1	Tracing historical changes, degradation, and original sources of airborne
2	polycyclic aromatic hydrocarbons (PAHs) in Jilin Province, China, by Abies
3	holophylla and Pinus tabuliformisneedle leaves
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#### **Abstract:** 22

Due to their wide distribution and availability, plant leaves are interesting biomonitoring 23 candidates for the evaluation of atmospheric pollution. In addition, leaves from some species 24 can also retain historical information, for example related to environmental pollution, due to 25 their class age. In this study, the content of polycyclic aromatic hydrocarbons (PAHs) in Abies 26 holophylla and Pinus tabuliformis needles in function of their class age have been investigated 27 to obtain information regarding the degradation constant for each PAH under investigation, and 28 to evaluate the possibility to correlate the presence of PAHs in needles with some important 29 environmental factors. It has been demonstrated that the total PAH concentration in the needles, 30 for both species, increased proportionally with the age of needles, ranging from 804 to 3604 ng 31  $g^{-1}$  (dry weight), showing a general tendence to accumulate these substances through years, 32 while degradation rates increased with molecular complexity. Finally, a study on the possible 33 correlation between adsorbed PAH contents in needle leaves and pollution emission sources 34 have been carried out, demonstrating that this biomonitoring system could be fruitfully used to 35 trace historical changes and original sources of airborne PAHs. 36

Keywords: Polycyclic aromatic hydrocarbons (PAHs); Air pollution biomonitoring; PAH 37 degradation constants; Abies holophylla; Pinus tabuliformis 38

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## 46 **1. Introduction**

The determination of polycyclic aromatic hydrocarbons (PAHs) level caused by air 47 pollution is one of the problems of major concern in recent decades. Airborne PAHs are 48 produced by incomplete combustion of industrial emissions and transportation (Chang et al., 49 2006; Rodgman et al., 2000), as well as wildfire (Choi, 2014; Lao et al., 2018), and their 50 monitoring has been considered a challenging and important activity, due to the carcinogenic, 51 mutagenic, and teratogenic effects of these substances on human health. PAHs determination 52 anyway could encounter technical, physical, and economical limitations when active air 53 samplers are used, while sampling tree components like needles, leaves and barks, suitable for 54 the determination of the spatial distribution of PAHs, are usually employed only to measure the 55 current levels of atmospheric PAHs (Amigo et al., 2011; De Nicola et al., 2011; Zhou et al., 56 57 2014). Recent studies on the historical record of PAHs have been conducted in the growth rings, lake sediments and ice cores of specific areas (Cai et al., 2016; Kuang et al., 2015; Wang et al., 58 2008), but these sampling sites impose several constrains, limiting the monitoring to specific 59 conditions in terms of spatial recognition and time intervals. 60

Plants can be considered a very interesting and advantageous biomonitoring substrate in 61 particular for PAHs, due to the hydrophobic lipid layers that can retain these substances (Kim 62 et al., 2014; Paterson et al., 1991). For example, it has been demonstrated that diffusive uptake 63 and storage function of polymeric lipids of plants play a key role in the transport and fate of 64 phenanthrene (Li and Chen, 2014). A unique and very useful characteristic of conifer needles 65 is that they can grow up and last for several years, thus preserving the history of the 66 environmental conditions. Furthermore, some PAHs can be used as a marker of certain pollution 67 sources: for example, it was found that the prevalence of lower molecular weight compounds 68 in ambient air samples can be essentially related to the road traffic, more specifically with 69 vehicles of diesel engines (Albuquerque et al., 2016). There have also been variation studies 70 using some specific PAH content ratio in needles, such as Ant/(Ant + Phen) or Flt/(Flt + Pyr), 71 to identify local sources of contamination (Esen et al., 2008; Galarneau, 2008; Kong et al., 72

73 2015).

On the base of these considerations, *Abies holophylla* needles from 1 to 6 years old (2016-2011) old and *Pinus tabuliformis* from 1 to 4 years old (2015-2012) have been collected to investigate the species-specificity historical variations of PAHs and to investigate the possible relation between emitted air pollutants and PAHs in the needles over the years, also taking into account the PAH degradation into the needle leaves. To pursue this target a mathematical model to interpret the PAH degradation has been proposed. Finally, PAHs content in conifer needles were correlated with pollutant sources of airborne PAHs throughout the studied years.

