

20 **Abstract**

21 Water treatment sludge (WTS) management is a growing global problem for water treatment
22 plants (WTPs) and governments. Considering the scarcity of raw materials in many parts of the
23 planet and unique properties of WTS, extensive research has been conducted on the application
24 of WTS in the production of construction materials such as roof tiles, bricks, lightweight
25 aggregates, cement, concrete and geopolymers. This paper critically reviews the progress in the
26 application of WTS in construction materials, by synthesizing results from recent studies.
27 Research findings have revealed that incorporation of $\leq 10\%$ alum-based sludge in ceramic
28 bricks is satisfactory with a small reduction of mechanical performance. Using the iron-based
29 sludge, the bricks presented better mechanical strength than the reference clay-bricks.
30 Concerning WTS application in concrete, 5% replacement of cement or sand by WTS was
31 considered as the ideal value for the application in a variety of structural and non-structural
32 concrete without adverse effect on concrete mechanical performance. Furthermore, this paper
33 discusses sludge-amended concrete in terms of durability, potential leaching of toxic elements
34 and cost, and suggests topics for future research on the sustainable management of WTS.

35

36 **Keywords:** Water treatment sludge; Construction materials; Mechanical properties; Durability;
37 Solid waste recycling

38 1. Introduction

39 To protect public health, raw water is treated to a high standard at water treatment plants (WTPs)
40 before drinking water distribution, through the process of coagulation, flocculation,
41 sedimentation, filtration and disinfection (Ahmad et al., 2016; 2017). Large quantities of water
42 treatment sludge (WTS) are produced daily by WTPs, and its management is becoming a global
43 problem of major concern (Dassanayake et al., 2015; Ahmad et al., 2016; Hidalgo et al., 2017;
44 Lee et al., 2018). A recent survey by Maiden et al. (2015) showed that the amount of WTS
45 generated by Australian WTP could reach up to 43500 tons per year. In 2011, Japan generated
46 over 290000 dry tons of sludge across the country (Fujiwara, 2011), while the UK produced
47 around 131000 tons in 2014 (Finlay, 2015). **Figure 1** shows an estimated amount of sludge per
48 capita produced in selected countries. It is estimated that an ordinary WTP generates over 10000
49 tons/day and 100000 tons/year of WTS (Babatunde and Zhao, 2007; Ahmad et al., 2016).

50 A major environmental challenge for water treatment is the disposal of excessive sludge
51 produced in the process, which are either discharged into waterways or disposed to landfills
52 (Ooi et al., 2018). The raw water quality and the treatment process are the key factors
53 determining the quantity and quality of the sludge produced. Thus, any change in the quality of
54 natural water, seasonal change, as well as the dosage of the coagulants used in the treatment
55 system, will alter the quantity and quality of the sludge produced (Ahmad et al., 2016).

56 Curiously, in the scientific literature, there are limited statistical data on the production and
57 costs associated with WTS on a global or national scale (Babatunde and Zhao, 2007; Keeley et
58 al., 2014; Ahmad et al., 2016). Furthermore, other problems such as the risk of water
59 contamination with heavy metals, high costs of sludge treatment and disposal are challenges
60 faced by water companies and governments around the world. In Australia, the cost associated
61 with the sludge disposal without transportation is \$130-200/ton through landfill or/and sewer
62 disposal, with an approximate cost of over \$6.2 million per annum just in the state of Victoria
63 (Maiden et al., 2015). The costs associated with sludge disposal in the Netherlands are on

64 average between £30 and £40 million (Evuti and Lawal, 2011), while in the UK the annual
65 disposal costs £5.5 million (Keeley et al., 2014).

66 WTS has been used in various industrial and commercial manufacturing processes.
67 Particularly, some studies have been conducted by using WTS for contaminant removal in
68 wastewater treatment, coagulant recovery, in ceramic products, and land application (Li et al.,
69 2013; Ahmad et al., 2016; Cremades et al., 2018; Hagemann et al., 2019). As clay and WTS
70 have a similar mineralogical composition (essentially hydroxides and oxides of silica,
71 aluminum and ferric), the use of WTS has been highly encouraged to partially replace the clay
72 used for fabrication of cement and others sintered ceramic materials (Cremades et al., 2018;
73 Hagemann et al., 2019). Furthermore, because of clay scarcity in many parts of the world and
74 the high environmental impact generated by the clay harvesting, many studies are focused on
75 designing and development of new sustainable materials using less conventional raw materials,
76 with available local sources and in accordance with international standards requirements. In
77 order to protect the natural environment and the limited clay resources, countries such as China
78 have begun to limit the use of clay-bricks and publicly recommend the incorporation of waste
79 materials into the ceramic products (Chen et al., 2011).

80 There are a few review papers on WTS application and reuse (Babatunde and Zhao, 2007;
81 Raut et al., 2011; Ahmad et al., 2016), most of which have focused on general aspects of brick
82 or cement production from the last decade. Furthermore, the majority of the studies do not
83 differentiate between the different types of sludge such as iron-based, aluminum-based, or lime
84 (calcium carbonate, CaCO_3) which have been applied in the building materials, yet they have
85 different properties and behaviour. This review therefore evaluates the different types of WTS
86 and their latest applications in building materials such as ceramic bricks, paving tiles,
87 lightweight aggregates, cement, concrete and composite materials. In addition, it discusses
88 important aspects such as durability, environmental risks and cost-benefit analysis from the
89 application of different types of WTS. Recommendations are provided for future studies in this

90 emerging **research topic**.

91

92 **2. Characteristics of water treatment sludge**

93 2.1. Chemical properties

94 In most water treatment systems around the world with elevated suspended particles, inorganic
95 coagulants are more commonly applied (**Kashyap and Datta, 2017**). They are composed mainly
96 of iron and aluminum salts such as aluminum sulfate [$\text{Al}_2(\text{SO}_4)_3$], ferric sulfate [$\text{Fe}_2(\text{SO}_4)_3$] and
97 ferric chloride (FeCl_3). Whereas in the softening process with high levels of hardness, the main
98 coagulant used is lime (Sales et al., 2011). These coagulants are effective in removing a wide
99 variety of impurities from the water, including colloidal particles and dissolved organic
100 substances. The final process of water filtration results in a range of by-products known as
101 WTS, which consists of a mixture of microorganisms, organic suspended compounds,
102 coagulant and chemical oxides (Babatunde and Zhao, 2007; **Ahmad et al., 2016**). The final
103 composition of WTS is the result of the coagulant type and other chemical treatment used for
104 the purification of water. The typical sludge composition is shown in **Table A1**.

105 Particularly, aluminum sulfate has been used worldwide for more than 100 years in
106 different WTPs, due to its low cost, easy manipulation and storage (Renault et al., 2009). Most
107 WTPs in New Zealand and Australia use aluminum-based coagulants (**Ministry of Health,**
108 **2017**). **However, in the most populated NSW State in Australia iron-based coagulants are used**
109 **in the majority of its WTPs** due to severe manganese limits in the **finished water**. The United
110 Kingdom uses an average of 107000 tons of aluminum-based coagulant and 165000 tons of
111 **iron-based coagulants** per year at WTPs across the country (Henderson et al., 2009). In 1994,
112 in the United States and Canada, 72% of all water produced came from coagulation with
113 aluminum salts and 23% with iron salts. There are several factors that lead to the use of one or
114 another coagulant in water treatment, including the search for better quality of treated water,

115 local availability, lower cost, smaller volume of sludge and best dehydration conditions (Pizzi,
116 2010).

117

118 2.2. Geotechnical properties

119 **The knowledge of WTS geotechnical properties is essential to facilitate its application (Keeley**
120 **et al., 2014). The** properties such as mass loss on ignition (LOI), liquid limit (LL), plastic limit
121 (PL), plasticity index (PI) and specific gravity are crucial not only for **WTS** identification or
122 classification as a soil material, but also to predict future behavior of the application of this
123 material (Komlos et al., 2013; O'Kelly, 2008). **Table 1** shows that the majority of geotechnical
124 characterization studies with WTS **have** demonstrated a substantial similarity with clayey soils,
125 especially WTS from aluminum and iron salts. **However, WTS is different to clays due to the**
126 **presence of organic matter and high concentrations of chemicals (Lee et al., 2018). Such organic**
127 **matter in the soil usually results in high plasticity and low permeability, and decreases the**
128 **specific gravity (Mitchell and Soga, 2005).**

129 O'Kelly (2008) analyzed the geotechnical properties of WTP residuals that had aluminum
130 as a coagulant. The study characterized the sludge as a clay with high plasticity, high
131 compressibility, and very low permeability, which were attributed to the high affinity of the
132 coagulant metal by water and the high content of organic particles. These characteristics of
133 natural and untreated sludge make it sometimes inappropriate for applications in building
134 materials such as aggregates and structural elements, and help to explain the difficulty of
135 handling and transporting such material. Thus, by proper treatment to the sludge and its
136 incorporation into the most suitable material, the negative aspects of these characteristics can
137 be mitigated (Keeley et al., 2014).

138

139 **3. Application of water treatment sludge in construction materials**

140 3.1. Bricks and ceramic products

141 Considering the vast potential of solid waste from various industries as clay replacement for
142 the production of ceramic artefacts (roof tiles and bricks), the latest research has focused on
143 offering alternatives and environmentally friendly destination for these by-products. In
144 addition, materials resulting from the sintered ceramics are tolerant of additions of other raw
145 materials even in high quantities, **promoting the solidification and immobilization of the**
146 **elements with toxic potential (Dondi et al., 2016; Jonker and Potgieter, 2005) and high**
147 **durability (Toya et al., 2007).** Therefore, due to the fact that clay and WTS have a similar
148 chemical composition (essentially hydroxides and oxides of silica, aluminum and ferric), the
149 use of WTS has been highly encouraged to partially substitute the clay used for fabrication of
150 bricks and ceramic materials (Toya et al., 2007; Cremades et al., 2018). This is especially the
151 case in countries such as Brazil, where the production of sintered construction materials (bricks,
152 tiles, blocks) is still more advantageous than cement and geopolymeric materials which will
153 remain so for the next two decades (Monteiro and Vieira, 2014).

154 The amount of sludge added in the ceramic materials depends partly on its property (particle
155 size, chemical and mineral composition), which will directly influence the quality of the bricks
156 produced (Teixeira et al., 2011). In most studies, the increase in the addition of sludge in the
157 brick mix resulted in a decrease in compressive strength and higher water absorption. The high
158 values of loss on ignition are related to the high concentration of organic matter contained in
159 the sludge, which was burned during the sintering process, combined with the presence of
160 inorganic components found in WTS and clay (Wolff et al., 2015).

