

Performance and Mechanism of Autogenous Self-healing in Cementitious Composites Materials

by Caihong Xue

Thesis submitted in fulfilment of the requirements for the degree of

DOCTOR OF PHILOSOPHY

under the supervision of

Principal Supervisor:Professor Vute SirivivatnanonCo-Supervisor:Professor Jianchun LiCo-Supervisor:Dr Marie Joshua TapasExternal Supervisor:Professor Kejin WangExternal Supervisor:Professor Caijun Shi

University of Technology Sydney Faculty of Engineering and Information Technology

Sep 2021

CERTIFICATE OF ORIGINAL AUTHORSHIP

I, *Caihong Xue*, declare that this thesis, is submitted in fulfilment of the requirements for the award of Doctor of Philosophy, in the School of Civil and Environmental Engineering, Faculty of Engineering and Information Technology at the University of Technology Sydney.

This thesis is wholly my own work unless otherwise referenced or acknowledged. In addition, I certify that all information sources and literature used are indicated in the thesis.

This document has not been submitted for qualifications at any other academic institution.

This research is supported by the Australian Government Research Training Program.

Signature: Production Note: Signature removed prior to publication.

Date: Sep 20, 2021

Acknowledgements

Completing the Ph.D journey is never a solitary effort, and as always, I am extremely grateful to many people and feeling really lucky to have them in my life. On the top of the list are my supervisors, Professor Vute Sirivivatnanon, Professor Jianchun Li, Dr Marie Joshua Tapas, Professor Kejin Wang and Professor Caijun Shi. More than anything, they supported me when I felt the most helpless, and that gave me the confidence to begin again, to follow my dreams and to be myself. I keep telling myself, if possible in the future, I want to be a supervisor like them to deliver all the kindness and help that I received from them to others, always considering the needs of students, spending time on student's work, giving the student freedom to do research in their own way and providing technical guidance for further improvement. Moreover, thanks for their guidance, comments and patience on my dissertation. Without out them, I will never get there.

I would also like to thank Professor Hadi Khabbaz for helping me go through the most difficult time.

I am also extremely thankful to Professor Arnaud Castel for the handwritten comments on my work, which inspired me to work hard on research. Likewise, I am thankful to Dr Paul Thomas for his detailed comments and explanation on test methods and results, allowing me to learn faster and efficiently.

I would like to thank UTS Mortar and Concrete lab, UTS Science, UTS Environmental Lab and UTS workshop for providing the facilities and equipment. I would also like to thank Mr Rami Haddad, Muller Hailu, Herbert Yuan, Mohammed. Johir, Scott. Graham and Miss Ann Yan for providing professional technical assistant. I would like to thank my friends from UTS for making my PhD life colourful and memorable. Thanks for accompanying me to finish the Sydney Marathon and enjoy the beauty of Australia.

I would like to thank my life friend in China for the trust and encouragement. Thanks for always being there for me.

I am extremely grateful to my mother, for her unconditional love and support. Finally, the little girl is no longer a student.

Lastly, I would like to thank China Scholarship Council (CSC) and UTS for providing financial support for my study.

List of Publications

- Effect of chloride ingress on self-healing recovery of smart cementitious composite incorporating crystalline admixture and MgO expansive agent, *Cement and Concrete Research*, vol. 139.
- Numerical investigation on interface crack initiation and propagation behaviour of self-healing cementitious materials, *Cement and Concrete Research*, vol. 122.
- Self-healing efficiency and crack closure of smart cementitious composite with crystalline admixture and structural polyurethane, *Construction and Building Materials*, vol. 260.

- Novel experimental and numerical investigation on bonding behaviour of crack interface in smart self-healing concrete, *Smart Materials and Structures*, vol. 29.
- Effect of incompatibility between healing agent and cement matrix on self-healing performance of intelligent cementitious composite, *Smart Materials and Structures*, vol. 29.