## 81 2. Material and methods

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## 2.1 Materials and standards

Standard mixtures of 16 PAHs (naphthalene (Naph), acenaphthylene (Acy), acenaphthene 83 (Ace), fluorene (Fluo), phenanthrene (Phen), anthracene (Ant), fluoranthene (Flt), pyrene (Pyr), 84 benzo[a]anthracene (BaA), chrysene (Chry), benzo[b]fluoranthene (BbF), 85 benzo[k]fluoranthene (BkF), benzo[a]pyrene (BaP), indeno[1,2,3-cd] pyrene (IcdP), 86 dibenzo[a,h]anthracene (DahA) and benzo[ghi]perylene (BghiP)) and 3 deuterated PAHs 87 (Phenanthrene-d<sub>8</sub>, Fluoranthene-d<sub>12</sub> and Perylene-d<sub>12</sub>) were purchased from Sigma Aldrich (St. 88 Louis, MO, USA). The purity of all standards was up to 98%. Silica (100-200 mesh) and Al<sub>2</sub>O<sub>3</sub> 89 adsorbent was obtained from Sigma Aldrich (St. Louis, MO, USA). To remove the 90 contaminants, all adsorbents and glasswares were heated at 400°C for 12 h. 91

## 92 **2.2 Sampling and processing**

Two plant species with 4 years and 6 years old needles were selected for this study. Needles from 1 (2016) to 6 years (2011) and from 1 (2015) to 4 years (2012) old were collected from *Abies holophylla* and *Pinus tabuliformis*, respectively. Age of needles was evaluated by growth ring and branching pattern, and it was confirmed in combination with the situation at the branch cut (**Fig. 1**) (Pérez-Harguindeguy et al., 2013). During collection, in order to ensure a better authenticity and accuracy, needles of the same year at the upper 1/3 and lower 1/3 of the canopy
of tree species were mixed into one sample, cleaned, freeze-dried, crushed by 200 mesh, and
stored at -20°C for further analysis.



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Fig. 1. Photographs that show the branch information of A) *Abies holophylla*, and the growth
 ring patterns of B) *Abies holophylla* and C) *Pinus tabuliformis*.

Eleven air pollutant emission variables with an expected environmental impact on the concentrations of PAHs, namely coal (Ton year<sup>-1</sup>), crude oil (Ton year<sup>-1</sup>), and electricity (MKh year<sup>-1</sup>) consumption; number of civil vehicles and trucks (Unit year<sup>-1</sup>); emission of industrial sulfur, nitrogen oxide, industrial smoke (dust), and soot (Ton year<sup>-1</sup>); urban living gaseous pollutant emissions (Ton year<sup>-1</sup>); generating capacity (KWh year<sup>-1</sup>) were obtained for the interested areas from the databases of the China National Statistics Institute ("SBJ (Statistic bureau of Jilin-Statistics China)," 2016).

111 2.3 Measurement of lipid content

Needle lipid contents were measured as follows: 3 g of dry needles were extracted 3 times using ultrasonication with 20 mL n-hexane:acetone (50:50, v:v) for 20 min each. The extract was concentrated, dried and finally weighed to calculate the lipid content (Zhao et al., 2018).

#### 2.4 PAHs analysis

## 116 **2.4.1 Extraction of PAHs**

All the samples were spiked using internal standards. The sample pre-treatment procedures 117 were performed according to the method in our previous publication with slight modification 118 (Jin et al., 2020). In brief, a total of 0.5 g homogenized, crushed and sieved needles powder was 119 spiked with internal standards, then mixed with 10 mL of dichloromethane (DCM) and 120 ultrasonically extracted 3 times per 15 min. The combined extract was replaced with n-hexane 121 and concentrated to 0.5 mL prior to SPE clean up. The SPE column (down-top: 2.0 g Na<sub>2</sub>SO<sub>4</sub>, 122 1.5 g silica gel, 0.8 g Al<sub>2</sub>O<sub>3</sub> (deactivated with 2% ultrapure water), and 4.0 g Na<sub>2</sub>SO<sub>4</sub>) was 123 conditioned with n-hexane before sample loading. It was eluted with 3 mL n-hexane (discarded) 124 followed by 9 mL n-hexane/DCM (1:1, v/v) (first 3 mL was discarded). The 6 mL eluted extract 125 was concentrated via a gentle stream of nitrogen gas at room temperature, and reconstituted 126 with 100 µL of n-hexane. Then, gas purge microsyringe extraction (GP-MSE) was used for 127 further purification, as described in a previous work (Yang et al., 2011), and the final volume 128 was adjusted to 100 µL. Finally, 2 µL of the purified extract was injected to the GC-MS system 129 for analysis. 130