161 **Teixeira et al. (2011) showed that aluminium-based sludge was more deleterious than iron-**
162 **based coagulants to ceramic materials.** In general, the iron-based sludge applied in ceramic
163 products obtained better results concerning the mechanical properties and also a reduction in
164 the firing temperature of bricks (Anderson et al., 2003; Teixeira et al., 2011; Kizinievič et al.,
165 2013). Furthermore, Kizinievič et al. (2013) concluded that usually the iron-sludge conferred a
166 more reddish colour to the bricks, acting as a natural pigment. However, the results with the

167 addition of high proportions of aluminium-based sludge show a considerable reduction in the
168 mechanical performance of the ceramic bodies with increasing sludge addition (Huang et al.,
169 2001; Teixeira et al., 2011). In contrast, Benlalla et al. (2015) obtained bricks with mechanical
170 properties higher than reference clay bricks, even with the substitution of clay by around 30%
171 WTS at a firing temperature of 1000 °C.

172 A significant amount of studies discussing the use of WTS in combination with other
173 alternative materials have been reported in **Table A2**. In a UK study, Anderson et al. (2003)
174 investigated the incorporation of two by-products in the manufacture of bricks. They used
175 sludge generated during water and sewage treatment, which was incinerated and added as
176 partial substitutes for the traditional raw materials of the brick, in proportions of 5% (Dry
177 weight), meeting local standards and parameters for ceramic blocks. In a similar study in
178 Taiwan, Huang et al. (2001) mixed the sludge from the water treatment with dam sediment for
179 the production of ceramic samples and showed that WTS (< 20%) combined with dam sediment
180 can generate quality bricks under Chinese National Standard. In Taiwan, Chiang et al. (2009;
181 2010) produced lightweight bricks by mixing dry WTS and calcined rice husk, which showed
182 improvements in mechanical properties. **In addition, several other studies have demonstrated**
183 **successful application of waste materials including WTS with a high concentration of silica in**
184 **making ceramics and bricks with good mechanical performance** (Hegazy et al., 2012; Wolff et
185 al., 2015; Ewais et al., 2017).

186 In general, the addition of aluminum-based sludge decreases compressive/flexural strength
187 and increases the water absorption of the ceramic bodies. According to the majority of the
188 studies summarized in **Table A2** (Huang et al., 2001; **Teixeira et al., 2011**), the replacement of
189 clay by aluminum-based sludge below 10% did not significantly reduce the mechanical
190 properties of the ceramic bodies, although above 10% a reduction of 24.6-45.45% in flexural
191 strength was reported. On the other hand, iron-based sludge increased the compressive/flexural
192 strength of the ceramic samples by 7-97% compared to the reference values, for sludge addition

193 below 10% (Kizinievič et al., 2013; Hassan et al., 2014). Furthermore, studies have shown that
194 it is possible to produce ceramic bricks with mechanical properties similar to those of clay, with
195 more considerable additions (> 50%) of sludge, combined with material with high silica
196 concentration (Lin and Lin, 2005; Chiang et al., 2009; Hegazy et al., 2012; Wolff et al., 2015;
197 Ewais et al., 2017).

198

199 3.2. Lightweight aggregates

200 Lightweight aggregates (LWA) are highly porous and spherical ceramic products with low
201 density (0.8-2.0 g/cm³) and commonly used in the manufacture of various building products
202 (Soltan et al., 2016). **Figure A1** shows the manufacturing process of traditional LWA from
203 clay. The use of residual waste materials in the production of LWA has been extensively
204 exploited in recent years, mainly because LWA has wide applications in construction materials
205 such as lightweight concrete/mortar (structural and non-structural), water treatment process and
206 gardening (Dondi et al., 2016). A significant number of studies with sludge and industrial waste
207 have used the thermal stabilization process to produce composites LWA, immobilizing the
208 contaminants e.g. heavy metals through the sintering process in the matrix of the new materials,
209 and thus reducing the migration of contaminants into the environment (Huang et al., 2007;
210 Wang et al., 2008; Corrochano et al., 2011; González-Corrochano et al., 2017).

211 3.2.1. Lightweight coarse aggregate

212 Huang et al. (2005) used the aluminum-based sludge from a Taiwanese WTP to produce
213 lightweight concrete coarse aggregates. Spherical samples of approximately 2 cm in diameter
214 were produced, which were oven dried at 105 °C for 24 h and sintered at 1000, 1050 and 1100
215 °C. The WTS-derived LWA had water absorption and specific gravity of 14.47-37% and 1.12-
216 1.78 g/cm³, respectively. After composite production, the LWA was applied to a concrete mix
217 with proportions of 2:1:1 (Natural sand: WTS-LWA: cement). The results, when compared to
218 the Taiwanese construction standard, showed that only the WTS-aggregates sintered at 1100

219 °C met the requirements of specific gravity (1.4-2.0 g/cm³) and compressive strength (175-420
220 kgf/cm²) of the concretes made with natural LWA.

221 Huang and Wang (2013) investigated the alum-based sludge from 10 WTPs in Taiwan to
222 produce lightweight coarse aggregates (**Figure A2**). By using air-dry sludge mixed with an
223 adequate amount of water which was then extruded to **produce cylindrical pellets of 8-12 mm**
224 **in diameter**, they obtained samples which were sintered at 500 °C for 5-15min and then 1150-
225 1275 °C for 7.5-15 min. They **reported satisfactory results** with LWA concrete made with WTS,
226 five were classified as viable for structural and non-structural LWA production, the other five
227 only for structural, all with similar properties of high-quality LWA, with densities of 0.65-2.05
228 g/cm³ and water absorption of 0.5-15%. Moreover, the researchers produced LWA in large
229 scale using a rotary kiln with sludge from one of the WTPs. **The results demonstrated that it**
230 **was possible to produce structural and non-structural LWA with average particle density of**
231 **1.35 g/cm³ and bulk density of 726 kg/m³ meeting the requirements of the American Society**
232 **for Testing and Materials (ASTM, 2017b).**

233 In a different approach, Sales et al. (2010; 2011) produced a composite lightweight coarse
234 aggregate for concrete with alum-based WTS and sawdust. For the lightweight composite
235 production, dry and milled WTS was homogenized with water and sawdust, in a ratio of 6: 4.5:
236 1, to generate rounded samples with a diameter of 14±2 mm. After molding, the samples were
237 dried at 105 °C and then immersed in boiled linseed oil for 1 min. Then the samples were dried
238 at room temperature and mixed in different concretes batches. Based on the **ACI (2014)**
239 classification, the concrete produced with the lightweight composite was in the non-structural
240 lightweight concrete category, showing an axial compression strength of 11.1 MPa and an
241 **apparent density of 1847 kg/m³.**

242 3.2.2. *Ceramsite*

243 Ceramsite is a type of fine LWA that has been used as construction materials in a variety of
244 applications, such as concrete composites, bricks, wetlands layers and filters for wastewater

245 treatment process (Xu et al., 2008b; Jia et al., 2014). Xu et al. (2008a) tested the feasibility of
246 ceramsite production with alum-based WTS and wastewater treatment sludge (WWTS). For
247 the manufacture of ceramsite, different proportions of WTS, WWTS and water glass (sodium
248 silicate) were used to obtain pelletized samples (5-8 mm) which were maintained at about 20
249 °C for 3 days and then pre-heated to 24 h at 110 °C. The heating ramp started at 20 °C and
250 continued at a rate of 8 °C/min in a muffle until reaching 1000 °C for 35 min. The results
251 showed that the optimal values for the manufacture of ceramsite were: WTS/WWTS = 45/55,
252 sodium silicate/(WTS+WWTS) = 20%, SiO₂ = 14-26%, and Al₂O₃ = 22.5-45%, at a sintering
253 temperature of 1000 °C. Under these conditions, the compressive strength of ceramsite ranged
254 from 13.63 MPa to 16:25 MPa, above the minimum (7.50 MPa) required by the National
255 Standards in China. In a similar research, Zou et al. (2009) investigated the effect of different
256 proportions of Fe₂O₃, CaO and MgO oxides on the production of ceramsite with WTS and
257 WWTS. The methodology used was the same as used by Xu et al. (2008a), however varying
258 and adjusting the amount of iron, calcium, and magnesium oxide. The variation of Fe₂O₃
259 content directly influenced the mechanical strengths of ceramsite mixtures. Fe content of 5-8%
260 was considered optimal for lower water absorption and higher compressive strength which
261 ranged from 14.97 MPa to 15.67 MPa. The mechanical strength of ceramsite was not influenced
262 directly by MgO, but an increase in the bulk density was noticed as MgO was increased,
263 resulting in an ideal rate of 1.6-4.0%. The increase in CaO amount decreased the compressive
264 strength of ceramsites, and an optimal value of 2.75-7% was proposed.

265

266 3.3. Cement production

267 Global Portland cement and concrete production (Figure A3) is a technological activity of
268 intense demand for natural resources. The raw materials commonly used in the manufacture of
269 cement are calcium carbonate (75-80 wt.%) and clay (20-25 wt.%) (Bignozzi, 2011). Due to
270 the mineral and chemical similarity of Portland cement and WTS, alternatives are being sought

271 to incorporate WTS into cement production process or as a supplementary cementitious
272 material.

273 Tay and Show (1991) used an iron-based WTS mixed with different proportions of lime
274 powder (CaCO_3) and sintered at a temperature of 1000 °C for 4 h in three different blends (1:3,
275 1:1 and 3:1 by weight). The results indicate that cement made from Fe-sludge, with a sludge-
276 lime mixing ratio of 1:1, **sintered at 1000 °C for 4 h**, can be used for general masonry work
277 according to **ASTM**. On an industrial scale, Baker et al. (2005) replaced 15% of **limestone** for
278 lime sludge (Ca-sludge) in a cement plant and produced around 80 tons of composite cement.
279 The properties of the new cement were satisfactory in comparison to the **ordinary Portland**
280 **cement**. Similarly, Pan et al. (2004) evaluated the possibility of replacing clay used in **Portland**
281 **cement** clinker production by using alum-sludge. The cement containing WTS as a substitute
282 of clay showed adequate results regarding the setting time of cement paste. The **compressive**
283 **strength of concrete was** increased as the sludge was incorporated, meeting the **Chinese national**
284 **standard requirements** for first grade **Portland cement**.