- A review study on encapsulation based self-healing for cementitious materials, Structural Concrete, vol. 20.
- Self-healing performance of cementitious composite in marine environments-A prospect in Australia *Concrete in Australia*, vol. 46.

Table of Contents

| Certificate of Original Authorshipi |
|---|
| Acknowledgementsii |
| List of Publicationsiv |
| List of Figuresx |
| List of Tablesxv |
| List of Abbreviationsxvi |
| Abstractxxvi |
| 1. Introduction |
| 1.1. Objectives |
| 1.2. Significance of the research |
| 2. Literature review |
| 2.1. Introduction |
| 2.2. Self-healing cementitious composite materials (SHCCMs)7 |
| 2.2.1. Autogenous self-healing |
| 2.2.2. Autonomous self-healing12 |
| 2.2.3. Methods for enhancing autogenous self-healing13 |
| 2.3. Cracking and self-healing on properties of cementitious composites |
| 2.3.1. Cracking and crack closure measurement |
| 2.3.2. Mechanical properties |
| 2.3.3. Durability |
| 2.4. Characterization of self-healing products |
| 2.5. Summary |
| 3. Mechanical properties recovery assessment for autogenous self-healing in water61 |
| 3.1. Introduction |
| 3.2. Experimental investigation |
| 3.2.1. Raw materials and specimen fabrication63 |
| 3.2.2. Crack inducing and self-healing exposure64 |

| 3.2.3. Mechanical properties recovery assessment | 66 |
|--|-------------|
| 3.2.4. Interface between self-healing products and crack surface | 72 |
| 3.3. Experimental results and discussion | 72 |
| 3.3.1. Flexural capacity recovery | 73 |
| 3.3.2. Compressive properties recovery | 75 |
| 3.3.3. Interface between self-healing products and crack surface | |
| 3.4. Conclusions | |
| 4. Performance and mechanism of autogenous self-healing in water | and NaCl |
| solutions | |
| 4.1 Introduction | 85 |
| 4.2 Experiment investigation | 86 |
| 4.2.1 Raw materials and specimen preparation | 86 |
| 4.2.2. Crack inducing and self-healing exposure | |
| 4.2.3. Self-healing performance | |
| 4.2.4 Characterization of self-healing products and pastes | 92 |
| 4.3. Experimental results | |
| 4.3.1. Crack closure ratio | |
| 4.3.2. Recovery of flexural properties | 96 |
| 4.3.3. Characterization of self-healing products and pastes | |
| 4.4. Discussion on self-healing mechanism under chloride attack | 111 |
| 4.4.1. Effects of unhydrated cement particles | 111 |
| 4.4.2. Effects of crystalline admixtures | 112 |
| 4.4.3. Effects of the MgO-type expansive binder | 114 |
| 4.4.4. Effect of chloride on the self-healing reaction | 115 |
| 4.5. Conclusions | 117 |
| 5. Influence of autogenous self-healing on chloride penetration in NaCl solution | ons and sea |
| water | |
| 5.1 Introduction | 120 |
| 5.2 Experimental investigation | 120 |
| 5.2.1 Raw materials and specimen preparation | 120 |
| 5.2.2. Crack inducting and self-healing exposure | |
| J.2.2. Crack inducting and sen-incaring exposure | 121 |