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## 2.4.2 GC-MS analysis

PAHs were analyzed quantitatively and qualitatively by Shimadzu gas chromatograph 132 equipped with DB-5MS fused silica capillary column (30 m  $\times$  0.25 mm; thickness: 0.25  $\mu$ m, 133 Restek Corporation, Bellefonte, PA, USA) and QPMS 2010 mass spectrometry detector. The 134 conditions for GC-MS analysis were as follows: sample injection was carried out in splitless 135 mode at 280°C, using He as a carrier gas at a flow rate of 1 ml min<sup>-1</sup>. GC-MS interface 136 temperature was set at 280°C, ionization voltage was 70 eV and ion source temperature was set 137 at 200°C. The initial oven temperature was held at 80°C for 1 min, then was brought to 100°C 138 at the rate of 20°C min<sup>-1</sup>, from 100°C to 200°C at 10°C min<sup>-1</sup> and 200°C to 280°C at 20°C min<sup>-1</sup> 139 <sup>1</sup>, holding this last value for 2 min. Data was collected in selected ion monitoring mode (SIM). 140

The relative percentage difference for individual PAH determined in paired duplicate samples (n = 2) was always < 24%. The obtained recovery for all the PAHs was between 64 and 120% with respect to the certified values.

144 **3. Results and discussion** 

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## 3.1 Adsorbed PAHs in needles

Usually, PAHs content in plant samples is normalized by lipid content or dry weight. 146 Studies have shown that both cuticle and suberin (a cell wall component) may act as lipids in 147 different parts of the plant, but they are not solvent extractable (Chen et al., 2012; Ockenden et 148 al., 1998; Taiz and Zeiger, 2010). For this reason, the needle lipid content obtained by the 149 classical method may not reflect the total amount of substances acting as PAH extracting lipids. 150 The total measured extractable PAH concentrations found in this work are given in Table 1. 151 The needle lipid contents in Abies holophylla and Pinus tabuliformis ranged from 27.46 to 152 32.16%, and from 6.85 to 9.56% (respect to dry weight), respectively. By comparing the total 153 extractable PAH concentrations with lipid contents of needles, no significant correlation was 154 found (R = -0.12 - 0.646, p > 0.354). Hence PAHs contents were expressed on a dry weight 155 basis in this study. 156

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## Pinus tabuliformis.

Table 1 Temporal trends of PAH total contents recorded in needles of Abies holophylla and

Dinga			Abies holopl	nylla (ng g <sup>-1</sup> )		
Kings	2016	2015	2014	2013	2012	2011
2	$43 \pm 2$	$52 \pm 17$	$42\pm 5$	$38 \pm 1$	$38 \pm 2$	$45\pm7$
3	$981\pm76$	$1503\pm10$	$1669 \pm 13$	$1820\pm82$	$2135 \pm 149$	$2340\pm34$
4	$438 \pm 133$	$588 \pm 98$	$762\pm204$	$815\pm158$	$893\pm84$	$1228 \pm 130$
5,6	$18 \pm 1$	$14 \pm 1$	$9\pm2$	$16 \pm 2$	$10 \pm 4$	$21\pm7$
Total PAHs	$1449 \pm 212$	$2113 \pm 106$	$2446 \pm 189$	$2663\pm73$	$3044 \pm 67$	$3604\pm96$

Dinga	Pinus tabuliformis (ng g <sup>-1</sup> )				
Kings	2015	2014	2013	2012	

2	$119\pm4$	$100 \pm 1$	$105 \pm 25$	$72\pm8$
3	$498 \pm 112$	$829 \pm 102$	$845\pm 66$	$910\pm 39$
4	$152\pm10$	$241\pm22$	$319\pm51$	$308\pm30$
5,6	$34\pm7$	$28\pm3$	$65 \pm 39$	$145\pm77$
Total PAHs	$804\pm99$	$1197 \pm 127$	$1334 \pm 131$	$1435\pm93$