285 Lin and Lin (2005) used WTS ash combined with two more industrial by-products for
286 replacement of **Portland cement** raw materials. The results regarding compressive strength and
287 microstructure of cement paste showed that the substitution up to 20% of these materials was
288 feasible for the manufacture of clinker cement. Moreover, all hydrated compounds commonly
289 found in **ordinary Portland cement** paste [$\text{Ca}(\text{OH})_2$ and C-S-H] were detected after the addition
290 of these residues. Chen et al. (2010) studied the feasibility of replacing the siliceous raw
291 material (Shale) by using alum-sludge from **4%** to 10% for cement production. All specimens
292 with WTS had higher 3-day and 7-day strength than the control specimens, and also superior
293 percentage of tricalcium silicate (C_3S). However, after 28-day **only samples with sludge**
294 **additions** below 7% had higher strengths than the control. In more recent studies, Dahhou et al.
295 (2016; 2018) produced several clinker compositions with partial limestone replacements by
296 alum-sludge and sintering temperature at 1300-1500 °C. Regarding the flexural and

297 **compressive strength** of the cement pastes burned from 1450 and 1500 °C, all clinker
298 compositions showed superior performance and also identical mineralogical values compared
299 to ordinary Portland **clinker**, with a significant presence of alite (C₃S) and belite (C₂S).

300 The production of cement with the partial addition of WTS has been shown to be potentially
301 **feasible**. Most of the studies focused on cement paste parameters such as mechanical strength,
302 setting time, chemical and mineralogical characterization and some leaching tests. **However,**
303 **due to the lack of microstructure and long-term durability testings on concrete and the presence**
304 **of some substances in sludge such as chloride and sulfate ions which are potentially deleterious**
305 **to concrete, it is unclear how the use of WTS will affect the microstructure and long-term**
306 **performance of concrete. For example, in this case whether the expansive products (such as**
307 **ettringite) will be formed, and whether corrosion in the bars of reinforced concrete will occur.**
308 **More studies should be conducted to assess the potential negative impacts of chloride and**
309 **sulphate ions on the long-term performance of sludge-amended cement materials, in terms of**
310 **structural integrity and mechanical properties.**

311

312 3.4. Supplementary cementitious material and inert addition

313 **Generally, supplementary cementitious materials (SCMs) are classified into two categories:**
314 **self-cementing and pozzolanics.** Self-cementing materials react similarly to hydraulic cement,
315 **as they set and harden in the presence of water, forming solid cementitious products (Snellings**
316 **et al., 2012). In comparison, pozzolanic materials (siliceous and aluminous composition) only**
317 **hydrate and form cementitious compounds in the presence of moisture and an activator, usually**
318 **calcium hydroxide (ASTM, 2017a).**

319 3.4.1. Self-cementing

320 **WTS has a potential to be used as a low-cost internal curing agent for concrete (Nowasell and**
321 **Kevern, 2015).** El-Didamony et al. (2014) investigated the substitution of a SCM, granulated
322 blast furnace slag (GBFS), by alum sludge in proportions of 5%, 10% and 15% by weight for

323 the manufacture of composite hydraulic cement (**Table 2**). The results showed that free
324 portlandite increased in the first 7 days of curing and then decreased at 90 days. Furthermore,
325 the replacement of GBFS by WTS led to **an increase** in the chemically combined water and
326 hydration products. Compressive strength was increased by up to 5% of GBFS replacement and
327 decreased with sludge increasing up to 15% by weight. Dahhou et al. (2018) investigated the
328 partial replacement of PC (class CPJ55) by various amounts (5-25%) of dry Alum sludge in
329 mortar samples with dimensions of 40 × 40 × 160 mm. They observed that the addition of 5%
330 of WTS in PC did not affect the mineralogy of the final product. However, based on the flexural
331 and compression tests at 28 days, mortars with 5% sludge substitution were classified as
332 belonging to class 32.5 R according to the Moroccan Standard.

333 *3.4.2. Pozzolanic addition*

334 Rodríguez et al. (2010) used the atomized-dry WTS (Aluminium-based) as a pozzolanic
335 addition in the cement mortars (**Table 2**). The mortars were prepared with the substitution of
336 10-30% PC by atomized sludge and tested in relation to hydration, water demand, setting time
337 and mechanical strength. The cement replacement by the sludge considerably delayed the rates
338 of cement paste hydration and setting time, even in the samples with only 10% of sludge.
339 Mortars with 10-30% of atomized alum-sludge showed a substantial drop in flexural strength,
340 with a decrease of 30% and 45% respectively. Furthermore, FTIR spectra testing showed the
341 formation of amorphous ettringite for the mortars mixes with atomized sludge. Alqam et al.
342 (2011) investigated the use of an iron-based WTS as a partial replacement of pozzolanic
343 **Portland cement** in the production of paving tiles for external use. The sludge was added in
344 10%, 20%, 30%, 40% and 50% (by cement weight), in making samples of 400 mm × 400 mm
345 × 35 mm. **The results showed that sludge addition did not affect water absorption of the samples,**
346 **with an increase of only 6% with the maximum sludge replacement (50%).** Except for samples
347 with 50% sludge, all others obtained a minimum breaking strength of 2.8 MPa for external tiles
348 according to the British Standard. Nevertheless, it was observed there was a decrease in

349 mechanical strength ranging from 1.5% to 28%, when the sludge addition was increased from
350 10% to 50%.

351 Al-Tersawy and El Sergany (2016) compared the pozzolanic properties of calcined ferric-
352 sludge and rice husk ash (RHA) in the partial substitution of cement in several concrete
353 compositions, by ranging from 0% to 30% with two different types of aggregates (dolomite and
354 gravel). Regarding the mechanical properties, the substitution of 10% of RHA obtained better
355 results, with compressive and tensile strength higher than the control sample. However, the
356 same was not observed for the replacement of calcined sludge, which obtained reductions in
357 mechanical strength ranging from 30% to 62% after 28 days of age. The study concluded that
358 the calcined sludge can be used in percentages less than 10% as a filler material for concrete
359 and still reach the minimum limits for structural concrete in Egypt (ES:1524/993). In a recent
360 and similar study, Ahmad et al. (2018) studied the feasibility of partial replacement of Portland
361 cement by an Al-based sludge from the backwashing filtration beds, calcined at 800 °C for
362 incorporation in cement mortar and concrete. Concerning the pozzolanic activities of the sludge,
363 the results showed that, as well as traditional pozzolanic materials, the calcined sludge reacted
364 with $\text{Ca}(\text{OH})_2$ and generated significant quantities of hydrated products, and could therefore be
365 classified as an artificial pozzolana. The addition of sludge had a small influence at the initial
366 setting time, however, regarding the final setting time an increase of 2.6-40% was observed.
367 The compressive strength of the mixes decreased as the sludge was added. The results showed
368 that the replacements up to 20% could meet the Indian Standard for the paste made by Portland
369 pozzolana cement (BIS, 2015).

370 3.4.3. Sand replacement

371 Hoppen et al. (2005) evaluated the partial replacement of sand by wet Al-based sludge in
372 concrete mixtures. Four compositions of concrete-sludge were prepared with replacements of
373 3%, 5%, 7% and 10% based on the weight of the fine aggregate. The amount of mixing water
374 in each mixing was adequate regarding the weight and moisture content in the sludge. The

375 results indicated that 10% of sludge in the concrete is a limiting content for its practical
376 application, first due to low workability, as well as low mechanical strength, being less than 15
377 MPa (**Figure 2**). The substitutions of 4-8% of sand by wet sludge in the concrete resulted in
378 compressive strength being higher than 27 MPa at 28 days. Thus, it is suggested that the
379 possible applications for this type of concrete should be in non-structural applications such as
380 subfloor, blocks, non-load bearing walls, decorative concrete pieces, sidewalks, residential
381 floors, among others. Using a similar approach, Fernandez et al. (2018) obtained results that
382 supported Hoppen et al. (2005).

383 Tafarel et al. (2016) also partially replaced the natural sand present in the concrete by a wet
384 aluminum-based sludge, in proportions of up to 10% of the dry weight of the sand. Considering
385 the results, only the samples with 5% sludge substitution showed satisfactory compressive
386 strength performance of 15.5 MPa at 28 days, a decrease in the strength close to 11% when
387 compared to the reference concrete. The incorporation of 5% and 10% sludge led to an increase
388 in water absorption by 12% and 32%, respectively. In similar research, Gomes et al. (2017)
389 investigated the effects of aluminum-based sludge in its natural (wet) form, ranged from 0, 5, 7
390 and 10% of sand replacement in concrete. The results showed that the addition of wet sludge
391 reduced mechanical strength and increased water absorption, as even 5% of sludge substitution
392 led to a 50% reduction in compressive strength and 45% increase in water absorption.

393 Based on the studies presented, the substitution of natural sand for the production of
394 structural and non-structural concretes is viable and satisfactory. The replacement of fine
395 aggregate by up to 5% WTS in concrete and mortar led to a relatively small reduction in
396 compressive strength (less than 20%) compared with no sludge material (**Figure 2**). Thus, 5%
397 substitution of fine aggregate is a critical value that could be considered safe for sludge
398 application in structural or non-structural concretes according to the Brazilian and Chinese
399 codes (ABNT, 2014; CCES, 2005). However, further studies need to be performed to verify the
400 influence of sludge addition on water demand, hydration of PC, and the durability of concrete.

401 In addition, the leaching and safety tests according to specific technical standards should be
402 conducted to ensure the effectiveness of sludge applicability while minimizing potential
403 damages to nature and human health.

404

405 3.5. Geopolymers

406 Geopolymers were designated in analogy to the raw materials (geological elements) used in
407 their manufacture. While carbon structures form conventional polymers, geopolymers require
408 source materials of Si and Al (Precursor), water and an alkaline reagent (activator) (Figure A4),
409 which is responsible for triggering the polymerization of the components (Davidovits, 1991).
410 Geopolymers are considered sustainable materials, mainly because both industrial wastes and
411 by-products such as fly ash, metallurgical slag and mining residues, and geological materials
412 such as activated clays, zeolites and kaolin can be utilized for their production (Phair and Van
413 Deventer, 2002; Komljenović et al., 2010; Salwa et al. 2013; Singh et al. 2015; El-Eswed et al.,
414 2017; Sudagar et al. 2018). Therefore, there have been studies on the potential of WTS as a
415 geopolymeric feedstock, since WTS is usually rich in Si and Al oxides.

416 Waijarean et al. (2014) used aluminum-based WTS as alumino-siliceous material to
417 synthesize geopolymers, aiming for their application in heavy metal waste immobilization. The
418 results showed that the non-calcined sludge did not show compressive strength in the early ages,
419 and after 60 days of curing achieved 0.76 MPa. The XRD profile showed that the main reason
420 for this phenomenon was the non-occurrence of dehydroxylation of halloysite. The geopolymer
421 samples calcined at 600 and 900 °C presented a lower compressive strength than those calcined
422 at 800 °C achieving 8.8 MPa after 60 days of curing.