| 5.2.3. Self-healing performance assessment | 124 |
|---|-------------|
| 5.2.4. Chloride ingress, elemental profile and phase alteration in matrix . | |
| 5.2.5. Self-healing products formed in sea water | 129 |
| 5.3. Experimental results | 130 |
| 5.3.1. Self-healing performance assessment | 130 |
| 5.3.2. Chloride ingress, elemental profile and microstructure alterat | tion in the |
| matrix | 137 |
| 5.3.3. Self-healing products formed in sea water | 147 |
| 5.4. Effects of self-healing on the chloride penetration | 155 |
| 5.4.1. In sea water | 155 |
| 5.4.2. In sodium chloride solutions | 157 |
| 5.4.3. Chloride penetration affected by wet/dry cycles | 157 |
| 5.5. Conclusions | 160 |
| 6. The hydration and leaching behaviour of SHCCMs affected by CAs | 163 |
| 6.1. Introduction | |
| 6.2. Experimental investigation | |
| 6.2.1. Materials and specimen preparation | |
| 6.2.2. Effects of CAs on physical properties and heat flow | 166 |
| 6.2.3. Effects of CAs on leaching and pH | 166 |
| 6.2.4. Effects of CAs on hydration products | 167 |
| 6.3. Experimental results and discussion | 167 |
| 6.3.1. Setting time, workability and compressive strength | 167 |
| 6.3.2. Hydration heat | 170 |
| 6.3.3. Leaching and pH | |
| 6.3.4. Hydration products | |
| 6.4. Conclusions | |
| 7. Conclusions and future work | 195 |
| 7.1. The assessment of self-healing on mechanical properties recovery | 195 |
| 7.2. The interaction between self-healing and chloride penetration | 197 |
| 7.3. The self-healing mechanism of SHCCMs with CAs | 198 |
| 7.4. Recommendations for future work | |

| Appendices | |
|------------|--|
| Appendix A | |
| Reference | |

List of Figures

| Fig. 2.1 Microcracks used to track the autogenous self-healing due to continued hydration |
|---|
| in the CO ₂ -free environment9 |
| Fig. 2.2 Autogenous self-healing process due to carbonation |
| Fig. 2.3 Crack closure of cement paste and Ca(OH) ₂ activated GGBFS paste in water. 14 |
| Fig. 2.4 Effects of calcium sulfoaluminate-based expansive additive (CSA) or/and |
| crystalline additives (CAs) on the crack closure16 |
| Fig. 2.5 Self-healed crack in specimens with CSA17 |
| Fig. 2.6 SEM image of hydration products of concere with CAs20 |
| Fig. 2.7 Caclulation of the crack area |
| Fig. 2.8 BSE imaging and EDS analysis of a self-healed crack (200 μ m) submerged in |
| seawater for 28 days. Area that are "yellow" of the EDS images representing calcium, |
| "blue" representing magnesium and "violet" representing silicate |
| Fig. 2.9 Crack volume change measured by µCT26 |
| Fig. 2.10 Uniaxial tensile tests |
| Fig. 2.11 Splitting tensile test set-up for inducing a singe crack |
| Fig. 2.12 The flexural stress-crack opening displacement curves of deflection-hardening |
| HPFRCCs with fibers distributed parallel to the axis of the beams |
| Fig. 2.13 Typical load-displacement curve of the pull-out test |
| Fig. 2.14 Effect of COD on the water permeability |
| Fig. 2.15 Setup for measuring the gas permeability of cementitious composite materials. |
| |
| Fig. 2.16 The gas permeability of cracked cylinders associated with the damge of elastic |
| modulus45 |

| Fig. 2.17 Effect of crack width created by splitting tensile loading on the increase of gas |
|---|
| permeability |
| Fig. 2.18 Effects of cracks on chloride profiles measured by EPMA51 |
| Fig. 2.19 Cracks on carbonation-induced corrosion |
| Fig. 2.20 SEM-EDS points analysis of self-healing products of cement due to further |
| hydration |
| Fig. 2.21 SEM-EDS analysis of self-healing products formed in ECC with CSA |
| Fig. 3.1 Morphology and XRD pattern of the used crystalline admixture admixture64 |
| Fig. 3.2 Three-points bending test setup with the crack opening displacement control |
| using an extensometer |
| Fig. 3.3 Pre-cracking and reloading of mortar prisms using three-points bending under |
| different loading levels |
| Fig. 3.4 Schematic diagram of SDI and PDI from the stiffness damage tests |
| Fig. 3.5 Cyclic compressive loading/unloading procedure for self-healing performance |
| assessment |
| Fig. 3.6 The flexural load-crack opening displacement curves of the steel rebar reinforced |
| prisms before cracking and after self-healing74 |
| Fig. 3.7 Effects of loading/unloading cycles on the compressive properties77 |
| Fig. 3.8 The effect of loading/unloading cycles and self-healing on the stress-strain curves. |
| |
| Fig. 3.9 Effects of cracking and self-healing on the compressive properties |
| Fig. 3.10 Autogenous self-healing products in surface cracks |
| Fig. 3.11 Autogenous self-healing products fomed in internal cracks and pores |
| Fig. 4.1 Self-healing observation |

