

1        **Influence of thermal hydrolysis pretreatment on physicochemical properties and**  
2        **anaerobic biodegradability of waste activated sludge with different solids content**

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27 **Abstract**

28 The influence of thermal hydrolysis pretreatment (THP) on physicochemical properties (pH,  
29 total solids, volatile solids, chemical oxygen demand, total nitrogen, ammonium nitrogen,  
30 volatile fatty acids, viscosity, and cell morphology) and anaerobic biodegradability of highly  
31 concentrated waste activated sludge (WAS) with TS content ranging from 1-7% was  
32 evaluated at different temperatures ranging from 100-220 °C. The biomethane potential (BMP)  
33 of the WAS was systematically analyzed and evaluated. Images of its cellular structure were  
34 also analyzed. The results indicated that THP is a useful method for solubilizing volatile  
35 solids and enhancing CH<sub>4</sub> production regardless of the TS content of the WAS feed. The  
36 ultimate CH<sub>4</sub> production determined from the BMP analysis was 313-348 L CH<sub>4</sub>/kg VS (72.6-  
37 74.1% CH<sub>4</sub>) at the optimum THP temperature of 180 °C. The results showed that THP could  
38 improve both the capacity and efficiency of anaerobic digestion, even at a high TS content,  
39 and could achieve the dual purpose of sludge reduction and higher energy recovery.

40 **Keywords:** thermal pretreatment; sludge properties; wastewater activated sludge; anaerobic  
41 digestion; biomethane potential

## 42 **1. Introduction**

43 A large amount of excess sludge is produced daily at wastewater treatment plants (WWTPs)  
44 worldwide (Zhang et al., 2017). In South Korea , the average generation of sewage sludge (as  
45 wetted solids) in 2013 was about 3.6 million t/y, thereby accounting for 19.65% of the  
46 municipal solid waste stream by volume (Korea Ministry of Environment, 2016). Since  
47 sewage sludge contains an array of pathogens, nutrients, degradable organics, and possibly  
48 some hazardous substances and heavy metals, it needs to be handled carefully and its mass  
49 must be reduced appropriately before disposal or reuse (Raheem et al., 2017; Semblante et al.,  
50 2014).

51 Standards and policies for sewage sludge management vary by country (Villar et al., 2016). In  
52 general, they aim to promote cost-effective and sustainable sludge management through best  
53 practice as well as research and development. The management of sewage sludge presents  
54 many technical challenges, and the associated capital and operating costs can exceed 50% of  
55 the total wastewater treatment costs (Serrano et al., 2015). Additionally, because sewage  
56 sludge contains mostly degradable organics, it is a valuable source of sustainable biomass  
57 energy (Leng et al., 2018). Consequently, it is expected to become a mainstream energy  
58 source for environmentally and socially sustainable development (Nguyen et al., 2017a; Tyagi  
59 and Lo, 2013).

60 There are several technologies for sewage sludge treatment, including alkaline stabilization,  
61 aerobic digestion, anaerobic digestion (AD), composting, landfilling, and incineration  
62 (Bougrier et al., 2007; Chiu et al., 2015; Zhang et al., 2017). Among them, AD has likely been  
63 the most widely used process for treating sewage sludge at WWTPs because of its many  
64 advantages, such as high renewable energy production, low environmental impacts, and low  
65 solid residue with fewer pathogens (Chiu et al., 2015; Nizami et al., 2017). However, further  
66 studies and development are necessary to optimize the performance of AD in a retrofit design  
67 or new project. These limitations include i) slow and incomplete decomposition of organic

68 matter; ii) long digestion time and large digester volume causing high capital and operating  
69 costs; (iii) sensitivity to temperature, pH, loading changes, and inhibitors; and iv) low biogas  
70 production and CH<sub>4</sub> content (Appels et al., 2008; Bougrier et al., 2007; Koch and Drewes,  
71 2014; Ushani et al., 2018). To overcome the aforementioned issues and to improve the  
72 efficiency of AD, various pretreatments can be used, including biological (e.g., enzyme and  
73 ensilage) (Liu et al., 2017), mechanical (e.g., ultrasonic, agitator, high pressure  
74 homogenization, and jetting) (Le et al., 2016; Nguyen et al., 2017c), chemical (e.g., alkali or  
75 acid) (Li et al., 2012; Rajesh Banu et al., 2017), thermal (Appels et al., 2010; Eftaxias et al.,  
76 2018; Li et al., 2018), and combined (Peng et al., 2018) processes.