423 Suksiripattanapong et al. (2015a) researched the mechanical performance of masonry units
424 made with WTS-fly ash geopolymer, using air-dried alum sludge as aggregate, fly ash as
425 precursor combined with alkaline activation mixture based on NaOH and Na₂SiO₃. The
426 production of alkaline activator liquid used a fixed composition (by weight) of 9% Na₂O, 30%

427 SiO₂, and NaOH (10 M). The ratio of WTS/fly ash was 70:30, while the ratio of Na₂SiO₃/NaOH
428 varied at 100:0, 90:10, 80:20, 70:30 and 50:50. Furthermore, the samples were heated at
429 temperatures of 65, 75 and 85 °C, lasting up to 120 h and then cured at room temperature (27-
430 30 °C). The best formulation results for maximum unit weight and compressive strength were
431 at Na₂SiO₃/NaOH ratio of 80:20 and the alkaline activator/fly-ash ratio of 1.3. The best results
432 were found at the temperature of 75°C and heat period of 72 h, generating samples with
433 compression strength of 12-20 MPa. Therefore, WTS was considered feasible for the
434 production of masonry units according to the Thai standard. Suksiripattanapong et al. (2015b)
435 also investigated the use of the same by-products, WTS and fly ash as aggregates and precursor,
436 respectively, for the production of a cellular geopolymer without using PC as a cementing agent.
437 A foaming agent based on anionic surfactants, foam stabilizers and liquid air entraining were
438 used to reduce the unit weight. The parameters used were the same as the previous study with
439 masonry units and the influential factors studied were air content, the liquid content of alkaline
440 activator and duration of heat and cure time. The higher compressive strengths (7.5-19 MPa)
441 with different air contents were obtained at Na₂SiO₃/NaOH ratio of 80:20 and a heat duration
442 of 72 h at 65 °C, which is associated with the growth of geopolymerization.

443 Nimwinya et al. (2016) studied the feasibility of a geopolymer precursor with calcined
444 mixtures of WTS and RHA. Calcined aluminum sludge and RHA powder were mixed in
445 various WTS/RHA ratios of 100:0, 85:15, 70:30, 60:40 and 50:50. The alkaline activator
446 solution was based on the NaOH mixture with calcium silicate (Na₂SiO₃) with fixed proportions
447 of Na₂O (8.0%) and SiO₂ (27.0%). A delay in the settling time was observed as the SiO₂/Al₂O₃
448 ratio was increased, mainly because of decrease in the WTS/RHA ratio. Thus, the initial settling
449 time was increased with increasing RHA content, reaching 13.5 h for mixture with 50% RHA.
450 The higher 7-day compressive strength (16-24 MPa) of the geopolymer paste cured at room
451 temperature and 60 °C was reached when the ratio of SiO₂/Al₂O₃ was 4.9 and 5.9 (30% and
452 40% RHA replacement, respectively) (Table 3).

453 Geraldo et al. (2017) investigated the partial substitution of metakaolin (MK) by WTS for
454 the composition of geopolymeric mortar. An alternative alkaline activator solution was
455 prepared with NaOH and RHA (replacing sodium silicate). Furthermore, due to the amount of
456 SiO₂ being lower in the WTS compared to the MK, an extra amount of RHA was added in the
457 mixtures to balance and increase the SiO₂/Al₂O₃ ratio. The results showed that as the proportion
458 of WTS was increased, setting time increased simultaneously. The final settling time of the
459 samples ranged from 3.22 to 5.32 hours; which were higher compared to fly ash geopolymers
460 for the same temperature. The compressive strength and the workability of the samples
461 decreased as the addition of WTS. Mechanical strengths at all curing ages were higher than the
462 minimum required for various types of building components, according to the Brazilian
463 Standard (clay-fire brick > 1.5 MPa, soil-cement brick > 2.0 MPa, concrete block > 3.0 MPa).

464 Duxson et al. (2005) and Tang et al. (2019a) showed with aluminosilicate-base that the
465 strength development of geopolymer matrix depended on the type of precursor and SiO₂/Al₂O₃
466 ratio. Thus, these few studies with WTS as a precursor have focused on improving and
467 modifying the SiO₂/Al₂O₃ ratio when it was not satisfactory. Therefore, considering the recent
468 research with WTS, the production of geopolymers with WTS would be a feasible solution to
469 convert such solid waste to construction materials. However, for WTS-geopolymer to be
470 competitive and alternative for PC-based materials, more studies need to be carried out to
471 examine the structural performance of such geopolymers under realistic environmental
472 conditions such as different pH, ambient temperature, humidity, alkalinity, and salt content, so
473 as to identify potential relationships and key controls. The results obtained will provide
474 guidance in designing further formulations such as geopolymer combinations with
475 superplasticizers, handle retarders, polymers, natural and synthetic fibres and different alkaline
476 activators, under different setting and curing time, so as to obtain the best products in terms of
477 mechanical performance.

478

479 **4. Durability**

480 Natural deterioration causes decreased performance in construction materials; however, it is
481 often considered a second or forgotten topic during the design phase (Bijen, 2003). The primary
482 tests for the durability of materials incorporating WTS are discussed, based on studies which
483 have been carried out on the long-term life cycle.

484 The durability of ceramic bricks depends primarily on water absorption (Huang et al., 2001;
485 Hegazy et al., 2012; Ewais et al., 2017). According to Fernandes et al. (2010), the deterioration
486 of ceramic bodies is mainly due to the high capacity of the fluid to be stored in bricks, causing
487 the reduction of mechanical resistance and useful life. The major causes of the increased water
488 absorption in the bricks were the burning of the organic matter present in WTS between 300-
489 600 °C and the decomposition of carbonates and sulfates at temperatures above 800 °C, which
490 eventually contributed to the increase in porosity (Teixeira et al., 2011; Kizinievič et al., 2013;
491 da Silva et al., 2015). The primary test used to measure water absorption is quite similar in most
492 of the international standards used, the majority of which limit the value of water absorption
493 for load-bearing bricks at no more than 15-25% by weight (BIS, 1992; ABNT, 2015; ASTM,
494 2018). The increase in water absorption was noticeable as the sludge was added to the ceramic
495 bodies. Among the various studies, the vast majority of which obtained water absorption values
496 below 20%, except where the concentration of organic material in the sludge was very high
497 (Huang et al., 2001; Teixeira et al., 2011). In addition, according to Raut et al. (2011), in their
498 review on ceramic bricks with addition of residues and waste materials, accelerated weathering
499 tests should also be conducted to ensure a long-term cycle for the material.

500 LWA and ceramsite are clay-sintered materials which exhibit similar behavior as the
501 ceramic bricks. The water absorption, porosity and recrystallization of the liquid CaCO_3 are the
502 predominant factors adversely affecting the durability of the raw materials and the light
503 concrete produced (Huang et al., 2005; Zou et al., 2009). Huang et al. (2005) showed that among
504 the three firing temperatures (1000, 1050 and 1100 °C) used to produce LWA made from

505 sludge, only the specimens produced at 1100°C obtained low water absorption (14.47%),
506 similar to commercial LWA and therefore feasible for application in concrete. Sales et al.
507 (2010) showed that due to the high water absorption of the LWA made with WTS, all concrete
508 formulations had relatively high porosity and water absorption (8.8%) compared to
509 conventional concrete, which can affect the long-term life of the material depending on its
510 exposure and application (Figure A5). Zou et al. (2009) demonstrated that the chemical
511 durability of ceramsite is better when there are more complex crystalline phases and few pores.
512 Through the tests of X-ray diffraction, morphological structures analysis, water absorption and
513 porosity, it was confirmed that ceramsite characteristics are mostly influenced by Fe₂O₃ content,
514 which should ideally be between 6% and 8%.

515 Concrete is the most commonly used construction material in the world, and its
516 combination with steel makes it a material with broad applications. However, their durability
517 performance depends on their interaction with the environment, where the penetration of
518 substances through the pores has a significant impact. The most common forms of deterioration
519 of concrete are corrosion of the steel bars through the carbonation and ingress of chlorine ions,
520 freezing and thawing action, sulfate attack and alkali aggregate reaction (Basheer et al., 2001;
521 Tang et al., 2019b). Therefore, the additions of sludge in the concrete must be evaluated for
522 durability, mainly because sludge is a by-product with the presence of organic matter, chloride
523 and sulfates ions, and other deleterious elements such as heavy metals (Cremades et al., 2018;
524 Ooi et al., 2018). In studying the application of WTS in cement mortar, Rodríguez et al. (2010)
525 showed, by FTIR spectroscopy test, that the presence of 12-14% organic matter (mainly fatty
526 acids) had influenced the formation of ettringite and retardation of calcium silicate hydration
527 process, affecting the long-term durability of concrete. In studying the addition of sludge to
528 concrete, Hoppen (2005) reported that sludge additions above 5% caused a considerable
529 increase in water absorption, making the material less durable and more vulnerable to
530 penetration of chloride and sulfate ions. The concrete durability tests by accelerated aging in

531 alkaline solution (ASTM, 2015a) with 4% of WTS showed results of potential corrosion similar
532 to the reference concrete. However, the specimens with 8% of sludge showed a tendency of
533 potential corrosion with 90% probability of occurrence, but no cracks or other defects were
534 found in the concrete surface.

535 Concerning sulfate attack, Suksiripattanapong et al. (2015b) showed that geopolymers
536 based on clay-fly ash had greater durability than clay-cement materials. Using the methods for
537 wetting and drying cycles (ASTM, 2015b), Horpibulsuk et al. (2016) showed that wet-dry
538 durability of WTS-fly ash geopolymer after 12 cycles was higher than that of the WTS-cement
539 and silty clay-cement under the same conditions. Even after 12 wet-dry cycles, WTS-FA
540 geopolymer with better heat curing condition (85 °C for 72 h) showed a satisfactory
541 compressive strength of 7 MPa, compatible with durable bearing masonry blocks. Using the
542 water absorption, voids and capillary absorption tests (ABNT, 2014), Geraldo et al. (2017)
543 showed that the air permeability of the geopolymer with 60% substitution of metakaolin by
544 WTS was five times higher than that specimens with no sludge addition. The porosity of the
545 material was increased as the sludge was added, and the water absorption values were between
546 10.6% and 13.7%. The voids varied from 9.6% to 12%, suggesting that the sludge addition in
547 high proportions affected the long-term behavior of the material. Up to now few studies focused
548 on the aspect of durability and long-term effects of the addition of sludge residue. Therefore,
549 further research is needed to assess the long-term performance of sludge-amended concrete
550 under a range of environmental stresses such as freeze-thaw deterioration, chloride ingress,
551 sulfate attack, and carbonation.