| Fig. 4.2 Flexural load and crack opening displacement curves of the fiber-reinforced91 |
|---|
| Fig. 4.3 Crack closure ratio of self-healing in different chloride environments |
| Fig. 4.4 Self-healed cracks under different environmental conditions |
| Fig. 4.5 Effects of chloride penetration from NaCl solutions on the flexural properties of |
| the cracked and uncracked specimens97 |
| Fig. 4.6 The flexural load and crack opening displacement curves of the original and the |
| self-healed cracked specimens |
| Fig. 4.7 Morphology of self-healed cracks and elemental compositions at different spots |
| (CMA90 in Cl-2) |
| Fig. 4.8 XRD analysis on the self-healing products and cement pastes under different |
| |
| Fig. 4.9 FTIR curves of self-healing products |
| Fig. 4.10 FTIR curves of cement pastes after self-healing |
| Fig. 4.11 Quantification of self-healing products based on the TG/DTG curves 108 |
| Fig. 4.12 Characterization of cement pastes exposed to different solutions |
| Fig. 4.13 Microstructure of the self-healed crack in C100 exposed to water |
| Fig. 4.14 Morphology of self-healing products of CA100 exposed to distilled water (the |
| marked yellow line was the approximate boundary) |
| Fig. 4.15 Self-healing mechanism of cementitious composite with CAs and MgO: (a) |
| unhyrated cement (UHC), CAs and unhydrated MgO grains (UMGs) exposed to the crack |
| solution; (b) various ions diffusion to crack solution to saturation; (c) crystals formation; |
| and (d) carbonation of crystals114 |
| Fig. 4.16 Self-healing products contributed by the hydration and carbonation of MgO (the |
| marked yellow line was the approximate boundary) |

| Fig. 4.17 Comparison on Ca/Si and Al/Si of the self-healing products of C100 and CA100, |
|---|
| the CH and C-S-H in C100 and CA100, exposed to the 0.545M NaCl solution |
| Fig. 5.1 Splitting tensile test setup with DIC |
| Fig. 5.2 Crack closing measurement by 3D laser scanning |
| Fig. 5.3 P_{split} -COD curves of the original, cracked and self-healed cementitious |
| composites |
| Fig. 5.4 Sample preparation for microstructural analysis |
| Fig. 5.5 Crack closure ratio of cracked specimens exposed to different solutions 131 |
| Fig. 5.6 Self-healed cracks in CMA90 after three months exposure in different solutions. |
| |
| Fig. 5.7 Self-healing on splitting tensile properties |
| Fig. 5.8 Self-healing on water absorption |
| Fig. 5.9 Effects of self-healing on the chloride penetration |
| Fig. 5.10 The elemental profiles of the matrix exposed to sea water and Cl-2141 |
| Fig. 5.11 TGA for different mortar after 3 months of wet/dry cycles in water, sea water, |
| Cl-1 solution and Cl-2 solution |
| Fig. 5.12 Effects of the exposure environments on the phase alteration in the mortar |
| matrix |
| Fig. 5.13 SEM-EDS mapping of the self-healed cracks |
| Fig. 5.14 SEM-EDS points analysis results of healing products of CMA90152 |
| Fig. 5.15 Quantification of healing products formed in different solutions |
| Fig. 5.16 Morphology of the self-healing products formed in natural sea water |
| Fig. 5.17 The change of the micrtostucture in the matrix exposed to the saline solutions. |
| |