77 Thermal hydrolysis pretreatment (THP) has been proposed as an effective technology for  
78 accelerating the hydrolysis of organic matter (Jin et al., 2016; Li et al., 2016; Urrea et al.,  
79 2018). During THP, complex organic substances can be broken down into smaller and soluble  
80 molecules suitable for biodegradation (Yang et al., 2017) and cell walls rupture, releasing  
81 their contents into the homogenized solution of which the largest are lysed microbial cells that  
82 subsequently become available for metabolic activity and microbial growth. In addition, THP  
83 provides a homogenized sludge solution with stable conditions, thereby preventing shock  
84 loading during AD, which consequently leads to shortened digestion time, decreased sludge  
85 mass, enhanced biodegradation, and CH<sub>4</sub> production (Ariunbaatar et al., 2014; Bougrier et al.,  
86 2007; Kim et al., 2015; Liao et al., 2016). The general advantages are outlined above, but the  
87 specific physicochemical and biological properties (organic/insoluble fraction, solubilization,  
88 viscosity, biodegradability, cell morphology, etc.) of thermally pretreated waste activated  
89 sludge (WAS) in different experimental conditions (solid load and temperature) and their  
90 relationships have not been systematically elucidated (Liu et al., 2012). In addition, in Korea,  
91 there are still very few investigations on the use of the thermal pretreatment method to  
92 enhance sludge hydrolysis, anaerobic biodegradability, and biogas production to meet local  
93 conditions (environment, organic components, waste sources, etc.) (Yang et al., 2017).

94 The major goal of this study was to investigate and evaluate the performance of THP for  
95 highly concentrated WAS. The results will be helpful for integrating THP processes into  
96 existing and new WWTPs in South Korea, with the goal of improving sludge digester  
97 performance and biogas yields. The physicochemical properties of sludge, namely volatile  
98 solids (VS), chemical oxygen demand (COD) solubilization, ammonium nitrogen ( $\text{NH}_4^+\text{-N}$ ),  
99 volatile fatty acids (VFA), viscosity, and cellular structure, were measured. Biomethane  
100 potential (BMP) was also tested. Furthermore, the optimum temperature of THP for various  
101 solids concentrations was evaluated in order to obtain the most effective economic and  
102 environmental benefits.

## 103 **2. Materials and Methods**

### 104 **2.1. Preparation of substrate and inoculation**

105 The raw WAS and inoculum sludge used in this study were sourced from a WWTP with a  
106 capacity of 85 000 m<sup>3</sup>/d and a mesophilic AD plant, respectively. Both plants were located in  
107 “N” city, South Korea. Large debris and inert materials were eliminated from the samples  
108 using a No. 18 mesh sieve (0.85 mm). Samples were then stored in a plastic bottle at 4 °C and  
109 in an incubator at 35 °C, respectively, before experimental use. WAS of high total solids (TS)  
110 concentration (69.6-72.4 g TS/L) was utilized by diluting it with deionized water to desired  
111 solids concentrations of 9.9 g TS/L, 29.91 g TS/L, 49.90 g TS/L, and 70.06 g TS/L, which  
112 corresponded to WAS<sub>1%</sub>, WAS<sub>3%</sub>, WAS<sub>5%</sub>, and WAS<sub>7%</sub>, respectively. Characteristics of the  
113 WAS samples are summarized in Table 1.

114 Table 1: Characteristics of raw waste activated sludge (WAS) used for the experiment.

115 [Insert Table 1]

## 116 2.2. Thermal pretreatment and evaluation of chemical oxygen demand solubilization

117 A batch-type processor was used for the high-pressure THP. The processor consisted of a  
118 control unit, pressurized vessel, pressure-reducing valve, pressure gauge, heating element,  
119 hydraulic cooling system, and thermometer (Fig. 1); these were designed by our lab members  
120 for laboratory testing only. The stainless steel reactor had a total capacity of 1 L. The  
121 automatic temperature control apparatus was designed to maintain a constant temperature with  
122 an uncertainty of  $\pm 1$  °C by activating the heating or cooling system as needed. Thermal  
123 pretreatment of WAS was conducted at temperatures ranging from 100-220 °C with the  
124 sludge sample volume held at a constant value of 700 mL for each experiment.