As expected, due to the accumulating effect, PAH contents generally increase with the 160 needles age, accordingly to the results obtained by previous works (Odabasi et al., 2015; Ratola 161 et al., 2010a). Furthermore, Abies holophylla needles generally showed a higher tendence to 162 accumulate PAHs in respect to Pinus tabuliformis needles. The concentration of 3-4 ring PAHs 163 increase over time, and this result is consistent with a previous work (Ratola et al., 2010a). 164 These values were found to be higher than 5-6 ring PAHs, which are in good agreement with 165 the accumulation pattern of PAHs in needles of other pine species, including Pinus halepensis 166 and P. pinea (Librando et al., 2002), Pinus pinaster and Pinus nigra (Piccardo et al., 2005) and 167 *P. pinaster* and *P. pinea* (Ratola et al., 2010a). Because of the volatile nature of Naph (2 ring), 168 its fast photodegradation process and re-suspension capability leads to a non-stable 169 accumulation in the needles (Choi, 2014; Wang et al., 2005). Lighter PAHs (3-4 ring), which 170 are predominantly present in the gaseous phase, interact strongly with the waxy layer of the 171 needles and this effect can promote PAH uptakes in needles (Lehndorff and Schwark, 2004). 172 The low content of heavier PAHs (5-6 rings) observed in this work confirms what already 173 noticed in previous works, and it could be related to the fact that their adsorption in needle 174 leaves can be hindered by their different distribution in the environment. These PAHs are 175 present in fact preferentially in the particulate phase of the atmosphere, and for this reason their 176 migration from atmosphere can be slower if compared with other PAHs (Yang et al., 2017). The 177 accumulation process may vary from species to species, leading to different concentration in 178 needles (Ratola et al., 2011). This behavior has been noticed also for the species in this work, 179 even if a similar PAH trend for years 1 to 4 with Abies holophylla and Pinus tabuliformis 180 needles were noticed. More than 64 % of adsorbed PAHs in the needles are composed of 3-4 181 rings: this value roughly reflects PAH composition in the air in both urban and industrial areas 182 (Li et al., 2016; Odabasi et al., 2015; Tomashuk et al., 2012). 183

## 184 **3.2** A study on PAH degradation constants in *Abies holophylla* needles

As already pointed out, PAHs are present in gas and particle phases of the atmosphere and 185 diffuse into plant leaves through dry and wet deposition. Their sources are closely related to the 186 anthropogenic combustion processes (e.g. car exhaust, local heating facilities, industrial related 187 activities). It has been already demonstrated that the PAH degradation follows a first-order 188 kinetics (Haritash and Kaushik, 2009), even if dependent from external variables: for example, 189 fly ash and carbon black, that can protect PAHs from photo decomposition (Behymer and Hites, 190 1985; Korfmacher et al., 1980). Anyway, in principle the PAH amount collected in different 191 years by needles undergo through a degradation mechanism that follows an exponential trend, 192 which can be described by these equations: 193

$$C_{year}^{Tot} = \sum_{i=0}^{n} C_i e^{-\alpha t_i}$$
 Eq.1

<sup>194</sup> Where  $C_{year}^{Tot}$  is the content (obtained experimentally) for each PAH,  $C_i$  is the PAH <sup>195</sup> amount that can be attributed to a specific year,  $\alpha$  is the degradation kinetic constant and  $t_i$  is <sup>196</sup> the time expressed in years. This model was then applied to PAH contents in *Abies holophylla* <sup>197</sup> needles, considering that for this species data were collected from 2011 to 2016, a longer period <sup>198</sup> if compared to data obtained by *Pinus tabuliformis*. In this case, it is possible to write for each <sup>199</sup> year the following equations:

$$C_{2011}^{Tot} = C_{\le 2010} e^{-\alpha} + C_{2011}$$
 Eq.2

$$C_{2012}^{Tot} = C_{\le 2010} e^{-2\alpha} + C_{2011} e^{-\alpha} + C_{2012}$$
Eq.3

$$C_{2013}^{Tot} = C_{\le 2010}e^{-3\alpha} + C_{2011}e^{-2\alpha} + C_{2012}e^{-\alpha} + C_{2013}$$

$$C_{2014}^{Tot} = C_{\le 2010}e^{-4\alpha} + C_{2011}e^{-3\alpha} + C_{2012}e^{-2\alpha} + C_{2013}e^{-\alpha} + C_{2014}$$
 Eq.4

$$C_{2015}^{Tot} = C_{\le 2010}e^{-5\alpha} + C_{2011}e^{-4\alpha} + C_{2012}e^{-3\alpha} + C_{2013}e^{-2\alpha} + C_{2014}e^{-\alpha} + C_{2015}$$
 Eq.5