| Fig. 5.18 Expansion of specimens after exposure to different solutions |
|--|
| Fig. 6.1 XRD of two CAs |
| Fig. 6.2 DTG curves of the original CAs powder165 |
| Fig. 6.3 Effects of CAs on the setting time of pastes |
| Fig. 6.4 Mortar workability and compressive strength |
| Fig. 6.5 Hydration-related heat evaluation of pastes with CA-α |
| Fig. 6.6 Hydration-related heat evaluation of pastes with CA-β172 |
| Fig. 6.7 Leaching of ions and pH values of pastes with CA- α . The values marked as |
| "calculated" for the concentration of Na and K were calculated from oxides (Na2O and |
| K ₂ O) in the raw materials |
| Fig. 6.8 Leaching assessment and pH values of pastes with CA-β179 |
| Fig. 6.9 XRD pattern of hydration products of pastes with CA- α at 7 d and 28 d; U: U- |
| phase, E:ettringite; P: portlandite; T: thenardite; C:calcite; A; alite; B: Belite; Q;quartz; |
| CSH: Calcium-Silicate-Hydrates; MH: brucite; Hc: hemicarboaluminate; |
| Mc:monocarboaluminate and G: gypsum |
| Fig. 6.10 XRD pattern of hydration products of pastes with CA- β at 7 d and 28 d 183 |
| Fig. 6.11 TG and DTG curves of hydration products of pastes with CA- α at 7 d and 28 d. |
| |
| Fig. 6.12 DTG curves of hydration products of pastes with CA- β at 7d and 28 d 186 |
| Fig. 6.13 Mass percentage of Ca(OH) ₂ and CaCO ₃ |
| Fig. 6.14 SEM-EDS analysis of hydration products of Ca100191 |
| Fig. 6.15 SEM-EDS analysis of hydration products of Cβ100192 |

List of Tables

| Table 3.1 Chemical compositions of cement and crystalline admixture | 64 |
|---|-------|
| Table 3.2 Mix proportions of mortars | 64 |
| Table 3.3 Loaing/unloading cycles performed on mortar cubes to induce cracks | 75 |
| Table 3.4 The effect of self-healing on the compressive properties recovery | 80 |
| Table 4.1 Oxides in the MgO powder | 87 |
| Table 4.2 Physical properties of polyvinyl alcohol (PVA) fiber | 87 |
| Table 4.3 Mix design of self-healing cementitious composites | 87 |
| Table 4.4 Concentration of chloride in solutions (mol/L) | 88 |
| Table 5.1 Mix proportions | .121 |
| Table 5.2 Elemental composition of exposure solutions and 7-day pore solution (mme | ol/L) |
| | .123 |
| Table 5.3 Numbers of specimens for the different post-conditioning states | .123 |
| Table 6.1 Oxides in raw materials | .164 |
| Table 6.2 Mix proportions | .166 |
| Table 6.3 The Na ₂ O content and the expected Na concentration in filtrate | .177 |

List of Abbreviations

| AE | Acoustic Emission Analysis |
|------|---|
| ASTM | American Society for Testing and Materials |
| BSE | Backscattered Electron |
| BN | Expansive Bentonite |
| СТ | X-ray computed tomography |
| CA-a | Notation for one of the used crystalline admixtures |
| CA-β | Notation for one of the used crystalline admixtures |
| Cl-1 | The 0.545 mol/L sodium chloride solution |
| Cl-2 | The 2 mol/L sodium chloride solution |
| DIC | Digital Image Correlation |
| DTT | Uniaxial Direct Tensile Test |
| DTG | Derivative Thermogravimetry |
| EGC | Engineered Geopolymer Composites |
| ECC | Engineered cementitious composites |
| EDS | Energy Dispersive-X-ray Analysis |
| EPMA | Electron Probe Micro-Analysis |
| FRC | Fiber Reinforced Concrete |
| FTIR | Fourier Transform Infrared Spectroscopy |

