IS SEMI-FLOCCULATION EFFECTIVE AS PRETREATMENT TO ULTRAFILTRATION IN WASTEWATER TREATMENT?

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ABSTRACT

In this study, ferric chloride $(FeCl_3)$ flocculation was used as a pretreatment to ultrafiltration (UF) in treating synthetic wastewater containing synthetic organic matter (SOM). The effect of flocculant dose was studied in terms of organic removal and membrane flux decline.

The UF with optimum dose of FeCl₃ (68 mg L⁻¹) did not experience any flux decline during the whole operation of 6 hours. The preflocculation with a smaller dose of 20 mg L⁻¹ of FeCl₃ led to a severe flux decline in the UF (more than 65% in 6 hours). To understand the phenomenon of the flux decline of UF, the MW ranges of SOM removed by different doses of FeCl₃ and by the post treatment of UF were studied. Flocculation with at least 50 mg L⁻¹ of FeCl₃ dose was found to be necessary to avoid any significant flux decline and to obtain superior DOC removal.

Keywords: flocculation, molecular weight distribution, pretreatment, synthetic organic matter, ultrafiltration

1. INTRODUCTION

Membrane processes such as ultrafiltration (UF), nanofiltration (NF) and reverse osmosis (RO) are successfully used in wastewater treatment, especially in water reuse applications. One of the major drawbacks of membrane processes is the membrane fouling by organic contaminants (Lin et al., 1999). Effective pretreatment is an efficient way of reducing membrane fouling.

Conventional sewage treatment includes primary treatment to remove the majority of suspended solids, secondary biological treatment to degrade the biodegradable dissolved organics and nitrogen, and tertiary to remove most of the remaining organic and inorganic solids and pathogenic microorganisms. The presence of organic pollutant in water and wastewater has been the cause of public concern in past decades due to their potential health hazard (Wang et al., 2001; Imai et al., 2002). Effluent organic matter (EfOM) in the biological treated wastewater consists of mixed particulates and soluble substance, which is combined with natural organic matter (NOM) from drinking water and soluble microbial product (SMP) from biological treatment. EfOM can thus be broadly classified into three different groups by their origins i.e. 1) refractory NOM derived from drinking water sources, 2) synthetic organic compounds (SOC) produced during domestic use and disinfection by-products (DBPs) generated during disinfection processes of water and wastewater treatment and 3) SMP derived during biological processes of wastewater treatment (Drewes and Fox, 1999).

Many researchers found that flocculation is one of the most effective pretreatment methods to remove EfOM (Abdessemed and Nezzal, 2002; Abdessemed et al., 2002; Shon et al., 2004; Kim et al., 2002). Abdessemed and Nezzal (2002) observed that flocculation with FeCl₃ removed 77% of COD in wastewater. They also found that flocculation increased the permeate flux by 46.6%. Shon et al. (2004) showed that FeCl₃ flocculation removed 68% of organic matter (in terms of DOC) from the biologically treated wastewater. The majority of organics removed were the ones with the large molecular weight (MW). Al-Malack and Anderson (1996) determined the optimum coagulation conditions for wastewater. It was 200 mg L⁻¹ FeCl₃ at a pH of 9. The COD removal with this optimum dose of FeCl₃ was 99.3%. According to the results of Aguiar et al. (1996), the optimum dose of coagulant was 2.1 ± 0.2 mg Fe per mg of total organic carbon (TOC).

It may not be economical to use large doses of flocculants. Larger chemical doses will also lead to larger quantity of chemical sludge. Therefore, it is advisable to use as smaller dose of flocculent as possible to achieve significant organic removal that can lead to minimum membrane fouling in the post treatment of membrane filtration. Thus, in this study, the effect of flocculant dose was studied in terms of organic removal and membrane flux decline. The performance of flocculation as pretreatment and UF as post treatment was also studied for their ability to remove organic matter of different MW.

2. EXPERIMENTAL

2.1 Synthetic wastewater

This study was carried out with synthetic wastewater. The composition of the synthetic wastewater is presented in Table 1. This synthetic wastewater represents effluent organic matter (EfOM) generally found in the biologically treated sewage effluent (BTSE) (Seo et al., 1997). Tannic acid, peptone, sodium lignin sulfornate, sodium lauryle sulfate and arabic acid represent larger MW portion, while peptone, beef extract and humic acid consist of organic matters of smaller MW.

2.2 Synthetic organic matter (SOM) characterization

2.2.1 Dissolved organic carbon (DOC)

DOC was measured by using the Dohrmann Phoenix 8000 UV-persulphate TOC analyzer with an autosampler. All samples were filtered through 0.45 μ m membrane prior to the DOC measurement. This is the reason the values are given in terms of DOC instead of TOC.

2.2.2 Molecular weight (MW) distribution

The synthetic wastewater after flocculation with different doses and after UF was subjected to MW distribution measurements. High pressure size exclusion chromatography (HPSEC, Shimadzu Corp., Japan) with a SEC column (Protein-pak 125, Waters Milford, USA) was used to determine the MW distributions of organic matter. Standard polystyrene sulfonates (PSS: 210, 1800, 4600, 8000, and 18000 daltons) were used to calibrate the equipment. All samples were filtered through 0.45 μ m membrane prior to the measurement. Details of the measurement methodology are given elsewhere (Her, 2002 and Cho, 1998).

2.3 Experimental set-up of partial flocculation and filtration

Figure 1 presents experimental schematic of flocculation and UF unit. The flocculation was carried out with a range of ferric chloride (FeCl₃) doses from 20 mg L^{-1} to 68 mg L^{-1} . The samples were stirred rapidly for 1 minute at 100 rpm, followed by 20 minutes of slow mixing at 30 rpm, and 30 minutes of settling. The supernatant then underwent UF.

In this study, the cross flow ultrafiltration unit (Nitto Denko, Corp.) was used to study the organic removal from synthetic wastewater. The synthetic wastewater after undergoing flocculation was pumped to the membrane module (effective membrane area 0.006 m²; flat sheet configuration). The operating pressure and cross-flow velocity were controlled at 300 kPa and 0.5 m/s by means of by-pass and regulating valves. The Reynold's number and shear stress at the wall were 735.5 and 5.33 Pa, respectively. The membrane used in this study was NTR 7410 (Nitto Denko Corp., Japan) (Table 2).

3. RESULTS AND DISCUSSION

3.1 Removal of DOC by partial FeCl₃ flocculation

The removal of synthetic organic matter (SOM) from the wastewater by ferric chloride (FeCl₃) flocculation and by the post treatment of UF was investigated in terms of DOC removal (Figure 2). DOC removal was the highest (78.2%) with the FeCl₃ flocculation at a

dose of 68 mg L⁻¹. The optimum FeCl₃ dose was found to be 68 mg L⁻¹ from the Jar test experiments. Experiments were also conducted with reduced concentrations of FeCl₃ (semi optimum doses) followed by UF in order to study the effect of semi flocculation on DOC removal. For example, the DOC removal from wastewater was 87.8% with preflocculation (68 mg L⁻¹ of FeCl₃) and post UF application. The removal efficiency reduced when lower doses of FeCl₃ were employed. Although the semi flocculation (with reduced FeCl₃ doses) led to less pretreatment removal efficiency, the post treatment of UF compensated the total net removal i.e. 82.8% (with 40 mg L⁻¹ of FeCl₃) and 82.2% (with 20 mg L⁻¹ of FeCl₃). In other words, the post treatment of UF after 68 mg L⁻¹ FeCl₃ flocculation removed 48.2% of additional DOC removal. Thus, the semi flocculation as pretreatment delegated the majority of organic removal to UF.

3.2 Flux decline of UF with pretreated wastewater

The flux decline of UF in treating synthetic wastewater was studied in terms of normalized permeate flux (J/J_0) . It was studied both with and without the pretreatment of FeCl₃ flocculation at different doses (Figure 3). The flux decline (J/J_0) with no pretreatment was at 0.68 after 6 hours of UF operation. After the pretreatment of flocculation with the optimum dose of FeCl₃ (68 mg L⁻¹), the UF did not experience any flux decline. It may be due to the removal of the majority of SOM by flocculation and complexation (Shon et al., 2004a). However, the preflocculation with partial optimum dose of FeCl₃ led to significant flux decline. For example, the preflocculation with 20 mg

 L^{-1} of FeCl₃ led to a flux decline (J/J₀) in the post treatment of UF of 65% in 6 hours. This indicates that flocculant dose should be sufficient enough to avoid or minimize the flux decline.

3.3 Molecular weight (MW) distribution

To understand the phenomenon of the flux decline, it is necessary to know the types of organics and/or the range of MW distribution of SOM removed from the wastewater by flocculation and ultrafiltration. This will help in the selection of suitable pretreatment method and a correct membrane for the given application (Tandanier et al., 2000). Synthetic wastewater has a number of known compounds at a known concentration. Hence, the MW distribution of each component in SOM in synthetic wastewater was first analyzed (Figure 4). The MW of the mixed synthetic wastewater ranged from 291 daltons to about 34118 with the highest fraction of 943 – 1196 daltons. Although sodium lignin sulfornate and tannic acid showed the peaks at 12120 and 6343 daltons respectively, the corresponding peaks were not found in the mixed synthetic wastewater (Figure 4 (c)). It may be due to aggregations between SOM and inorganic and/or organic compounds in the synthetic wastewater.

Figure 5 describes the MW distribution of UF effluent without any pretreatment and with preflocculation (using FeCl₃). It should be noted that the settled flocs were removed after FeCl₃ flocculation and only the supernatant underwent ultrafiltration. Flocculation with larger doses (closer to the optimum dose) from 68 mg L^{-1} to 50 mg L^{-1} FeCl₃ removed

practically all the large MW SOM such as tannic acid, sodium lignin sulfornate, sodium lauryle sulfate and arabic acid. The FeCl₃ dose of 40 mg L⁻¹ removed the majority of large MW SOM, but not all of them. Further, this preflocculation was also helpful to remove some of the small MW compounds (573 - 1002 daltons) such as peptone, beef extract and humic acid. However, the smallest MW range of compounds in the range of 248 daltons could not be removed by flocculation. The phenomenon of the small MW organic matter removal (573 daltons to 1002) by FeCl₃ flocculation may be due to the complexation of Fe at wide range of pH (Vilge-Ritter et al., 1999). FeCl₃ dose of 40 mg L⁻¹ and less did not remove all the large MW compounds and the majority of small MW compounds. The post treatment of UF with the pretreatment of flocculation removed practically the organic compounds of more than 1000 daltons.

Table 3 presents the weight-averaged MW of the compounds in the supernatant after different FeCl₃ doses. The weight-averaged MW value of the influent was 29759 daltons, which was similar to the effluent after flocculation with 20 mg L^{-1} (around 29594 daltons).

A correlation between the amount of FeCl_3 dose and the averaged-weight MW is presented in Figure 6. The deviant crease circle shows that the pretreatment of flocculation with reduced FeCl_3 dose (less than the optimum dose) is possible as an adequate pretreatment to minimize the flux decline and to obtain high DOC removal. In the present study, a dose of 50 mg L⁻¹ of FeCl_3 was sufficient to run the UF with no (or minimum) flux decline and higher DOC removal. Here, it should be noted that the above finding and the quantitative values are true only for the synthetic wastewater used, as wastewater characteristics differ from area to area and from season to season. As such, one needs to conduct experiments with particular wastewater at the time of operation to determine the suitable dose of FeCl₃.

CONCLUSIONS

A detailed experimental study was conducted to investigate the capability of semi flocculation as a pretreatment to UF. The organic matter (DOC) removal and the flux decline in the post treatment of UF were examined. These results were related to MW distribution of the effluent. The results obtained lead to the following conclusions.

1. Pretreatment of flocculation (with optimum FeCl_3 dose of 68 mg L⁻¹) removed 75% of DOC, which led to only 9.6% of additional DOC removal by the UF as post treatment. On the other hand, a partial FeCl_3 dose of 20 mg L⁻¹ removed only 34% of DOC and the UF removed another 48%.

2. The flux decline (in terms of J/J_0 in the UF after 6 hours of operation) with no pretreatment was 0.68. On the other hand, the UF with the preflocculation with the optimum dose of FeCl₃ (68 mg L⁻¹) did not experience any flux decline during the operation of 6 hours. The preflocculation with sub-optimal doses of FeCl₃ of 20 – 30 mg L⁻¹ led to a significant flux decline, whereas a dose of 40 – 50 mg L⁻¹ of FeCl₃ showed only a minimum flux decline.

3. The peaks corresponding to larger MW (36258 daltons) were not observed in the flocculated effluent with a FeCl₃ dose of 50 mg L^{-1} and more. The effluent after flocculation with FeCl₃ of less than 40 mg L^{-1} showed peaks corresponding to large MW.

4. The weight-averaged MW values of the compounds in the effluent after flocculation with more than 50 mg L^{-1} FeCl₃ was much lower (less than 700 daltons) as compared to the one with less than 40 mg L^{-1} FeCl₃ (around 29000 daltons which is in the similar range of the influent).

The results showed that a FeCl_3 dose of 50 mg L⁻¹ was necessary to minimize the flux decline in the UF and to obtain higher DOC removal.

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Table 1 Constituents of the synthetic wastewater

Table 2 Characteristics of UF membrane used

Table 3 Weight-averaged MW values of the effluent samples after pretreatment (weight-

averaged MW of initial = 29759 daltons)

Table 1	Constituents	of the	synthetic	wastewater
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Commound	Concentration	Main molecular	Fraction by DOC	
Compound	(mg L ⁻¹)	weight (Daltons)		
Beef extract	1.8	298, 145, 65	0.065	
Peptone	2.7	34265, 128, 80	0.138	
Humic acid	4.2	1543, 298	0.082	
Tannic acid	4.2	6343	0.237	
Sodium lignin sulfonate	2.4	12120	0.067	
Sodium lauryle sulphate	0.94	34265	0.042	
Arabic gum powder	4.7	925, 256	0.213	
Arabic acid (polysaccharide)	5.0	38935	0.156	
(NH ₄) ₂ SO ₄	7.1		0	
K ₂ HPO ₄	7.0		0	
NH ₄ HCO ₃	19.8		0	
MgSO ₄ •7H ₂ O	0.71		0	

Table 2 Characteristics of UF membrane	used
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Code	Material	$MWCO^*$	Contact	Zeta potential	PWP ^{**} at 300	R_m (membrane resistance,
		(daltons)	angle(°)	at pH 7 (mV)	kPa (m/d)	$x 10^{12} m^{-1}$)
NTR 7410	Sulfonated polysulfones	17,500	69	-98.63	1.84	14.1

* MWCO: molecular weight cut off

** PWP: pure water permeability

Table 3 Weight-averaged MW values of the effluent samples after pretreatment (weightaveraged MW of initial = 29759 daltons)

		FeCl ₃ concentration					
	68 mg L ⁻¹	60 mg L ⁻¹	$50 \text{ mg } \text{L}^{-1}$	40 mg L ⁻¹	$30 \text{ mg } \text{L}^{-1}$	20 mg L ⁻¹	
MW	517	584	685	26226	29324	29594	

All units: Daltons

Figure 1 Schematic drawing of partial flocculation and UF unit

Figure 2 DOC removal (a) by semi flocculation and (b) by semi flocculation followed by UF

Figure 3 Temporal variation of filtration flux of UF after a pretreatment of flocculation at different FeCl₃ doses (NTR 7410 UF membranes, $J_0 = 1.84$ m/d at 300 kPa; crossflow velocity = 0.5 m/s; MWCO of 17,500 daltons; Reynold's number: 735.5; shear stress: 5.33 Pa)

Figure 4 MW distribution of SOM in the synthetic wastewater (a), b): individual components in the wastewater; c) wastewater (with all compounds mixed together) Figure 5 MW distribution of the effluent of flocculation ($J_0 = 1.84$ m/d at 300 kPa; crossflow velocity = 0.5 m/s; MWCO = 17,500 daltons; Reynold's number.: 735.5; shear stress: 5.33 Pa; a) MW distribution of SOM with higher doses of FeCl₃ (50 - 68 m/L); b) with FeCl₃ of lower doses (40 - 20 mg L⁻¹ flocculation); c) flocculation followed by UF Figure 6 Correlation between the FeCl₃ concentrations and the corresponding weight-averaged MW values in the flocculated effluent.



Figure 1 Schematic drawing of partial flocculation and UF unit



Figure 2 DOC removal by semi flocculation followed by UF



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