$$C_{2016}^{Tot} = C_{\le 2010}e^{-6\alpha} + C_{2011}e^{-5\alpha} + C_{2012}e^{-4\alpha} + C_{2013}e^{-3\alpha} + C_{2014}e^{-2\alpha} +$$
Eq.6

$$C_{2015}e^{-\alpha} + C_{2016}$$

Eq.7

200 Considering  $e^{-\alpha t} = y$  we have:

$$C_{2011}^{Tot} = C_{\le 2010}y + C_{2011}$$
Eq.8
$$C_{2012}^{Tot} = C_{\le 2010}y^2 + C_{2011}y + C_{2012} = (C_{\le 2010}y + C_{2011})y + C_{2012} = C_{2011}^{Tot}y + C_{2012}$$
Eq.9

$$C_{2013}^{Tot} = C_{2012}^{Tot}y + C_{2013}$$
 Eq.10

$$C_{2014}^{Tot} = C_{2013}^{Tot} y + C_{2014}$$
Eq.11

$$C_{2015}^{Tot} = C_{2014}^{Tot}y + C_{2015}$$

$$C_{2016}^{Tot} = C_{2015}^{Tot}y + C_{2016}$$

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On the base of Eq. 9-13, considering that  $C_{year}$  and y should be always  $\geq 0$ , it is possible 202 to individuate the minimum acceptable  $\alpha$  values and the relevant minimum PAH degradation 203 percentage per year (Table 2). To our knowledge, no studies have been carried out regarding 204 the PAH degradation in leaves. Cutright (Cutright, 1995) reported a study to determine the 205 specific degradation rates for the bioremediation of PAH-contaminated soils. Other studies 206 demonstrated that the degradation rate of high molecular weight PAHs is slower than other 207 PAHs (Johnsen et al., 2005), and as an important factor for the PAH-degradation activity, of the 208pH changes by the bacteria could be the reason (Kästner et al., 1998). 209

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 Table 2 Minimum percentage of PAH degradation year<sup>-1</sup> and minimum values of the degradation constant.

Rings	Compound	Minimum degradation per year (%)	Minimum <i>a</i> value
2	Naph	32	0.387
3	Acy	16	0.173
3	Ace	34	0.421
3	Fluo	39	0.501
3	Phen	27	0.309
3	Ant	41	0.533
3	Flt	35	0.428

4	Pyr	35	0.423
4	BaA	39	0.496
4	Chry	32	0.382
4	BbF	35	0.434
4	BkF	62	0.962
4	BaP	25	0.987
5	IcdP	81	1.681
5	DahA	85	1.870
6	BghiP	81	1.647

On the contrary, in this study it is interesting to note that higher is the number of rings (i.e. 212 IcdP, DahA and BghiP), higher is the degradation constant, with the consequence of high 213 degradation rates. This different behavior is most probably due to the fact that degradation rates 214 in soil are governed by bacteria activity (Roslund et al., 2018), which is not present in the 215 needles where decomposition is mostly dependent on PAH chemical stability. The proposed 216 degradation model allows an estimation of the PAH contents retained by needles every year. In 217 Fig. 2, the normalized PAH contents in needle are displayed. 3 and 4 rings PAHs show a 218 decrease from 2011 to 2016, while for 2 and 5-6 rings PAHs the behavior seems not following 219 any particular trend. 220



Fig. 2. Normalized PAH content differentiated by the number of rings, in function of different

## 3.3 Correlation between adsorbed PAHs in Abies holophylla needles and air pollutant 224 225

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# emission variables in needles

PAHs sources are closely related to the anthropogenic combustion processes (e.g. car 226 exhaust, local heating facilities, industrial related activities). For this reason, it could be 227 hypothesized a correlation between adsorbed PAHs and some air pollutant emission parameters. 228 From China National Statistics Institute, it is possible to obtain data concerning: annual 229 consumption of coal (Ton year<sup>-1</sup>), crude oil (Ton year<sup>-1</sup>) and electricity (MKh year<sup>-1</sup>) 230 consumption; number of civil vehicles and trucks (Unit year<sup>-1</sup>); emission of industrial sulfur, 231 nitrogen oxide, industrial smoke (dust), and soot (Ton year-1); urban living gaseous pollutant 232 emissions (Ton year<sup>-1</sup>); generating capacity (KWh year<sup>-1</sup>). Correlation coefficients were then 233 obtained considering these parameters and PAHs average values from Abies holophylla. 234

Table 3 Correlation coefficients (R) between air pollution parameters and the average values 235 of adsorbed PAHs obtained from *Abies holophylla*. Only R > 0.8 values are displayed. 236

Air pollutant parameter	РАН	Correlation coefficient(R)
	Fluo	0.817
Coal	Phen	0.947
	Ant	0.874
	Flt	0.915
Floatricity	Pyr	0.925
Electricity	BaA	0.906
	Chry	0.947
Trucks	BbF	0.814
Industrial sulfur omissions	Fluo	0.862
	Phen	0.857
	Асу	0.869
	Ace	0.982
Industrial nitrogen evide emissions	Fluo	0.956
industrial introgen oxide emissions	Phen	0.934
	Ant	0.848
	Flt	0.908

	Pyr	0.833
Vehicle number	BaP	0.888
	Ace	0.816
Industrial gaseous pollutant emissions	Fluo	0.866
	Phen	0.864
	Flt	0.915
Generating capacity	Pyr	0.925
Generating capacity	BaA	0.911
	Chry	0.947

The concentration of PAH congener shows good correlation with meteorological 237 parameters (Table 3). For example, Fig. 3 shows a typical correlation graph obtained between 238 coal consumption per year and Ant contents in needle. Several interesting considerations can 239 be made based on Table 3. For instance, Fluo, Phen and Ant are well correlated with coal 240 consumption, and this correlation can be ascribed to their production during combustion 241 (Dameng et al., 2011). In particular, in this last process, Phen production is the highest when 242 compared to the PAHs investigated in this work, and this could lead to the very high correlation 243 value (0.947). On the other hand, these three PAHs are not the only ones produced during coal 244 combustion, but, most probably, the lack in correlation could be due to other interfering 245 processes. The correlation existing between the number of trucks and BbF can be ascribed to 246 its production by diesel engines (Kuusimäki et al., 2002). It is also worth to note the high 247 number of PAH that are correlated with industrial nitrogen oxide emissions, suggesting that 248 NOx production can be strongly related to the emission of these PAHs in the atmosphere. 249



Fig. 3. A graph showing the correlations between coal consumption per year and anthracene content in *Abies holophylla* leaves. The solid line and dotted line curves represent confidence limits for the prediction and confidence limits for the regression line at 95% confidence level, respectively.

In the urban areas, traffic emission is the dominant source of PAHs (Ratola et al., 2010b), in particular 3-4 rings PAHs: this could explain the good correlation between vehicle number and BaP. Surprisingly, none of the 5 and 6 ring PAHs shows correlation with any of the pollution sources, probably because of their fast degradation constant ( $\alpha \ge 1.65$ ) that does not allow to recover time dependent information from needles, and due to the molecule dimensions that make it more difficult to be adsorbed onto pine needles (Yang et al., 2007).

## 261 **4.** Conclusions

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In this study, the possibility to evaluate PAH historical changes and their correlation with pollution emission sources by PAH contents in the needles of *Abies holophylla* and *Pinus tabuliformis* has been investigated. Using the PAH content in *Abies holophylla* needles, a mathematical model that consider a first order PAH degradation process has been applied in order to obtain the minimum degradation per year and the minimum  $\alpha$  value for each of the considered PAH. Results indicate that higher is the number of rings (i.e. IcdP and DahA and

BghiP), higher is the degradation constant in the needles, with the consequence of high 268degradation rates. Furthermore, considering that PAHs sources are closely related to the 269 anthropogenic pollution processes, the adsorbed PAHs content calculated by the mathematical 270 model from 2011 to 2016 in Abies holophylla leaves have been statistically compared with air 271 pollutant emission parameters obtained from the databases, obtaining in some cases a good 272 correlation (R > 0.8) among data. The obtained results indicate that adsorbed PAHs in needle 273 leaves are good candidate as biomonitoring system to evaluate the historical changes of PAHs 274 induced by pollutant emission on the regional scale. 275

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## 277 **Conflicts of interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## **Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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## Highlights

- PAH contents in *Abies holophylla* and *Pinus tabuliformis* needles have been evaluated.
- A mathematical model to calculate PAH degradation in pine needles was developed.
- Degradation constants in needles increased with the number of rings in PAHs.
- Correlations between specific adsorbed PAHs in needle leaves and air pollutant sources was found.



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