| GP | General Purpures Cement |
|--------|---|
| GGBFS | Ground Granulate Blast Furnace Slag |
| HPFRC | High Performance Fiber Reinforced Concrete |
| HPFRCC | High Performance Fibre Reinforced Cementitious Composites |
| HPC | High Performance Concrete |
| HMCs | Hydrated Magnesium Carbonates |
| ICP-MS | Inductively Coupled Plasma Mass Spectrometry |
| IC | Ion Chromatography |
| L | Lime |
| LVDTs | Linear Variable Displacement Transducers |
| NMR | Nuclear Magnetic Resonance |
| NSC | Normal Strength Concrete |
| ОМ | Optical Microscope |
| RCPT | Rapid Chloride Permeability Test |
| RMC | Magnesium-based Concrete |
| PLC | Portland Limestone Cement |
| PVA | Polyvinyl Alcohol |
| SDT | Stiffness Damage Test |
| SCMs | Supplementary cementitious materials |

| SHCCMs | Self-healing cementitious composites materials |
|--------|--|
| SF | Silica fume |
| Т | Temperature |
| TEM | Transmission Electron Microscopy |
| TG | Thermogravimetry |
| TGA | Thermogravimetry Analysis |
| TRF | Transverse Resonant Frequency |
| UHC | Unhydrated Cement Particles |
| UHPC | Ultra-High-Performance-Concrete |
| UMGs | Unhydrated MgO Grains |
| w/d | Wet/dry cycles |
| W | Wetting |
| XRD | X-ray Diffraction |
| XRF | X-ray Fluorescence |

Cement chemistry

| Al ₂ O ₃ | Aluminium oxide |
|--------------------------------|--|
| AH3 | Aluminium hydroxide (Al(OH)3) |
| AFm | Aluminate ferrite monosulfate |
| AFt | Al ₂ O ₃ -Fe ₂ O ₃ -trisulfate |

| В | Brucite |
|-------------------|---|
| С | Calcite |
| Ca ²⁺ | Calcium ions |
| C-A-S-H | Calcium aluminium silicon hydrate |
| CAs | Crystalline additives |
| C4AF | Tetracalcium aluminoferrite |
| CaO | Calcium oxide |
| Cc | Calcium carbonates (CaCO ₃) |
| СН | Calcium hydroxide (Ca(OH)2) |
| CO3 ²⁻ | Carbonate |
| CO ₂ | Carbon dioxide |
| CSA | Sulfoaluminate-based expansive additive |
| C_2S | Belite |
| C ₃ S | Alite |
| C-S-H | Calcium silicate hydrates |
| Cl | Chloride |
| Cl | Chloride ions |
| E | Ettringite |
| Fs | Friedel's salt |

| G | Gypsum |
|---|--------------------------------|
| Нс | Hemicarboaluminate |
| HNO ₃ | Nitric acid |
| K | Potassium |
| K ₂ O | Potassium oxide |
| Mc | Monocarboaluminate |
| Mg | Magnesium |
| MgO | Magnesium oxide |
| MH | Brucite (Mg(OH) ₂) |
| M-S-H | Magnesium Silicate Hydrate |
| Ms | Monosulfoaluminate |
| МК | Metakaolin |
| Na | Sodium |
| Na ₂ CO ₃ | Sodium Carbonates |
| NaHCO ₃ | Sodium bicarbonate |
| Na ₂ O | Sodium oxide |
| Na2SO4-H2O | Sodium sulfate salt |
| Na ₂ SO ₄ | Sodium sulfate |
| Na ₂ SO ₄ ·10H ₂ O | Mirabilite |

| Р | Portlandite |
|-------------------|---|
| Q | Quartz (SiO ₂) |
| S | Sulfur |
| SO ₃ | Sulfur trioxide |
| SiO ₂ | Sulfur oxide |
| SO4 ²⁻ | Sulfate |
| Т | Thenardite (Na ₂ SO ₄) |