552

553 **5. Leachability and toxicity**

554 Considering the high potential of applying WTS in construction materials, environmental risk
555 assessment becomes an important issue, since WTS may contain heavy metals with potential
556 consequences for the environment and human health. The most common tests for this type of

557 evaluation are called leaching tests, which evaluate the release of contaminants to the
558 environment in the presence of water, in the different phases of the product life cycle
559 (manufacturing, distribution, construction, use, end of life) and not only in aspects of 'pass fail
560 tests' (Shih and Tang, 2011; Watanabe et al., 2011). When a leaching test is selected for material
561 analysis, the shape of the building material tested should be taken into consideration, because
562 the release mechanisms of heavy metals involved are different. For example, in monolithic
563 elements, heavy metals will be released due to surface washing, diffusion and dissolution
564 processes (NEN, 2004; Król, 2011). The most widely test applied to materials and waste is the
565 Toxicity Characteristics Leaching Procedure (TCLP) from the United States Environmental
566 Protection Agency (USEPA). According to Townsend et al. (2003), many chemical and
567 physical factors control waste leaching in products. The physical factors are particle size,
568 liquid/solid reaction, temperature and porosity. Chemical factors include pH, the redox
569 potential of the material, sorption processes and formation of complexes with organic or
570 inorganic compounds. Considering the large amount of literature evaluated in this review, only
571 limited studies have assessed the environmental risk caused by the incorporation of various
572 types of WTS into building materials (Table 4). This may be due to the fact that in most cases
573 the sludge is classified as non-hazardous waste according to local standards (Hidalgo et al.,
574 2017).

575 In a study with arsenic-iron sludge for the production of bricks, Hassan et al. (2014)
576 reported that for different pH only sludge-mix ratio of 3% showed less arsenic and iron leaching
577 than was acceptable by the Bangladesh and World Health Organization (WHO) Standards. The
578 highest values for leaching of the burned brick were found at pH around 3.0, showing values
579 above the limit values of 0.05 mg/l for arsenic. Sales et al. (2010; 2011) studied the leaching of
580 aluminum, iron and lead in a LWA produced with Al-based WTS and sawdust. The solubilized
581 extract of the concrete produced with the WTS-LWA showed an aluminum concentration
582 (19.96 mg/l) which is substantially higher than the solubilized extract concentration of the

583 reference concrete without this composite (1.12 mg/l). The result was mainly due to the high
584 concentration of aluminum (11.10 mg/l) present in the sludge. Nevertheless, the material
585 showed the necessary limits for environmental safety according to the NBR 10004 Brazilian
586 standard (ABNT, 2004) and classified as non-harmful and non-inert solid wastes. Xu et al.
587 (2008c; 2010) studied the stabilization of heavy metals in ceramists made with Al-WTS and
588 WWTS. Leaching tests focused on the effects of pH, oxidative condition, sintering temperature
589 and the ratio of oxides $(\text{Fe}_2\text{O}_3 + \text{CaO} + \text{MgO})/(\text{SiO}_2 + \text{Al}_2\text{O}_3)$. The effect of sintering
590 temperature and the ratio of oxides had a great influence on the leaching and solidification,
591 noting that above 1000 °C heavy metals such as Cd, Cr, Cu, and Pb can hardly be re-released
592 to the environment causing secondary pollution. The results also show that sample-ceramsites
593 exposed to conditions of $\text{pH} > 2$ had leaching rates below those recommended in the Chinese
594 Standard and TCLP. Similar findings were reported by Wei (2015) that the heavy metal
595 solidifying efficiencies of sludge-derived LWA were strongly enhanced by crystallization and
596 chemical incorporations within the aluminosilicate or silicate frameworks during the sintering
597 process. Therefore, it is concluded that heavy metals can be properly stabilized in LWA samples
598 containing sludge and prevented from release into the environment again to cause secondary
599 pollution.

600 When studying eco-cements clinkers made with two industrial waste and Al-based WTS
601 ash, Lin and Lin (2005) analyzed the concentrations and leachings of heavy metals Pb, Zn, Cd,
602 Ni, Cr and Cu. The results confirm a high stabilization rate of heavy metals, and all eco-cement
603 mixtures were below the regulatory limits established by TCLP. Similar analysis was made by
604 Chen et al. (2010) in their study of shale replacement by WTS for clinker production. The TCLP
605 results showed that for the hydrated samples made with the WTS-clinker, no heavy metal was
606 detected above the normalized standard, indicating that these components were fixed and
607 incorporated to the clinker. Lee et al. (2012) analyzed the leaching effects of the substitution of
608 natural sand by an Al-based WTS in the production of concretes, and the results of TCLP

609 **analysis showed** that no substantial quantities of heavy matter were detected in the concrete
610 samples. Proving that, even when large quantities of sludge (30%) were added, the heavy metals
611 were fixed and immobilized in the concrete matrix. Using TCLP, Alqam et al. (2011) evaluated
612 the heavy metals leaching of paving tiles made from the mixture of Al-sludge and cement. All
613 results showed very low percentage of heavy metal concentrations, even for paving tiles with
614 50% Al-WTS contents.

615 **Wajjarean et al. (2017) evaluated the leaching of Zn, Fe and Cr in geopolymers synthesized**
616 **by alkali activation of WTS. Leaching tests in simulated acid rain indicates that the**
617 **geopolymers with molar ratios of $\text{SiO}_2/\text{Al}_2\text{O}_3 = 1.78$ were capable of immobilizing all three**
618 **metals to within safe leachate levels in accordance with USEPA regulatory limits. It was**
619 **observed that as the $\text{SiO}_2/\text{Al}_2\text{O}_3$ ratio was increased, the WTS-based geopolymer became less**
620 **effective as an immobilizing agent. For Cr (the most toxic of the 3 metals), the results simulating**
621 **acid rain (pH 5) showed that at $\text{SiO}_2/\text{Al}_2\text{O}_3$ ratio of 3 Cr concentration in the leachate reached**
622 **8.9 mg/l, exceeding the USEPA recommended value of 5 mg/l.**

623

624 **6. Economic evaluation**

625 In their research on WTS application in structural ceramics, Wolff et al. (2015) showed that for
626 every ton of clay, it would be possible to save US\$318 by introducing sludge into the brick
627 production cycle. Included in this amount is a saving of US\$8/ton that would be spent in
628 transport to the landfill and other cost associated with clay extraction and energy. Kaosol (2010)
629 evaluated the cost involved in the production of hollow concrete blocks with fine aggregate
630 replacement up to 50% by using WTS in Thailand. The cost composition for the evaluation was
631 based on the production of 1200 blocks day, 2 laborers, machine cost, electricity and materials
632 for normal block production (cement, sand and aggregate). In their evaluation, for the
633 replacement of 10% and 20% of fine aggregate by WTS, the cost per unit would be reduced by
634 Thai-Baht ฿0.64 and ฿1.05 respectively, for a hollow load bearing concrete block. Furthermore,

635 for the manufacture of non-load bearing blocks with 30-50% of sludge, the total costs would
636 be reduced by Thai baht ฿1.48-2.35 per block.

637 **Lima and Zulanis (2016)** performed an economic feasibility analysis for the incorporation
638 of 5% WTS in the concrete mix sidewalks in Cubatão city, Brazil. The analysis considered the
639 Cubatão city's **dry-sludge production of 1925 kg/day** and a standard city block of 80 m × 274
640 m (1534.4 m³ for sidewalk of 7 cm thickness). According to their analysis, it would need three
641 days of sludge production to supply the required 5% addition in the 98-m³ concrete mix for 700
642 m of a sidewalk, and 39 days of water filtration process to achieve enough WTS to lay a city
643 block of concrete. Thus, in 1 year with a production of 9.4 city blocks of sidewalks (365 days
644 in a year/39 days to produce sufficient sludge for a city block), the city could consume all its
645 sludge production of the year. In addition, according to the researchers, the local civil
646 construction industry could save \$543069/year with the assumption that 1 ton of concrete gravel
647 (washed rock) is \$28.81 with 5% WTS substitution. Gastaldini et al. (2015) analysed the costs
648 (US\$56.04/m³) of collecting/transporting the sludge from water purification plant to the
649 landfill, and the fees applied to allocate the sludge in the Rio Grande do Sul, Brazil. They
650 estimated that the cost to carry 1 kg of sludge to the landfill was twice as the cost of a 19-mm
651 coarse aggregate (US\$0.018) and 5.6 times the cost of washed and sieved sand (US\$0.0083).
652 Therefore, the 5-30% substitution of cement by WTS ash (for the same concrete strength
653 without the addition of sludge) would lead to savings of 37-200 kg of cement per m³ of concrete.

654 **In evaluating the production of geopolymer binders with WTS, Nimwinya et al. (2016)**
655 **showed that the cost involved would be twice as practiced today in Thailand which is**
656 **US\$33/ton, when WTS is transported to landfill sites (although actual landfill cost is not**
657 **included). This relatively high cost is mainly related with the geopolymers ingredients,**
658 **calcination of the sludge and the electricity used in the curing process, which stands at US\$65**
659 **and US\$68 for geopolymers prepared at room temperature and cured at 60 °C, respectively.**
660 **However, such analyses have not considered the significant savings in CO₂ reduction, when**

661 WTS is replacing natural raw materials in construction material production. Furthermore, more
662 financial gains can be obtained by the recovery of metals e.g. Al from WTS (Ooi et al., 2018),
663 and the sales of geopolymers (Nimwinya et al., 2016).

664

665 **7. Conclusions**

666 This paper critically reviewed the research progress in the application of WTS in construction
667 materials. Some conclusions can be drawn up as follows:

668 ● The latest research results on the application of WTS in a range of construction materials
669 were critically examined. In summary, WTS has some unique characteristics to enable its
670 incorporation in construction materials such as bricks, ceramics, LWA, cement and
671 geopolymers.

672 ● The main challenges in the application of WTS in construction materials are derived from
673 its high variation in physicochemical properties and relatively high content of organic
674 matter, which will increase the porosity and water absorption therefore potentially
675 adversely affecting products' structural integrity and performance.

676 ● The majority of studies with ceramic bricks showed that the incorporation of alum sludge
677 up to 10% is satisfactory in maintaining the structure performance. For iron-based sludge,
678 an increase in the mechanical strength was observed when more significant proportions
679 were added. A similar effect was observed in studies with LWA, where Fe content in the
680 sludge had a high impact on water absorption and compressive strength.

681 ● Regarding the production of cement and concrete with the partial addition of WTS, it has
682 been shown to be potentially feasible, especially as a pozzolanic additions and replacement
683 of sand in concrete up to 5%.