125 [Insert Figure 1]

126 Fig. 1. Thermal pretreatment apparatus a) schematic diagram and b-c) photographs.

127 In order to create an anaerobic environment and avoid oxidation of the samples during  
128 hydrolysis, the headspace of the THP reactor was initially flushed using N<sub>2</sub> at a rate of  
129 approximately 0.5 L/min for 3 to 5 min before performing the thermal hydrolysis. The  
130 pretreatment temperature was increased to the set point and maintained for 30 min. During the  
131 pretreatment experiment, the pressure inside the pressurized vessel varied from 0.8 to 2.6 MPa  
132 depending on the setpoint pretreatment temperature. The agitation velocity was set at 180 rpm  
133 and maintained until sampling. After each reaction, the temperature of the sludge was allowed  
134 to cool to less than 60 °C, and saturated vapor was relieved to reduce pressure for sample  
135 collection.

136 COD solubilization served as an index indicating the efficiency of the pretreatment process by  
137 way of Eq. 1.

$$\text{COD solubilization (\%)} = \frac{C_s - C_{S0}}{C_o - C_{S0}} \times 100 \quad \text{Eq. 1}$$

where  $C_o$  = total chemical oxygen demand (TCOD) concentration before pretreatment in g/L;  
 $C_{so}$  = soluble chemical oxygen demand (SCOD) concentration before pretreatment in g/L;  
and  $C_s$  = SCOD concentration after pretreatment in g/L.

### 138 **2.3. Biochemical CH<sub>4</sub> potential tests**

139 In order to evaluate enhancements in the biodegradability of substrates and ultimate biogas  
140 yields due to THP, BMP tests were performed (Dwyer et al., 2008; Holliger et al., 2016;  
141 Nguyen et al., 2017a).

142 Tests were conducted in 630 mL serum bottles (effective volume of 330 mL) with coiled  
143 butyl rubber stoppers at 35 °C. Inoculum (10%, v/v) and substrate (2 g VS<sub>substrate</sub>/L) were  
144 placed in a growth medium containing 1.8 g/L NH<sub>4</sub>Cl, 0.7 g/L KH<sub>2</sub>PO<sub>4</sub>, 0.4 g/L MgCl<sub>2</sub>·6H<sub>2</sub>O,  
145 0.2 g/L CaCl<sub>2</sub>·2H<sub>2</sub>O, 20 mg/L FeCl<sub>2</sub>·4H<sub>2</sub>O, 5 mg/L CoCl<sub>2</sub>·6H<sub>2</sub>O, 1 mg/L MnCl<sub>2</sub>·4H<sub>2</sub>O, 1  
146 mg/L NiCl<sub>2</sub>·6H<sub>2</sub>O, 0.5 mg/L ZnCl<sub>2</sub>, 0.5 mg/L H<sub>3</sub>BO<sub>3</sub>, 0.5 mg/L Na<sub>2</sub>SeO<sub>3</sub>, 0.4 mg/L  
147 CuCl<sub>2</sub>·2H<sub>2</sub>O, and 0.1 mg/L Na<sub>2</sub>MoO<sub>4</sub>·2H<sub>2</sub>O (Del Río et al., 2011). If necessary, the pH was  
148 adjusted to 7.1 ± 0.1 using either 2 M HCl or 2 M NaOH solution. To prevent a sudden  
149 decrease in pH caused by acid fermentation during the early stages of AD, 2.6 g/L of NaHCO<sub>3</sub>  
150 was included in the serum bottle as a buffer. Pure N<sub>2</sub> was used to purge the serum bottles to  
151 create anaerobic conditions (Nguyen et al., 2017b).

152 The biomethane production was expressed as the volume of CH<sub>4</sub> produced per gram of VS of  
153 substrate (mL CH<sub>4</sub>/g VS<sub>add</sub>). CH<sub>4</sub> production was measured periodically using a 50 mL glass  
154 syringe to extract the gas. In order to take volume expansion and saturated steam pressure into  
155 account, the CH<sub>4</sub> volume was converted to standard state (STP: 0 °C, 1 bar) volume (V) using  
156 the ideal gas law and the 35 °C saturated vapor pressure of 42.2 mmHg.