Roman symbols (lowercase)

| а | Area of the specimen |
|-------------------------|--|
| С | Crack closing ratio |
| d | Density of the water |
| fpeak_virgin | Peak flexural strength of the virgin specimen |
| $f_{peak_self-healed}$ | Peak flexural strength of the self-healed specimen |
| f_c | Compressive strength |
| $f_{c_original}$ | Compressive strength of the original specimen |
| $fc_cracked$ | Compressive strength of the cracked specimen |
| fc_{self} -healed | Compressive strength of the self-healed specimen |
| mi | Mass change in grams |
| t | Self-healing duration |

| Wi | Width of a crack before self-healing |
|----|---------------------------------------|
| Wt | Width of the crack after self-healing |

Roman symbols (uppercase)

| Ai | Area of crack mouth before self-healing |
|----------------------------------|---|
| At | Area of crack mouth after self-healing |
| Eoriginal | Elastic modulus of the original specimen |
| Ecracked | Elastic modulus of the cracked specimen |
| ERI | Compressive elastic modulus recovery index |
| Ι | Slope of the best fit line for water absorption |
| IDaRUPV | Stiffness/damage recovery measured by UPV |
| ISR | Index of flexural strength recovery |
| ISRpre-peak | ISR for pre-peak cracked specimens |
| ISR _{post-peak} | ISR for post-peak cracked specimens |
| COD | Crack Opening Displacement |
| Kcracked | Gas permeability coefficient of the cracked specimen |
| Kself-healed | Gas permeability coefficient of the self-healed specimen |
| K flexural_self-healed | Slope of <i>P</i> _{flexural} - <i>COD</i> for the self-healed specimen |
| $K_{\mathit{flexural_cracked}}$ | Slope of <i>P</i> _{flexural} - <i>COD</i> for the cracked specimen |
| $K_{flexural_original}$ | Slope of <i>P</i> _{flexural} -COD for the original specimen |

| K_{split_self} -healed | Slope of <i>P_{split}-COD</i> for the self-healed specimen |
|-------------------------------------|--|
| $K_{split_cracked}$ | Slope of <i>P_{split}-COD</i> for the cracked specimen |
| $K_{split_original}$ | Slope of <i>P_{split}-COD</i> for the original specimen |
| Psplit | Splitting tensile load |
| $P_{split_unloading}$ | Unloading splitting tensile load |
| $P_{split_max_principal}$ | Principle peak splitting tensile load of the original specimen |
| $P_{split_max_secondary}$ | Secondary peak splitting tensile load of the original specimen |
| $P_{split_max_self-healed}$ | Maximum splitting tensile load of the self-healed specimen |
| $P_{split_cracking_original}$ | Splitting tensile cracking load of the original specimen |
| $P_{split_cracking_self_healed}$ | Splitting tensile cracking load of the self-healed specimen |
| PRIsplitting | Splitting tensile load capacity recovery |
| P_{self} -healed | Water permeability of the self-healed specimen |
| Pcracked | Water permeability of the cracked specimen |
| $P_{flexural_max_original}$ | Maximum flexural load of the original specimen |
| $P_{flexural_unloading}$ | Unloading flexural load |
| $P_{flexural_max_cracked}$ | Maximum flexural load of the cracked specimen |
| PDI | Plastic damage index |
| PRI | Plastic damage recovery index |
| PRIflexural | Recovery of the flexural load capacity |

| R_p | Relative water permeability |
|-----------------------------|---|
| Si | Initial water sorptivity |
| $S_{i,wet/dry,self-healed}$ | Initial sorptivity of the self-healed specimen |
| $S_{i,wet/dry,original}$ | Initial sorptivity of the original specimen |
| Ss | Secondary water sorptivity |
| $S_{s,wet/dry,uncracked}$ | Secondary sorptivity of the self-healed specimen |
| $S_{s,wet/dry,original}$ | Secondary sorptivity of the original specimen |
| SDI | Stiffness damage index |
| SDIcracked | Stiffness damage index of the cracked specimen |
| SDIself-healed | Stiffness damage index of the self-healed specimen |
| SDIoriginal | Stiffness damage index of the original specimen |
| SDIstrength | Compressive strength reduction index |
| SRIc_stiffness | Compressive stiffness reduction recovery index |
| SRIc_strength | Compressive strength recovery index |
| $SRI_{flexural, stiffness}$ | Flexural stiffness recovery index |
| SRI splitting, stiffness | Splitting tensile stiffness recovery |
| UPVself-healed | Ultrasonic pulse velocity of the self-healed specimen |
| UPV cracked | Ultrasonic pulse velocity of the cracked specimen |
| UPVvirgin | Ultrasonic pulse velocity of the virgin specimen |

Creek symbols

Compressive stress

 σ_c

| σ_N | Nominal flexural stress |
|-----------------------------------|--|
| σ_{N_loss} | Nominal flexural strength loss due to cracking |
| $\sigma_{N_Inherent}$ | Inherent nominal flexural strength regain |
| $\sigma_{N_unloading_virgin}$ | Unloading nominal flexural strength of virgin specimen |
| $\sigma_{N_unloading_cracking}$ | Unloading nominal flexural strength during cracking |
| η_{gas} | Recovery of the gas permeability of the cracked specimen |

Abstract

The mechanical properties and durability of concrete structures can be seriously impaired by cracks. The need to reduce the risk of cracking and repair the damage caused by concrete cracks led to the development of self-healing cementitious composite materials (SHCCMs). SHCCMs are able to heal cracks without human intervention and can generally be classified into either autogenous self-healing or autonomous self-healing. In comparison with autonomous self-healing, autogenous self-healing is more cost effective and easier to implement in full-scale application, while it is limited to healing crack widths of about 100-150 µm.

Supplementary cementitious materials (SCMs), expansive minerals and crystalline admixtures (CAs) are used to promote the autogenous self-healing of cementitious composite materials and develop SHCCMs. However, the absence of standardized test procedure and performance assessment criteria for self-healing resulted in disagreement with regard to the effectiveness of stimulated autogenous self-healing in concrete. The benefit of SHCCMs to durability of cracked concrete in chloride solutions and marine environments is not well understood. Moreover, in order to tailor self-healing reactions of SHCCMs for better performance, it is necessary to understand the mechanisms of stimulated autogenous self-healing. Currently, these issues are hindering the application and development of SHCCMs.

In this study, the methodology for evaluating the influence of autogenous self-healing on the mechanical properties and durability performance of concrete was developed and validated. Cracked specimens with and without CAs were exposed to water, sodium chloride solutions and seawater to facilitate self-healing. Afterwards, the effect of selfhealing on the compressive, flexural and splitting tensile properties, water absorptivity as well as chloride penetration of cracked materials was assessed. The mineralogy of selfhealing products was also characterized to reveal the mechanism of autogenous selfhealing. Furthermore, the influence of CAs on the hydration and leaching behaviour of cement was investigated, in order to explore factors that can help improve the autogenous self-healing of cementitious composites materials.

With regards to mechanical properties, autogenous self-healing improves the stiffness but barely affects the load capacity of cracked specimens under compressive, flexural and splitting tensile loading. Indexes that are related to stiffness are recommended for selfhealing performance assessment. The results also highlight that exposure environments affect self-healing mechanisms (reaction) and subsequently, self-healing performance. The rate of crack closure is fastest in seawater followed by NaCl solutions and water, which results in the lower chloride penetration into self-healed cracked specimens from seawater than NaCl solution. The rapid precipitation of Mg(OH)₂ in cracks dominates the self-healing in seawater. Compared to water, NaCl solutions accelerate self-healing by promoting the precipitation of CaCO₃ in cracks. This suggests that a faster rate of self-healing helps improve the durability of cracked structures more than the type of products that form during self-healing. Moreover, the hydration and leaching behaviour of cement incorporating CAs indicates that a higher pH of the pore solution and a reasonable degree of carbonation could benefit the self-healing of cementitious composite materials.