684 ● The effect of sintering temperature and the ratio of oxides had a great influence on heavy
685 metals leaching from WTS-amended products. Sintering above 1000 °C is recommended

686 to ensure the full immobilisation of heavy metals such as Cd, Cr, Cu and Pb hence
687 preventing further environmental pollution.

688 ● Used as a substitution, WTS can make saving for raw materials and energy, replace landfill
689 as solid waste management, and contribute to sustainable construction materials production.

690

691 **8. Recommendations for future research**

692 Extensive research has been conducted with satisfactory results on the application of WTS in
693 construction materials. However, due to the complexity of this material, much remains to be
694 done to improve its **incorporation** in the construction industry practice. As far as the sintered
695 material (Bricks and LWA) is concerned, the best results were achieved with WTS blended
696 with materials rich in silica (RHA, glass). Therefore, with more conclusive studies and
697 commercial scale **applications of such combinations**, the practice could provide a solution to
698 the **sound** management and valorisation of WTS **without adverse effects on the mechanical**
699 **performance of modified construction materials**.

700 Most of the studies have examined the engineering performance of WTS-incorporated
701 cement and concrete in terms of mechanical strength, setting time, chemical-mineralogical
702 characterization and leaching tests, in a relatively short time window. There is a lack of long-
703 term durability testing in concrete, which should be addressed in future studies. In addition, due
704 to the presence of certain substances in sludge such as chloride and sulfate ions which are
705 potentially deleterious to concrete, there is no guarantee that the use of WTS will not develop
706 expansive products (Ettringite) or corrosion in the bars of reinforced concrete. **Further studies**
707 **on this topic should be carried out to ensure the safety and sustainable use of sludge residue**.

708 Considering the fact that research on WTS-derived geopolymers is a relatively new
709 approach, future studies should explore ways to improve their mechanical strength and to
710 **resolve the problems associated with setting time and workability**. More studies are needed to
711 explore appropriate experimental conditions such as combination with superplasticizers,

712 polymers, fibres and different alkaline activators (KOH and K_2SiO_3). Furthermore, potential
713 leaching of toxic chemicals and material durability from alkali-silica reaction, efflorescence,
714 acid/sulfate attack, carbonation and steel reinforcement corrosion need to be addressed
715 urgently, to promote the sustainable recycling of WTS and solid waste in general in construction
716 materials.

717

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722

723 **References**

724 [ABNT, 2004. Brazilian Standard NBR 10004 : 2004: Solid Wastes - Classification. Brazilian](#)
725 [Association of Technical Standards.](#)

726 [ABNT, 2014. Brazilian Standard NBR 6118 : 2014: Design of Concrete Structures - Procedure.](#)
727 [Brazilian Association of Technical Standards.](#)

728 [ABNT, 2015. Brazilian Standard NBR ISO 5017: 2015: Dense Shaped Refractory Products -](#)
729 [Determination of Bulk Density, Apparent Porosity and True Porosity. Brazilian](#)
730 [Association of Technical Standards.](#)

731 [ACI, 2014. Guide for Structural Lightweight-Aggregate Concrete \(ACI 213R-14\). American](#)
732 [Concrete Institute, Farmington Hills, Michican, 38 pp.](#)

733 [Ahmad, T., Ahmad, K., Alam, M., 2016. Sustainable management of water treatment sludge](#)
734 [through 3'R' concept. J. Clean. Prod. 124, 1-13.](#)

735 [Ahmad, T., Ahmad, K., Alam, M., 2017. Sludge quantification at water treatment plant and its](#)
736 [management scenario. Environ. Monitor. Assess. 189\(9\), 1-10.](#)

737 Ahmad, T., Ahmad, K., Alam, M., 2018. Investigating calcined filter backwash solids as
738 supplementary cementitious material for recycling in construction practices. *Constr. Build.*
739 *Mater.* 175, 664-671.

740 Alqam, M., Jamrah, A., Daghlas, H., 2011. Utilization of cement incorporated with water
741 treatment sludge. *Jordan J. Civil Eng.* 5, 268-277.

742 Al-Tersawy, S.H., El Sergany, F.A., 2016. Reuse of water treatment plant sludge and rice husk
743 ash in concrete production. *Int. J. Eng. Sci. Res. Technol.* 5(12), 138-152.

744 Anderson, M., Biggs, A., Winters, C., 2003. Use of two blended water industry by-product
745 wastes as a composite substitute for traditional raw materials used in clay brick manufacture.
746 In *Recycling and Reuse of Waste Materials*, Thomas Telford Publishing, pp. 417-426.

747 ASTM, 2015a. Standard Test Method for Corrosion Potentials of Uncoated Reinforcing Steel
748 in Concrete. ASTM C876-15, West Conshohocken, PA, USA.

749 ASTM, 2015b. Standard Test Methods for Wetting and Drying Compacted Soil-Cement
750 Mixtures. ASTM D559/D559M, West Conshohocken, PA, USA.

751 ASTM, 2017a. Standard Specification for Coal Fly Ash and Raw or Calcined Natural Pozzolan
752 for Use in Concrete. ASTM C618-17a, West Conshohocken, PA, USA.

753 ASTM, 2017b. Standard Specification for Lightweight Aggregates for Structural Concrete.
754 ASTM C330 / C330M-17a, West Conshohocken, PA, USA.

755 ASTM, 2018. Standard Test Methods for Determination of Water Absorption and Associated
756 Properties by Vacuum Method for Pressed Ceramic Tiles and Glass Tiles and Boil Method
757 for Extruded Ceramic Tiles and Non-tile Fired Ceramic Whiteware Products. ASTM C373-
758 18, West Conshohocken, PA, USA.

759 Babatunde, A., Zhao, Y., 2007. Constructive approaches toward water treatment works sludge
760 management: an international review of beneficial reuses. *Critical Rev. Environ. Sci.*
761 *Technol.* 37(2), 129-164.

762 Baker, R.J., White, D.J., Leeuwen, J.V., 2005. Applications for Reuse of Lime Sludge from
763 Water Softening (IHRB Project TR-535). Department of Civil, Construction and
764 Environmental Engineering, Iowa State University.

765 Basheer, L., Kropp, J., Cleland, D.J., 2001. Assessment of the durability of concrete from its
766 permeation properties: a review. *Constr. Build. Mater.* 15(2), 93-103.

767 Benlalla, A., Elmoussaouiti, M., Dahhou, M., Assafi, M. 2015. Utilization of water treatment
768 plant sludge in structural ceramics bricks. *Appl. Clay Sci.* 118, 171-177.

769 Bignozzi, M.C., 2011. Sustainable cements for green buildings construction. *Procedia Eng.* 21,
770 915-921.

771 Bijen J., 2003. *Durability of Engineering Structures: Design, Repair and Maintenance.* CRC
772 Press, Cambriedge, pp. 261.

773 BIS (Bureau of Indian Standards), 1992. *Common Burnt Clay Building Bricks Specifications:*
774 *BIS 1077: 1992*, New Delhi.

775 BIS (Bureau of Indian Standards), 2015. *Portland Pozzolana Cement – Specification – Part 1:*
776 *Fly Ash Based.* *BIS IS 1489-1 : 2015*, New Delhi.

777 CCES, 2005. *Guide to Durability Design and Construction of Concrete Structures*, CCES 01-
778 2004, China Civil Engineering Society, China.

779 Chen, H., Ma, X., Dai, H., 2010. Reuse of water purification sludge as raw material in cement
780 production. *Cement Concrete Comp.* 32(6), 436-439.

781 Chen, Y., Zhang, Y., Chen, T., Zhao, Y., Bao, S. 2011. Preparation of eco-friendly construction
782 bricks from hematite tailings. *Constr. Build. Mater.* 25(4), 2107-2111.

783 Chiang, K.Y., Chen, Y.C., Chien, K.L., 2010. Scrap glass effect on building materials
784 characteristics manufactured from water treatment plant sludge. *Environ. Eng. Sci.* 27(2),
785 137-145.

786 Chiang, K.Y., Chou, P.H., Hua, C.R., Chien, K.L., Cheeseman, C., 2009. Lightweight bricks
787 manufactured from water treatment sludge and rice husks. *J. Hazard. Mater.* 171(1-3), 76-
788 82.

789 Corrochano, B.G., Azcárate, J.A., Gonzalez, M.R., 2011. Heavy metal chemical fractionation
790 and immobilization in lightweight aggregates produced from mining and industrial waste.
791 *Inter. J. Environ. Sci. Technol.* 8(4), 667-676.

792 Cremades L.V., Cusidó, J.A., Arteaga, F., 2018. Recycling of sludge from drinking water
793 treatment as ceramic material for the manufacture of tiles. *J. Clean. Prod.* 201, 1071-1080.

794 Da Silva, E., Morita, D., Lima, A., Teixeira, L.G., 2015. Manufacturing ceramic bricks with
795 polyaluminum chloride (PAC) sludge from a water treatment plant. *Water Sci. Technol.*
796 71(11), 1638-1645.

797 Dahhou, M., El Moussaouiti, M., Arshad, M.A., Moustahsine, S., Assafi, M., 2018. Synthesis
798 and characterization of drinking water treatment plant sludge-incorporated Portland cement.
799 *J. Mater. Cycles Waste Manage.* 20(2), 891-901.

800 Dahhou, M., El Moussaouiti, M., Benlalla, A., El Hamidi, A., Taibi, M., Arshad, M.A., 2016.
801 Structural aspects and thermal degradation kinetics of water treatment plant sludge of
802 moroccan capital. *Waste Biomass Valor.* 7(5), 1177-1187.

803 Dassanayake, K., Jayasinghe, G., Surapaneni, A., Hetherington, C., 2015. A review on alum
804 sludge reuse with special reference to agricultural applications and future challenges. *Waste*
805 *Manage.* 38, 321-335.

806 Davidovits, J., 1991. Geopolymers: inorganic polymeric new materials. *J. Thermal Anal. Calor.*
807 37(8), 1633-1656.

808 Dondi, M., Cappelletti, P., D'Amore, M., de Gennaro, R., Graziano, S.F., Langella, A.,
809 Raimondo, M., Zanelli, C., 2016. Lightweight aggregates from waste materials:
810 Reappraisal of expansion behavior and prediction schemes for bloating. *Constr. Build.*
811 *Mater.* 127, 394-409.

812 Duxson, P., Provis, J.L., Lukey, G.C., Mallicoat, S.W., Kriven, W.M., van Deventer, J.S.J.,
813 2005. Understanding the relationship between geopolymer composition, microstructure
814 and mechanical properties. *Colloid. Surface. A: Physicochem. Eng. Aspects* 269(1), 47-58.

815 El-Didamony, H., Khalil, K.A., Heikal, M., 2014. Physico-chemical and surface characteristics
816 of some granulated slag-fired drinking water sludge composite cement pastes. *HBRC J.*
817 10(1), 73-81.

818 El-Eswed, B.I., Aldagag, O.M., Khalili, F.I., 2017, Efficiency and mechanism of
819 stabilization/solidification of Pb(II), Cd(II), Cu(II), Th(IV) and U(VI) in metakaolin based
820 geopolymers. *Appl. Clay Sci.* 140, 148-156.

821 Evuti, A.M., Lawal, M., 2011. Recovery of coagulants from water works sludge: a review. *Adv.*
822 *Appl. Sci. Res.* 2(6), 410-417.

823 Ewais, E., Elsaadany, R., Ahmed, A., Shalaby, N., Al-Anadouli, B., 2017. Insulating refractory
824 bricks from water treatment sludge and rice husk ash. *Refract. Ind. Ceram.* 58(2), 136-144.

825 Fernandes, F.M., Lourenço, P.B., Castro, F., 2010. Ancient clay bricks: manufacture and
826 properties. In *Materials, Technologies and Practice in Historic Heritage Structures*,
827 Springer, pp. 29-48.

828 Fernandez, L.P., Mikowski, P.C.B., Macioski, G., Nagalli, A., Freire, F.B., 2018. Study of
829 water treatment sludge incorporation into interlocking concrete pavers. *Revista Materia*
830 23(3). <http://dx.doi.org/10.1590/s1517-707620180003.0490>.

831 Finlay, N., 2015. Using Water Treatment Residual to Immobilise Lead for in-situ Remediation
832 of Contaminated Soil, Doctoral dissertation, Durham University.

833 Fujiwara, M., 2011. Outline of sludge treatment & disposal at water purification plant in Japan.
834 In: 1st Japan e Singapore Workshop and Symposium, September 2011, Singapore.

835 Gastaldini, A., Hengen, M., Gastaldini, M., Do Amaral, F., Antolini, M., Coletto, T., 2015. The
836 use of water treatment plant sludge ash as a mineral addition. *Constr. Build. Mater.* 94, 513-
837 520.

838 Geraldo, R.H., Fernandes, L.F.R., Camarini, G., 2017. Water treatment sludge and rice husk
839 ash to sustainable geopolymer production. *J. Clean. Prod.* 149, 146-155.

840 Gomes, R.K., Edna, P., Santos, D.B.G.D., Mauricio, C., 2017. Potential uses of waste sludge
841 in concrete production. *Manage. Environ. Qual.* 28(6), 821-838.

842 González-Corrochano, B., Alonso-Azcárate, J., Rodríguez, L., Pérez Lorenzo, A., Fernández
843 Torío, M., Tejado Ramos, J.J., Corvinos, M.D., Muro, C., 2017. Effect heating dwell time
844 has on the retention of heavy metals in the structure of lightweight aggregates manufactured
845 from wastes. *Environ. Technol.* 39, 2511-2523.

846 Hagemann, S.E., Gastaldini, A.L.G., Cocco, M., Jahn, S.L., Terra, L.M., 2019. Synergic effects
847 of the substitution of Portland cement for water treatment plant sludge ash and ground
848 limestone: technical and economic evaluation. *J. Clean. Prod.* 214, 916-926.

849 Hassan, K.M., Fukushi, K., Turikuzzaman, K., Moniruzzaman, S.M., 2014. Effects of using
850 arsenic–iron sludge wastes in brick making. *Waste Manage.* 34, 1072-1078.

851 Hegazy, B., Fouad, H.A., Hassanain, A.M., 2012. Brick manufacturing from water treatment
852 sludge and rice husk ash. *Aust. J. Basic Appl. Sci.* 6(3), 453-461.

853 Henderson, J. L., Raucher, R. S., Weicksel, S., Oxenford, J., Mangravite, F., 2009. Supply of
854 critical drinking water and wastewater treatment chemicals – A whitepaper for
855 understanding recent chemical price increases and shortages (ProjectNo. 4225). Denver,
856 CO: Water Research Foundation.

857 Hidalgo, AM, Murcia, M.D., Gomez, M., Gomez, E., García-Izquierdo, C., Solano, C., 2017.
858 Possible uses for sludge from drinking water treatment plants. *J. Environ. Eng.* 143,
859 04016088.

860 Hoppen, C., Portella, K., Joukoski, A., Baron, O., Franck, R., Sales, A., Andreoli, C., Paulon,
861 V., 2005. Co-disposição de lodo centrifugado de Estação de Tratamento de Água (ETA)
862 em matriz de concreto: método alternativo de preservação ambiental. *Cerâmica* 51, 85-94.

863 Horpibulsuk, S., Suksiripattanapong, C., Samingthong, W., Rachan, R., Arulrajah, A., 2016.
864 Durability against wetting-drying cycles of water treatment sludge-fly ash geopolymer and
865 water treatment sludge-cement and silty clay-cement systems. *J. Mater. Civil Eng.* 28(1),
866 04015078.

867 Huang, C., Pan, J.R., Liu, Y., 2005. Mixing water treatment residual with excavation waste soil
868 in brick and artificial aggregate making. *J. Environ. Eng.* 131(2), 272-277.

869 Huang, C., Pan, J.R., Sun, K.D., Liaw, C.T., 2001. Reuse of water treatment plant sludge and
870 dam sediment in brick-making. *Water Sci. Technol.* 44(10), 273-277.

871 Huang, C.H., Wang, S.Y., 2013. Application of water treatment sludge in the manufacturing of
872 lightweight aggregate. *Constr. Build. Mater.* 43, 174-183.

873 Huang, S.C., Chang, F.C., Lo, S.L., Lee, M.-Y., Wang, C.F., Lin, J.D., 2007. Production of
874 lightweight aggregates from mining residues, heavy metal sludge, and incinerator fly ash.
875 *J. Hazard. Mater.* 144(1), 52-58.

876 Ippolito, J., Barbarick, K., Redente, E., 1999. Co-application effects of water treatment
877 residuals and biosolids on two range grasses. *J. Environ. Qual.* 28(5), 1644-1650.

878 Jia, H., Sun, Z., Li, G., 2014. A four-stage constructed wetland system for treating polluted
879 water from an urban river. *Ecol. Eng.* 71, 48-55.

880 Jonker, A., Potgieter, J.H., 2005. An evaluation of selected waste resources for utilization in
881 ceramic materials applications. *J. Europ. Ceramic Soc.* 25(13), 3145-3149.

882 Kaosol, T., 2010. Reuse water treatment sludge for hollow concrete block manufacture. *Energy*
883 *Res. J.* 1(2), 131-134.

884 Kashyap, S., Datta, D., 2017. Reusing industrial lime sludge waste as a filler in polymeric
885 composites. *Materials Today Proceed.* 4(2), 2946-2955.

886 Keeley, J., Jarvis, P., Judd, S.J., 2014. Coagulant recovery from water treatment residuals: a
887 review of applicable technologies. *Crit. Rev. Environ. Sci. Technol.* 44, 2675-2719.

888 Kizinievič, O., Žurauskienė, R., Kizinievič, V., Žurauskas, R., 2013. Utilisation of sludge waste
889 from water treatment for ceramic products. *Constr. Build. Mater.* 41, 464-473.

890 Komljenović, M., Baščarević, Z., Bradić, V., 2010. Mechanical and microstructural properties
891 of alkali-activated fly ash geopolymers. *J. Hazard. Mater.* 181(1), 35-42.

892 Komlos, J., Welker, A., Punzi, V., Traver, R., 2013. Feasibility study of as-received and
893 modified (dried/baked) water treatment plant residuals for use in storm-water control
894 measures. *J. Environ. Eng.* 139(10), 1237-1245.

895 Król, A., 2011. Problems of assessment of heavy metals leaching from construction materials
896 to the environment. *Architect. Civil Eng. Environ.* 4(3), 71-76.

897 Lee, Y.-C., Lo, S.-L., Kuo, J., Tsai, C.-C., 2012. Beneficial uses of sludge from water
898 purification plants in concrete mix. *Environ. Eng. Sci.* 29(4), 284-289.

899 Lee, Y.E., Kim, I.T., Yoo, Y.S., 2018. Stabilization of high-organic-content water treatment
900 sludge by pyrolysis. *Energies* 11(12), 3292.

901 Li, Z., Jiang, N., Wu, F., Zhou, Z., 2013. Experimental investigation of phosphorus adsorption
902 capacity of the waterworks sludges from five cities in China. *Ecol. Eng.* 53, 165-172.

903 Lima, D.D., Zulanis, C., 2016. Use of contaminated sludge in concrete. *Procedia Eng.* 145,
904 1201-1208.

905 Lin, K.-L., Lin, C.-Y., 2005. Hydration characteristics of waste sludge ash utilized as raw
906 cement material. *Cement Concr. Res.* 35(10), 1999-2007.

907 Maiden, P., Hearn, M., Boysen, R., Chier, P., Warnecke, M., Jackson, W., 2015. Alum sludge
908 re-use, Investigation (10OS-42) prepared by GHD and Centre for Green Chemistry
909 (Monash University) for the Smart Water Fund, Victoria, ACTEW Water & Seawater',
910 Melbourne, Australia.

911 Ministry of Health, 2017. Guidelines for Drinking-water Quality Management for New Zealand
912 (3rd edn). Wellington, New Zealand.

913 Mitchell, J.K., Soga, K., 2005. Fundamentals of soil behavior, Vol. 3, John Wiley & Sons New
914 York.

915 Monteiro, S.N., Vieira, C.M.F., 2014. On the production of fired clay bricks from waste
916 materials: A critical update. *Constr. Build. Mater.* 68, 599-610.

917 NEN (Netherlands Standardization Institute), 2004. Leaching Characteristics – Determination
918 of the Leaching of Inorganic Components from Moulded or Monolithic Materials with a
919 Diffusion Test – Solid Earthy and Stony Materials, NEN 7375:2004, Delft, The
920 Netherlands.

921 Nimwinya, E., Arjharn, W., Horpibulsuk, S., Phoo-ngernkham, T., Poowancum, A., 2016. A
922 sustainable calcined water treatment sludge and rice husk ash geopolymer. *J. Clean. Prod.*
923 119, 128-134.

924 Nowasell, Q.C., Kevern, J.T., 2015. Using drinking water treatment waste as a low-cost internal
925 curing agent for concrete. *ACI Mater. J.* 112(1), 5.

926 O’Kelly, B.C., 2008. Geotechnical properties of a municipal water treatment sludge
927 incorporating a coagulant. *Canadian Geotech. J.* 45(5), 715-725.

928 O’Kelly, B.C., Quille, M.E., 2009. Compressibility and consolidation of water treatment
929 residues, *Proceedings of the Institution of Civil Engineers-Waste and Resource*
930 *Management* 162, 85-97.

931 Ooi, T.Y., Yong, E.L., Din, M.F.M., Rezania, S., Arninudin, E., Chelliapan, S., Rahman, A.A.,
932 Park, J., 2018. Optimization of aluminium recovery from water treatment sludge using
933 response surface methodology. *J. Environ. Manage.* 228, 13-19.

934 Pan, J.R., Huang, C., Lin, S., 2004. Reuse of fresh water sludge in cement making. *Water Sci.*
935 *Technol.* 50(9), 183-188.

936 Phair, J.W., Van Deventer, J.S.J., 2002. Effect of the silicate activator pH on the microstructural
937 characteristics of waste-based geopolymers. *Int. J. Mineral Process.* 66(1), 121-143.

938 Pizzi, N.G., 2010. *Water Treatment*, 4th edition, American Water Works Association, USA.

939 Ramer, D., Wang, M., 2000. Performance of roadway embankment on lime waste. In
940 Geotechnics of High Water Content Materials, ASTM International.

941 Raut, S., Ralegaonkar, R., Mandavgane, S., 2011. Development of sustainable construction
942 material using industrial and agricultural solid waste: A review of waste-create bricks.
943 *Constr. Build. Mater.* 25(10), 4037-4042.

944 Renault, F., Sancey, B., Badot, P.M., Crini, G., 2009. Chitosan for coagulation/flocculation
945 processes – An eco-friendly approach. *Europ. Polym. J.* 45(5), 1337-1348.

946 Rodríguez, N.H., Ramírez, S.M., Varela, M.B., Guillem, M., Puig, J., Larrotcha, E., Flores, J.,
947 2010. Re-use of drinking water treatment plant (DWTP) sludge: characterization and
948 technological behaviour of cement mortars with atomized sludge additions. *Cement Concr.*
949 *Res.* 40(5), 778-86.

950 Sales, A., de Souza, F.R., Almeida, F.d.C.R., 2011. Mechanical properties of concrete produced
951 with a composite of water treatment sludge and sawdust. *Constr. Build Mater.* 25(6), 2793-
952 2798.

953 Sales, A., de Souza, F.R., dos Santos, W.N., Zimer, A.M., do Couto Rosa Almeida, F., 2010.
954 Lightweight composite concrete produced with water treatment sludge and sawdust:
955 Thermal properties and potential application. *Constr. Build. Mater.* 24(12), 2446-2453.

956 Salwa, M.S., Al Bakri, A.M., Kamarudin, H., Ruzaidi, C., Binhussain, M., Zaliha, S.S., 2013.
957 Review on current geopolymers as a coating material. *Austr. J. Basic Appl. Sci.* 7(5), 246-
958 257.

959 Shih, K., Tang, Y., 2011. Prolonged toxicity characteristic leaching procedure for nickel and
960 copper aluminates. *J. Environ. Monitor.* 13(4), 829-835.

961 Singh, B., Ishwarya, G., Gupta, M., Bhattacharyya, S.K., 2015. Geopolymer concrete: A review
962 of some recent developments. *Constr. Build. Mater.* 85, 78-90.

963 Snellings, R., Mertens, G., Elsen, J., 2012. Supplementary cementitious materials. *Rev. Mineral.*
964 *Geochem.* 74(1), 211-278.

965 Soltan, A.M.M., Kahl, W.-A., Abd El-Raouf, F., Abdel-Hamid El-Kaliouby, B., Abdel-Kader
966 Serry, M., Abdel-Kader, N.A., 2016. Lightweight aggregates from mixtures of granite
967 wastes with clay. *J. Clean. Prod.* 117, 139-149.

968 Sudagar, A., Andrejkovičová, S., Patinha, C., Velosa, A., McAdam, A., da Silva, E.F., Rocha,
969 F., 2018. A novel study on the influence of cork waste residue on metakaolin-zeolite based
970 geopolymers. *Appl. Clay Sci.* 152, 196-210.

971 Suksiripattanapong, C., Horpibulsuk, S., Boongrasan, S., Udomchai, A., Chinkulkijniwat, A.,
972 Arulrajah, A., 2015a. Unit weight, strength and microstructure of a water treatment sludge–
973 fly ash lightweight cellular geopolymer. *Constr. Build. Mater.* 94, 807-816.

974 Suksiripattanapong, C., Horpibulsuk, S., Chanprasert, P., Sukmak, P., Arulrajah, A., 2015b.
975 Compressive strength development in fly ash geopolymer masonry units manufactured
976 from water treatment sludge. *Constr. Build. Mater.* 82, 20-30.

977 Tafarel, N.F., Macioski, G., Carvalho, K.Q.d., Nagalli, A., Freitas, D.C.D., Passig, F.H., 2016.
978 Avaliação das propriedades do concreto devido à incorporação de lodo de estação de
979 tratamento de água. *Matéria (Rio de Janeiro)* 21, 974-986.

980 Tang, Z., Li, W., Hu, Y., Zhou, J.L., Tam, V.W.Y., 2019a. Review on designs and properties
981 of multifunctional alkali-activated materials (AAMs). *Constr. Build. Mater.* 200, 474-489.

982 Tang, Z., Li, W., Ke, G., Zhou, J.L., Tam, V.W.Y., 2019b. Sulfate attack resistance of
983 sustainable concrete incorporating various industrial solid wastes. *J. Clean. Prod.* 218, 81-
984 822.

985 Tay, J.H., Show, K.Y. 1991. Properties of cement made from sludge. *J. Environ. Eng.* 117(2),
986 236-246.

987 Teixeira, S., Santos, G., Souza, A., Alessio, P., Souza, S., Souza, N., 2011. The effect of
988 incorporation of a Brazilian water treatment plant sludge on the properties of ceramic
989 materials. *Appl. Clay Sci.* 53(4), 561-565.

- 990 Townsend, T., Jang, Y.-C., Tolaymat, T., Jambeck, J., 2003. Leaching tests for evaluating risk
991 in solid waste management decision making, Florida Center for Solid and Hazardous Waste
992 Management, Gainesville, FL. Report #04-0332007.
- 993 Toya, T., Nakamura, A., Kameshima, Y., Nakajima, A., Okada, K., 2007. Glass-ceramics
994 prepared from sludge generated by a water purification plant. *Ceramics Int.* 33(4), 573-577.
- 995 Waijarean, N., Asavapisit, S., Sombatsompop, K., 2014. Strength and microstructure of water
996 treatment residue-based geopolymers containing heavy metals. *Constr. Build. Mater.* 50,
997 486-491.
- 998 Waijarean, N., MacKenzie, K.J.D., Asavapisit, S., Piyaphanuwat, R., Jameson, G.N.L., 2017.
999 Synthesis and properties of geopolymers based on water treatment residue and their
1000 immobilization of some heavy metals. *J. Mater. Sci.* 52(12), 7345-7359.
- 1001 Wang, M., Hull, J., Jao, M., Dempsey, B., Cornwell, D., 1992. Engineering behavior of water
1002 treatment sludge. *J. Environ. Eng.* 118(6), 848-864.
- 1003 Wang, X., Jin, Y., Wang, Z., Mahar, R.B., Nie, Y., 2008. A research on sintering characteristics
1004 and mechanisms of dried sewage sludge. *J. Hazard. Mater.* 160(2), 489-494.
- 1005 Watanabe, Y., Komine, H., Yasuhara, K., Murakami, S., 2011. Batch leaching test focusing on
1006 clod size of drinking water sludge and applicability to long-term prediction using column
1007 leaching test, *Geo-Frontiers 2011: Advances in Geotechnical Engineering* 1075-1080.
- 1008 Wei, N., 2015. Leachability of heavy metals from lightweight aggregates made with sewage
1009 sludge and municipal solid waste incineration fly ash. *Int. J. Environ. Res. Public Health*
1010 12(5), 4992-5005.
- 1011 Wolff, E., Schwabe, W.K., Conceição, S.V., 2015. Utilization of water treatment plant sludge
1012 in structural ceramics. *J. Clean. Product.* 96, 282-289.
- 1013 Xu, G., Zou, J., Li, G., 2008a. Ceramsite made with water and wastewater sludge and its
1014 characteristics affected by SiO₂ and Al₂O₃. *Environ. Sci. Technol.* 42, 7417-7423.

1015 Xu, G.R., Zou, J.L., Li, G.B., 2008b. Effect of sintering temperature on the characteristics of
1016 sludge ceramsite. *J. Hazard. Mater.* 150, 394-400.

1017 Xu, G.R., Zou, J.L., Li, G.B., 2008c. Solidification and leaching behaviours of Cr⁶⁺ in sludge
1018 ceramsite. *J. Hazard. Mater.* 153, 1031-1035.

1019 Xu, G.R., Zou, J.L., Li, G.B., 2010. Stabilization of heavy metals in sludge ceramsite. *Water*
1020 *Res.* 44, 2930-2938.

1021 Zamora, R.M., Ayala, F.E., Garcia, L.C., Moreno, A.D., Schouwenaars, R., 2008. Optimization
1022 of the preparation conditions of ceramic products using drinking water treatment sludges.
1023 *J. Environ. Sci. Health A Toxic/Hazard. Substances Environ. Eng.* 43, 1562-1568.

1024 Zou, J.L., Xu, G.R., Li, G.B., 2009. Ceramsite obtained from water and wastewater sludge and
1025 its characteristics affected by Fe₂O₃, CaO, and MgO. *J. Hazard. Mater.* 165, 995-1001.

1026

1027 **After proof review:**

1028 Central Intelligence Agency, 2017. The World Factbook.
1029 [https://www.cia.gov/library/publications/resources/the-world-](https://www.cia.gov/library/publications/resources/the-world-factbook/fields/208rank.html)
1030 [factbook/fields/208rank.html](https://www.cia.gov/library/publications/resources/the-world-factbook/fields/208rank.html). Accessed 20022019.

1031 USEPA, 1992. Method 1311: Toxicity Characteristic Leaching Procedure.
1032 [https://www.epa.gov/hw-sw846/sw-846-test-method-1311-toxicity-characteristic-](https://www.epa.gov/hw-sw846/sw-846-test-method-1311-toxicity-characteristic-leaching-procedure)
1033 [leaching-procedure](https://www.epa.gov/hw-sw846/sw-846-test-method-1311-toxicity-characteristic-leaching-procedure). Accessed 20022019.