## 157 **2.4. Analytical methods**

158 TS, VS, TCOD, SCOD, TN, and  $\text{NH}_4^+$ -N analyses were performed using standard methods  
159 (APHA-AWWA-WEF, 2005). Soluble fractions of  $\text{SCOD}_{\text{Cr}}$ ,  $\text{NH}_4^+$ -N, and VFA were  
160 analyzed using filtrate obtained from filtering the supernatant with a 1.2  $\mu\text{m}$  GF/C microfiber  
161 filter (Whatman, UK) after centrifugation at 7000 rpm for 20 min. pH was measured using a  
162 pH meter (Hanna HI223, USA). VFA were analyzed by gas chromatography (GC) with a  
163 flame ion detector (Agilent 7890A, Agilent Technologies, Inc., USA) after solvent extraction  
164 (1:1 with ether) and adjustment of filtrate pH to  $2.2 \pm 0.2$  using 3% HCl (Nguyen et al., 2016).  
165  $\text{CH}_4$  content in the generated biogas was measured by employing the BMP test and was  
166 examined using gas chromatography with a thermal conductivity detector (Agilent 7890A,  
167 Agilent Technologies, Inc., USA). The GC operation conditions used for VFA and  $\text{CH}_4$   
168 analyses are summarized in Table 2.

169 Table 2: Gas chromatography analysis conditions for volatile fatty acids and  $\text{CH}_4$ .

170 [Insert Table 2]

171 The viscosities of the untreated and pretreated sludge samples were determined at 20 °C using  
172 a viscometer (Brookfield Viscometer DV2T, USA). The torque percentage was adjusted  
173 through a range of 20% to 800% by adjusting the rotational speed of the spindle (Xue et al.,  
174 2015). In order to observe changes in the microorganism cell structure, images were taken  
175 using a transmission electron microscope (80 Kv, JEM1010, JEOL, Japan).

## 176 **3. Results and Discussion**

### 177 **3.1. Effects of thermal hydrolysis pretreatment on waste activated sludge characteristics**

178 THP was undertaken in the range of 100 to 220 °C using four kinds of sludge ( $\text{WAS}_{1\%}$ ,  
179  $\text{WAS}_{3\%}$ ,  $\text{WAS}_{5\%}$ , and  $\text{WAS}_{7\%}$ ) with solid concentrations of 9.90 g TS/L, 29.91 g TS/L, 49.90



180 g TS/L, and 70.06 g TS/L, respectively. The changes in sludge physicochemical properties  
181 were analyzed and the results are shown in Figs. 2 and 3.

182 The TS, TCOD, and TN concentrations of the thermally pretreated sludge decreased  
183 marginally due to the formation of a scale layer on the inner wall of the thermal reactor (Fig  
184 2). On the other hand, SCOD,  $\text{NH}_4^+$ -N, and VFA in the soluble phase gradually increased  
185 directly proportional to the increase in pretreatment temperature from 100 to 220 °C in all  
186 four samples (Fig. 2). It is of particular note that THP resulted in significant COD  
187 solubilization. The most notable increase in SCOD was observed with WAS<sub>7%</sub> (sample with  
188 the highest solid content). In addition, there was a notable increase in the SCOD fraction of  
189 TCOD of this sample from 49.63 to 55.11% at the pretreatment temperature of 220 °C. Figure  
190 2 also shows a decrease in the pretreated sludge pH as the reaction temperature increased.  
191 This coincided with the increase in VFA concentration (Fig. 2), thereby confirming that  
192 thermal pretreatment was effective at promoting hydrolysis and solubilization. Additionally,  
193 during WAS thermal hydrolysis, the VFA concentrations in all WAS samples consistently  
194 increased with increased thermal hydrolysis temperature (Fig. 2). An increase in VFA  
195 concentration could possibly be ascribed to the hydrolysis of unsaturated lipids (Liao et al.,  
196 2018; Wilson and Novak, 2009).

197 [Insert Figure 2]

198 Fig. 2. Influence of thermal hydrolysis pretreatment temperature on physicochemical  
199 properties of treated sludge: a) WAS<sub>1%</sub>, b) WAS<sub>3%</sub>, c) WAS<sub>5%</sub>, and d) WAS<sub>7%</sub>.

200 At the pretreatment temperature of 220 °C, the  $\text{NH}_4^+$ -N concentration was 0.213, 0.596, 1.120,  
201 and 1.804 g/L for WAS<sub>1%</sub>, WAS<sub>3%</sub>, WAS<sub>5%</sub>, and WAS<sub>7%</sub>, respectively, and increased by 600-  
202 1675% depending on the TS concentration. Total ammonia nitrogen (TAN) exists in the form  
203 of free ammonia nitrogen and ionized ammonium nitrogen ( $\text{NH}_4^+$ -N). TAN disrupts anaerobic  
204 digestive reactions, thereby decreasing digestive efficiency and biogas generation (Nakakubo  
205 et al., 2008; Rajagopal et al., 2013).

206 Wilson and Novak (2009) demonstrated that proteins existing as particulate matter can be  
207 converted into  $\text{NH}_4^+\text{-N}$  through thermal pretreatment. Ammonia is an inhibitory intermediate  
208 during AD (Xie et al., 2016). AD performance starts to decline when the concentration of  
209 TAN in the digester reaches 1.7 to 2.7 g/L (Yenigün and Demirel, 2013). In this study, the  
210 concentration of  $\text{NH}_4^+\text{-N}$  reached 1.804 g/L for WAS<sub>7%</sub> at the pretreatment temperature of  
211 220 °C. Thus, there could be a relationship between effective hydrolysis and excessive  $\text{NH}_4^+\text{-N}$   
212 N concentration in the reactor. Since both of these were promoted by thermal pretreatment, an  
213 optimal pretreatment temperature was expected, as discussed in a later section.

214 [Insert Figure 3]

215 Fig. 3. Variations in volatile fatty acids composition and pH at different temperatures for  
216 thermally pretreated a) WAS<sub>1%</sub>, b) WAS<sub>3%</sub>, c) WAS<sub>5%</sub>, and d) WAS<sub>7%</sub>.

217 Figure 3 shows the variations in pH and VFA composition of each sample at different THP  
218 temperatures. Acetic acid, propionic acid, butyric acid, and valeric acid accounted for roughly  
219 73.6 to 85.9% of the VFA, with the percentage of acetic acid being the highest. More  
220 specifically, the acetic acid accounted for a significantly higher proportion (62.9-100%) in the  
221 group of four acids obtained above. Acetic acid can be readily converted into  $\text{CH}_4$  compared  
222 to other VFA. Thus, given the high concentration of acetic acid shown in Fig. 3, an enhanced  
223 AD performance was expected, as discussed in Section 3.5.

### 224 3.2. Organic matter fraction and solubilization

225 The VS over TS ratio of these sludge samples varied from 0.773 to 0.793, thereby indicating  
226 that the sludge consisted mainly of organic substances. The highest COD was associated with  
227 the solid fraction rather than the soluble fraction, as evidenced by the rather low SCOD to  
228 TCOD ratio in the range of 0.022 to 0.046.

229 Solubilization of all WAS samples (1%, 3%, 5%, and 7%) increased as pretreatment  
230 temperature increased (Fig. 4). The change was marginal between 100 and 120 °C, but then  
231 suddenly increased above 140 °C. All the WAS (1%, 3%, 5%, and 7%) showed maximum  
232 solubilization (51.2%, 47.3%, 45.9%, and 44.0%, respectively) at 220 °C.

233 Del Río et al. (2011) and Seviour et al. (2009) reported that aerobic granular sludge exists as a  
234 gel structure at temperatures around 115 °C. Moreover, the sludge formed a gel structure due  
235 to the high content of extracellular polymeric substances (EPS) with gel-forming properties at  
236 or below temperatures of 115 °C. They reported that the microorganisms release small  
237 amounts of EPS at moderate temperatures, which then act as a bond to maintain the gel  
238 structure. However, EPS lose these gel-forming properties at high temperatures. These results  
239 were confirmed in the current study, which showed low solubilization at temperatures around  
240 100-120 °C.

241 Evaluation of solubilization for each TS concentration of sludge revealed that as TS  
242 concentration increased, the maximum solubilization decreased. This is perhaps  
243 understandable that for WAS with high solids concentration, the heat transfer coefficient was  
244 lower than that in WAS with low solids concentration. Therefore, when applying THP to  
245 WAS with high solids concentration, higher thermal energy input or longer residence time  
246 might be required.

247 [Insert Figure 4]

248 Fig. 4. Impact of pretreatment temperature on solubilization of waste activated sludge: (a)  
249 WAS<sub>1%</sub>, b) WAS<sub>3%</sub>, c) WAS<sub>5%</sub>, and d) WAS<sub>7%</sub>.

### 250 3.3. Sludge viscosity

251 Viscosity is an important parameter that significantly affects the solubilization and  
252 biodegradability of sludge. It reflects the degree of interaction between the particles and  
253 sludge flocs within the mixed sludge (Markis et al., 2016). The variations in sludge viscosity

254 by different initial solids concentrations at an initial temperature of 20 °C and after THP are  
255 shown in Fig. 5.

256 At 20 °C, the viscosity of raw concentrated WAS increased proportionally to the solids  
257 concentration (Fig. 5a). The sludge viscosity at a TS concentration of 9.9 g/L was 40 MPa·s,  
258 and increased to 3102 MPa·s for sludge with a TS concentration of 70.63 g/L. As the  
259 solubilizing temperature increased, the viscosity of each sludge decreased gradually (Fig. 5b).  
260 At 140 °C, viscosity suddenly decreased, and at 220 °C, it had decreased to 3 MPa·s, 12  
261 MPa·s, 28 MPa·s, and 250 MPa·s for WAS of 1%, 3%, 5%, and 7%, respectively. As sludge  
262 concentration increased, viscosity also increased, which meant that more pumping energy  
263 would need to be supplied for pipe transfer and reactor agitation could be increased. Therefore,  
264 it is likely that as sludge viscosity decreases from thermal pretreatment, the energy required  
265 for facility operation (i.e., pumping and mixing) could be reduced (Brar et al., 2005; Xue et al.,  
266 2015) and particulate fraction biodegradability could be improved (Bougrier et al., 2008; Brar  
267 et al., 2005).

268 [Insert Figure 5]

269 Fig. 5. Change in sludge viscosity at (a) different initial solids concentrations and (b) after  
270 thermal pretreatment at various temperatures.

### 271 3.4. Transmission electron microscope analysis

272 Transmission electron microscope analysis was used to study the effects of THP on cell  
273 morphology (Fig. S1). The microorganism cell walls were destroyed by thermal pretreatment,  
274 thereby resulting in the release of cell contents (EPS and intracellular organic substances). At  
275 120 °C, the outer cell wall was partially destroyed (Del Río et al., 2011). As the pretreatment  
276 temperature increased, internal cellular substances were eluted through the outer cell wall.  
277 However, at 180 °C, the outer cell wall contracted. Because the cell walls of microorganisms  
278 comprising WAS exist as a solid structure, the hydrolysis process was delayed during the AD

279 stage. Cell walls of microorganisms comprising WAS are destroyed by THP, and the  
280 subsequent elution of the internal cellular solution causes the amount of dissolved substances  
281 to increase (Seviour et al., 2009). Increases in dissolved substances facilitate conversion to  
282 CH<sub>4</sub> later in the process, and thus increase the biodegradability and biogas output of AD  
283 (Nguyen et al., 2017c).

### 284 **3.5. Effects of thermal hydrolysis pretreatment on ultimate biomethane potential**

285 BMP is an important parameter to evaluate the biogas production potential of an organic  
286 substrate by AD because it can be used to assess anaerobic biodegradability, CH<sub>4</sub> production  
287 rate, and the ultimate biomethane yields of substrates. All of these variables affect the  
288 economics of the AD process. BMP test results obtained from samples both before and after  
289 THP are presented in Table 3.

290 Increasing the THP temperature from the ambient value to 220 °C resulted in CH<sub>4</sub> production  
291 increasing for all WAS samples (1%, 3%, 5%, and 7%), reaching peak values of 348, 345,  
292 320, and 313 L CH<sub>4</sub>/kg VS<sub>add</sub> (or 244, 254, 235, and 232 L CH<sub>4</sub>/kg COD<sub>add</sub>), respectively, at  
293 180 °C THP, which corresponded to CH<sub>4</sub> production increasing by 86.1%, 84.3%, 76.2%, and  
294 76.0%, respectively, compared to those without THP. Above 180 °C for THP, CH<sub>4</sub> production  
295 decreased gradually. This was probably caused by increases in NH<sub>4</sub><sup>+</sup>-N at high temperatures  
296 acting as an inhibiting factor in AD processes. It should also be noted that non-biodegradable  
297 substances generated by carbohydrates and proteins at a high enough temperature (optimal  
298 temperature) could cause decreases in CH<sub>4</sub> production, which has been reported in the  
299 literature. For example, Bougrier et al. (2008) undertook WAS solubilization at 95-210 °C,  
300 and reported that solubilization of carbohydrates and proteins in the sludge increased as  
301 temperature increased. However, in the case of carbohydrates, solubilization above 170 °C  
302 gradually decreased, which was attributed to a decrease in biodegradable substances through  
303 the Maillard reaction (i.e., a bonding between sugar and amino acids at high temperatures).

304 Furthermore, Dwyer et al. (2008) confirmed that thermal hydrolysis produces colored,  
305 recalcitrant compounds and melanoidins that can affect anaerobic biodegradability and CH<sub>4</sub>  
306 production. Del Río et al. (2011) also confirmed that at a higher sludge pretreatment  
307 temperature, dissolved carbohydrates and proteins decrease, which causes decreasing CH<sub>4</sub>  
308 production. Results from this study also indicated that at the same THP temperature, despite  
309 the increase in solids concentration from 1% to 7%, CH<sub>4</sub> production gradually decreased. This  
310 decline in CH<sub>4</sub> production could have been due to the higher TS concentration causing mass  
311 transfer limitations and reduced sludge fraction and solubility.

312 The results of this study showed that THP is an effective method for transforming WAS into  
313 easily biodegradable substrates with higher solids reduction regardless of the change in TS  
314 concentration of the sludge. Nevertheless, there exists an optimal pretreatment temperature,  
315 and the highest CH<sub>4</sub> production in this study was achieved at 180 °C (Table 3).

316 Compared with the biomethane production values from previous studies (e.g., 0.314 m<sup>3</sup>  
317 CH<sub>4</sub>/kg VS<sub>add</sub> or 0.217 m<sup>3</sup> CH<sub>4</sub>/kg COD<sub>add</sub> for 1.45% TS (Bougrier et al., 2007), 0.337 m<sup>3</sup>  
318 CH<sub>4</sub>/kg VS<sub>add</sub> or 0.219 m<sup>3</sup> CH<sub>4</sub>/kg COD<sub>add</sub> at 2.19% TS (Del Río et al., 2011), 0.261 m<sup>3</sup>  
319 CH<sub>4</sub>/kg COD<sub>add</sub> (Yang et al., 2010), 0.215 m<sup>3</sup> CH<sub>4</sub>/kg COD<sub>add</sub> at 4.6% TS (Mottet et al., 2009),  
320 and 0.286 m<sup>3</sup> CH<sub>4</sub>/kg VS<sub>add</sub> at 7.68% TS (Donoso-Bravo et al., 2011)), this study showed  
321 much higher CH<sub>4</sub> yields (Table 3). Higher CH<sub>4</sub> yields consequently led to high potential  
322 energy recovery, thereby allowing more widespread application in treating WAS both in  
323 facility retrofitting and in new facility designs.

324 [Insert Table 3]

325 Table 3: Biomethane yield during the biomethane potential test.

#### 326 4. Conclusions

327 THP was performed by targeting WAS with TS contents of 1%, 3%, 5%, and 7%.  
328 Physicochemical features for pretreated sludge and CH<sub>4</sub> production efficiency were analyzed.  
329 In all samples, dissolved substances increased by increasing pretreatment temperature. The  
330 CH<sub>4</sub> production was significantly increased in the samples with THP and reached 1.76 to  
331 1.86-fold compared with that of un-pretreated WAS samples. The optimum temperature for  
332 THP was 180 °C for achieving the highest CH<sub>4</sub> production. THP is an effective method for  
333 solubilizing WAS regardless of high sludge concentration. Therefore, THP is expected to  
334 greatly improve both the capacity and efficiency of the commercial AD process. Although  
335 THP offers numerous technological advantages for AD, it is still necessary to assess its  
336 economic feasibility at larger scales before practical application.

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