was then removed from the reactor and entered into a separation funnel left for 24 hours to remove the upper layer of cerbera odollam methyl ester (COME) oil from extra methanol and glycerol formed at the lower layer.

2.2.4 Characterization of crude Cerbera odollam oil and methyl ester

The crude cerbera odollam was found to be dark brown yellow. The physicochemical properties were tested according to the ASTM D6751 and EN 14214 standards. Several tests such as kinematic viscosity, density, flash point, acid value, oxidation stability, FAME content and heating value. Moreover, the fattyacid methyl ester (FAME) content in the biodiesel was measured by gas chromatography based on EN 14103 method which determined using the following equation [6]:

\[
FAME = \left[ \frac{\sum A - A_{EI}}{A_{EI}} \right] \times \left[ \frac{C_{EI} \times V_{EI}}{m} \right] \times 100 \%
\]

Where \(FAME\) represents the fattyacid methyl ester content (%), \(\sum A\) represents the total gas chromatography peak area of methyl ester, \(A_{EI}\) represents the gas chromatography area of internal standard, methyl heptadecanoate, \(C_{EI}\) represents the concentration of the standard solution (mg/ml), \(V_{EI}\) represents the volume of the standard solution (ml) and \(m\) represents the amount of the biodiesel sample (mg). The methyl ester yield was determined using the following equation [7]:

\[
\text{Methyl ester yield} \ (\%) = \left[ \frac{\text{FAME} \times W_{ME}}{W_{CO}} \right] \times 100 \%
\]

Where \(FAME\) represents the fattyacidmethyl ester content (%), \(W_{ME}\) represents the weight of the cerbera odollam methyl ester (g) and \(W_{CO}\) represents the weight of the crude cerbera odollamoil (g). The equipment used to analyse physiochemical properties of crude cerbera odollam oil is listed in Table 1.

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Table 1. List of equipment used for biodiesel properties

<table>
<thead>
<tr>
<th>Property</th>
<th>Equipment</th>
<th>Standard method</th>
<th>Accuracy</th>
</tr>
</thead>
<tbody>
<tr>
<td>Kinematic viscosity</td>
<td>Stabinger viscometer™ SVM 3000</td>
<td>ASTM D7042</td>
<td>± 0.01 mm²/s</td>
</tr>
<tr>
<td>Density</td>
<td>DM40 LiquiPhysics™ density meter</td>
<td>ASTM D127</td>
<td>± 0.1 kg/m³</td>
</tr>
<tr>
<td>Acid value</td>
<td>Automation titration rondo 20</td>
<td>ASTM D664</td>
<td>±0.001 mg KOH/g</td>
</tr>
<tr>
<td>Flash point</td>
<td>NPM 440 Pensky-martens flash point</td>
<td>ASTM D93</td>
<td>± 0.1°C</td>
</tr>
<tr>
<td>Heating value</td>
<td>6100EF Semi auto bomb calorimeter</td>
<td>ASTM D240</td>
<td>± 0.001 MJ/kg</td>
</tr>
<tr>
<td>FAME content</td>
<td>QP2010 Ultra, Shimizu</td>
<td>EN 14103</td>
<td>± 0.008% or 0.0008 min</td>
</tr>
<tr>
<td>Oxidation stability</td>
<td>873 Rancimat</td>
<td>EN 14112</td>
<td>± 0.01 h</td>
</tr>
</tbody>
</table>

3. Results and discussion

3.1 Physico-chemical properties of Cerberaodollam methyl ester

The physical and chemical properties of the cerbera odollam methyl ester (COME) produced were analyzed using ASTM specification: kinematic viscosity at 40°C (ASTM D445), density at 15°C (ASTM D1298), flash point (ASTM D93), acid value (ASTMD664), heating value (ASTM D240) etc. The physical and chemical properties of COME are listed in Table 2. The viscosity of the CPME is 4.56 mm²/s, thus meeting the requirements of both the ASTM D6751 and EN-14214 biodiesel standards, which prescribe viscosity ranges that should be between 1.9–6.0 and 3.5–5.0 mm²/s, respectively. The density of methyl ester usually varies between 860 and 900 kg/m³. The density of the COME is 878.10 kg/m³. The acid value of COME was observed to be 0.47%. These results are in agreement with the value specified in ASTM D664. The heating value of cerbera odollam methyl ester was found 39.46 MJ/kg respectively.

3.2 Fatty acid methyl ester of COME

The fatty acid methyl ester of the COME is shown in is shown in Fig.2 and a picture for the produced biodiesel is shown in Fig. 3. It can be seen that the COME biodiesel is mostly composed of oleic acid (19.33%), linoleic acid (20.15%), linolenic acid (9.82) and palmitic acid (17.10%). The total unsaturated fatty acid content of the COME is found to be 54.08% which contributes to the improve deoxidation stability [8]. The composition of fatty acids in methyl ester may affect oxidation stability. This can be explained that the content of unsaturated fatty acids are sensitive to oxidative degradation, as well as unsaturated fatty compounds have higher flash points than saturated fatty compounds [9,10].

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Table 2. Physicochemical properties of cerbera odollam methyl ester

<table>
<thead>
<tr>
<th>Property</th>
<th>COME</th>
<th>Limit ASTM D6751</th>
<th>Limit EN 14214</th>
</tr>
</thead>
<tbody>
<tr>
<td>Kinematic viscosity at 40°C (mm²/s)</td>
<td>847.9</td>
<td>880</td>
<td>860-900</td>
</tr>
<tr>
<td>Density at 15°C (kg/m³)</td>
<td>3.1578</td>
<td>1.9–6.0</td>
<td>3.5–5.0</td>
</tr>
<tr>
<td>Acid value (mg KOH/g)</td>
<td>0.4</td>
<td>Max. 0.5</td>
<td>Max. 0.5</td>
</tr>
<tr>
<td>Heating value (MJ/kg)</td>
<td>40.490</td>
<td>35</td>
<td>–</td>
</tr>
<tr>
<td>Flash point (°C)</td>
<td>214.0</td>
<td>Min. 130</td>
<td>Min. 101</td>
</tr>
<tr>
<td>Oxidation stability (h)</td>
<td>6.35</td>
<td>3</td>
<td>6</td>
</tr>
<tr>
<td>FAME content (%)</td>
<td>97.77</td>
<td>–</td>
<td>96.5</td>
</tr>
</tbody>
</table>

Fig. 2 Gas chromatography of cerbera odollam methyl ester

Fig. 3 Cerbera odollam biodiesel

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4. Conclusion

Cerbera odollam oil is one of non-edible feedstocks which can replace fossil diesel. The evaluation of biodiesel production using two step acid esterification and alkaline transesterification was performed in order to obtain high methyl ester yield. In this study, the FAME of cerbera odollam biodiesel attained 99.77 % w/w and the measured properties were within the limits EN14214 and ASTM D 6751 standards. The results show that all of the reaction variables which clearly influenced the acid esterification and alkaline transesterification in a positive manner. From all of these encouraging outcomes, and given further work testing on diesel engine performance, crude cerbera odollam oil could be recommended as a supplementary non-edible oil feedstock for biodiesel production.

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References: