- Evaluating the sustainability of free water surface flow constructed wetlands:
- 2 methane and nitrous oxide emissions
- 3 Haiming Wu^{a,b}, Jian Zhang^{b*}, Huu Hao Ngo^c, Wenshan Guo^c, Shuang Liang^b
- 4 College of Natural Resources and Environment, Northwest A&F University, Yangling,
- 5 Shaanxi 712100, China
- ^b Shandong Key Laboratory of Water Pollution Control and Resource Reuse, School
- of Environmental Science & Engineering, Shandong University, Jinan 250100, PR
- 8 China
- ⁹ School of Civil and Environmental Engineering, University of Technology Sydney,
- 10 Broadway, NSW 2007, Australia

11 **Abstract**

Constructed wetlands (CWs) have been used as a green technology to treat various 12 wastewaters for several decades, and greenhouse gases production in these systems 13 14 attracted increasing attention considering the contributions of methane and nitrous 15 oxide emissions to global warming. However, the detailed knowledge about the 16 contribution of CWs to methane and nitrous oxide emissions in treating sewage treatment plant effluent are still limited in particular for a better understanding of the 17 18 sustainability of CWs. The fluxes of methane (CH₄) and nitrous oxide (N₂O) from free 19 water surface (FWS) CWs in northern China were measured continuously using the static-stationary chamber technique from 2012 to 2013. The results showed that CWs 20 21 were the significant source of CH₄ and N₂O emissions. Average emission rates of CH₄ and N_2O ranged from -30.2 μ g m⁻² h⁻¹ to 450.9 μ g m⁻² h⁻¹, and -58.8 μ g m⁻² h⁻¹ to 22 1251.8 µg m⁻² h⁻¹, respectively. Obvious annual and seasonal variations of CH₄ and 23 N₂O emissions were observed over the 2-year period. In addition, temperatures and 24

- 25 plant species had an impact on CH₄ and N₂O emissions. The obtained results showed
- that FWS CWs, improving water quality but emitting lower CH₄ and N₂O, could be the
- 27 alternative method for sewage treatment plant effluent.
- 28 **Keywords:** Constructed wetlands; Methane; Nitrous oxide; Wastewater treatment

1. Introduction

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- 30 Over the last few decades, point and non-point pollution from agricultural, fishing,
- municipal and industrial drainage has become a worldwide environmental issue,
- 32 especially in developing countries (Wu et al., 2015). On the one hand, untreated
- wastewater is directly discharged into continental surface waters because large scale
- municipal wastewater treatment plants (WWTPs) have not been constructed or fully
- operated due to large capital investments and operating costs in rural areas.
- Furthermore, considering the stringent discharge guidelines and standards,
- 37 conventional wastewater treatment processes fail to remove large amount of nutrients
- 38 efficiently, and are also not specifically designed to eliminate micropollutants (Luo et
- 39 al., 2014; Kong et al., 2015; Lu et al., 2016; Wu et al., 2016). Consequently, untreated
- 40 wastewater and sewage effluent which contain a variety of excessive organics and
- 41 nutrients are discharged into rivers, estuaries and oceans, and may deteriorate the
- 42 water environment quality and impact aquatic ecosystem health. Thus, the potential
- 43 cost-effective treatment technologies of sewage/wastewater have been partially
- investigated in previous studies (Wu et al., 2011; Huang et al., 2013; Eveborn et al.,
- 45 2014; Pan et al., 2016).
- 46 In recent years, constructed wetlands (CWs), as a green wastewater treatment
- 47 technology by simulating natural wetlands, have been proven to be an effective
- 48 alternative for conventional wastewater treatment technologies owing to their lower

49	cost, less operation and maintenance requirements, and little reliance on energy
50	inputs (Vymazal, 2011; Wu et al., 2015). CWs are generally comprised of vegetation,
51	substrates, soils, microorganisms and water, and have been found to be able to
52	remove various pollutants (e.g., organics, nutrients and micropollutants) from
53	wastewater by utilizing a variety of physical, chemical, and biological mechanisms
54	(microbial degradation, plant uptake, sorption, sedimentation, filtration and
55	precipitation etc.) (Vymazal, 2011; Saeed and Sun, 2012; Wu et al., 2015). Such
56	natural-like systems can be mainly divided into free water surface (FWS) and
57	subsurface flow (SSF) CWs, and are usually used to treat different wastewaters such
58	as domestic sewage, industrial drainage, urban and agricultural, stormwater runoff,
59	animal wastewaters, leachates, mine drainage and polluted river water (Rai et al.,
60	2013; Li et al., 2014; Vymazal, 2014; Greenway, 2015; Saumya et al., 2015). In
61	addition, CWs might be utilized as a supplement to the existing conventional WWTPs
62	for reclaiming and reusing the sewage effluent (Greenway, 2005; Rai et al., 2013).
63	However, with the aim of improving the water quality and conserving aquatic
64	ecosystem, little attention has been paid to purification of WWTPs effluent which was
65	characterized by relatively low organic content and moderate nitrogen and
66	phosphorous concentrations. Meanwhile, as an artificial ecological system simulating
67	natural wetlands, greenhouse gases (GHG) production in these systems attracted
68	increasing attention considering the contributions of methane (CH ₄) and nitrous oxide
69	(N_2O) emissions to global warming (Kong et al., 2016). Compared to natural wetlands,
70	heavy nutrient loading to CWs stimulates bacterial processing, resulting in higher
71	fluxes of CH_4 and N_2O , and thus CWs might be significant sources of CH_4 and N_2O
72	emissions. Many studies investigated the emission of CH_4 and N_2O in various types of
73	CWs for treating various kinds of wastewaters (such as domestic wastewater, dairy

- farm wastewater, municipal wastewater and mining runoff) based on the lab-scale and
- full-scale experiments (Tanner et al., 1997; Mander et al., 2008; Van der Zaag et al.,
- 2010; Mander et al., 2014). From the current literature review by Mander et al. (2014),
- it indicated that average values of CH₄ and N₂O emissions in various types of CWs
- are 97-142 mg m⁻² h⁻¹ and 2.2-3.1 mg m⁻² h⁻¹, and can be influenced by various
- 79 physical, hydrological and operational factors such as dissolved oxygen (DO),
- 80 hydraulic retention time (HRT), water depth, inflow loading, influent C/N ratio, climate
- and vegetation. Therefore, in order to comprehensively evaluate the environmental
- benefit of using CWs as a sustainable wastewater treatment technology, a continuous
- measurement of CH₄ and N₂O emissions in treatment processes from CWs is
- absolutely necessary. Moreover, the detailed knowledge about the contribution of
- 85 CWs to CH₄ and N₂O emissions in treating WWTPs effluent would be required in
- particular for the potential of GHG mitigation.
- The aim of this work was to quantify the long-term CH₄ and N₂O emissions from FWS
- 88 CWs for treating sewage treatment plant effluent. Annual and seasonal variations of
- 89 CH₄ and N₂O emissions were analyzed over an approximate 2-year period. The CH₄
- and N₂O fluxes and their global warming potential were further comparatively
- compared with common CW treatments and current WWTPs.

92 **2. Material and methods**

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- 2.1 Experimental system and operation
- 94 Experimental FWS CW systems which were built in Baihua Park in Jinan, northern
- 95 China (36°40 '36"N, 117°03 '42"E) were designed to treat the effluent of sewage
- 96 treatment plant (Figure 1). The climate of the area is characterized by a
- 97 warm-temperature monsoonal climate. The experimental treatment system consisted

of twelve FWS CW systems with a surface area of approximately 0.13 m² (50 cm in depth and 40 cm in diameter), and each system had an outlet at the bottom. All CW system were filled with washed river sand (particle size <2 mm, 0.39 porosity) as the substrate with a depth of 25 cm. Nine of CW systems were planted with three macrophyte species (W1: *Phragmites australis*, W2: *Cyperus rotundus*, W3: *Zizania caduciflora*) with three duplicates, and three of CW systems were not planted (U4: control). The density of plants was 12, 20 and 20 rhizomes per system for W1, W2 and W3, respectively. Each system held 20 L water when filled. The water depth of each system was approximately 10 cm from the sand surface.

All experimental CW systems were operated for a period of approximate two years (from April 2012 to December 2013). The synthetic sewage treatment plant effluent was used as influent in each wetland in this study, The synthetic wastewater was prepared from tap water and mainly composed of sucrose, (NH₄)₂SO₄, KH₂PO₄ and

(from April 2012 to December 2013). The synthetic sewage treatment plant effluent was used as influent in each wetland in this study, The synthetic wastewater was prepared from tap water and mainly composed of sucrose, (NH₄)₂SO₄, KH₂PO₄ and KNO₃ based on Grade I treatment standard of municipal sewage treatment plants in China (Wu et al., 2011). Specially, the characteristics of the influents in the present study were COD 72.71 mg L⁻¹, NH₄⁺-N 8.36 mg L⁻¹, TN 21.14 mg L⁻¹ and TP 8.36 mg L⁻¹, respectively. Sequencing fills-and-draw batch mode was applied to influent in the whole experimental period. The HRT was 10 d from April to November and 15 d in November and March when temperature was low.

2.2 Sampling and analysis

2.2.1 Environmental parameters

The following environmental parameters and climatic data in the experimental site
were recorded: air temperature (°C) and relative humidity (%).

2.2.2 Water sampling and analysis

Water samples of influent and effluent were taken to evaluate their treatment performance. According to standard methods (APHA, 2005), and all samples were transferred immediately to the lab and analyzed immediately for the following water physicochemical parameters: chemical oxygen demand (COD; HACH DR 2008TM Spectrophotometer, USA), ammonia nitrogen (NH₄⁺-N), total nitrogen (TN) and total phosphorus (TP). Dissolved oxygen (DO) and pH were measured in situ by a DO meter (HQ 30d 53LEDTM HACH, USA) and a glass pH meter (SG2-T SevenGo pro TM MTD, Switzerland).

2.2.3 Gas sampling and analysis

CH₄ and N₂O fluxes from the FWS CWs have been investigated in this study. Gas sampling was done using the static-stationary chamber every two days during the whole experimental period. The transparent chamber system (50 cm \times 50 cm \times 50 cm \times 50 cm)was made of polymethyl methacrylate, and the details of collecting steps of gas samples were according to the method described in the previous studies (Wu et al., 2009). The N₂O concentration was determined using the gas chromatography (SP-3410, China) with an electron capture detector (ECD) and a Poropak Q column, using 30 mL/min high-purity nitrogen as the carrier gas. The temperature of the detector and column were set at 36 °C and 50 °C, respectively. The CH₄ concentration was determined using the gas chromatography (SP-6890, China) equipped with a flame ionization detector (GC-FID) and stainless steel packed columns (GDX502). The operating conditions for the GC were: 375 °C reformer temperature, 40 °C oven temperature and 200 °C detector temperature. The carrier gas was ultra-high purity N₂ (30 mL min⁻¹). CH₄ and N₂O fluxes (μ g m⁻² h⁻¹) were

- determined from the increase in concentration in the chambers over time with linear regression analysis according to the method described in the previous studies (Wu et al., 2009).
- 148 2.3 Statistical analysis
- Statistical analyses were performed through the software SPSS 11.0 (SPSS Inc.,
- 150 Chicago, USA). A two independent samples t-test was conducted to determine the
- significance of differences between means. In all tests, differences and correlations
- were considered statistically significant when P < 0.05.

3. Results and discussion

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3.1 Environmental variables and water characteristics

As shown in Figure 2, monthly mean air temperature during the whole monitoring period ranged from 2.1 °C to 29.5 °C, and the annual average air temperature in 2012 was 21.4 °C, which was slightly higher than that in 2013 (19.1 °C). Moreover, the maximum temperature was observed to appear from May to August, and the minimum temperature was recorded in January and February. The average relative humidity during the study period was 57.4%, with the higher value in the first year (59.6%) and the lower value in the second year (55.6%). The average effluent concentrations of COD, NH₄+-N, TN and TP in different FWS CW systems in the present study were 16.7-24.6 mg L⁻¹, 0.5-4.4 mg L⁻¹, 3.2-11.7 mg L⁻¹ and 0.5-1.1 mg L⁻¹, respectively. These results indicated a significant improvement in water quality of sewage treatment plant effluent by treatment through FWS CWs. However, the average removal performance of the planted FWS CW systems was higher than that of unplanted CW systems, which suggested that there was a positive correlation

among water purification and plant growth and establishment. On the whole, our results are consistent with other research which reported that reduction of pollutants was found to increase with growth and establishment of the plants (Rai et al., 2013).

3.2 Variation of CH₄ emission

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The variation of CH₄ emission from different FWS CW systems in 2012-2013 is shown in Figure 3a. Average CH₄ fluxes had obvious annual and seasonal variations in different FWS CWs, and ranged from -30.2 µg m⁻² h⁻¹ to 450.9 µg m⁻² h⁻¹. Specially, the average CH₄ flux in CW systems in the second year (138.6 µg m⁻² h⁻¹) was significantly higher than that (88.6 µg m⁻² h⁻¹) measured in the first year. The higher flux of CH₄ was observed in summer compared to spring and fall, and a general seasonal peak occurred at the end of the summer/beginning of fall. However, it should be noted that the weak absorption (the sink) of CH₄ was found in cold season (November and December) compared with other period when wetland became a source of CH₄. These results suggested that seasons might have a significant effect on CH₄ emissions in FWS CW systems. These results can also be illustrated by the line regression relationship between CH₄ fluxes and air temperature. As shown in Figure 3b, during the 2-year monitoring period, the rate of CH₄ emission in CW systems was significantly associated with air temperatures, and CH₄ flux generally increased with the temperature rising. However, clear difference was found between vegetation and non-vegetation systems. The possible reason may be that microbial activity in CWs would increase with the increasing of temperature at a certain climatic condition (Wu et al., 2011; Mander et al., 2014). On the other hand, it is well recognized that temperature affects plant photosynthesis and plant biomass directly. and thus increases organic matter and gas transportation, which would give a positive

192	effect on CH ₄ emission (Zhao et al., 2016). However, some other studies have
193	reported negative effect of plant biomass on CH ₄ emission, and these impacts on CH ₄
194	emission from CWs vary significantly among plant species (Bhullar et al., 2014).
195	Specifically, significant difference of CH ₄ emission rate was found among different
196	CWs with various plant species in this study. The wetlands (W1) vegetated with
197	Phragmites australis emitted higher CH ₄ (164.1 µg m ⁻² h ⁻¹) than wetlands (W3) planted
198	with Zizania caduciflora (152.1 μg m ⁻² h ⁻¹), following by and the wetlands (W2) with
199	Cyperus rotundus (104.5 µg m ⁻² h ⁻¹) throughout experimental period. Similarly, Zhao
200	et al. (2016) studied the effects of plant diversity on CH ₄ emission and nitrogen
201	removal, and concluded that the best combination of low CH ₄ emission and high N
202	removal rates could be achieved in CW microcosms planting P. arundinacea. These
203	results about difference among various plant species suggested that CH ₄ emission
204	could be affected by other various factors such as CW type, oxygen level, inflow
205	loading and C/N ratio (Mander et al., 2014; Zhao et al., 2016).
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206	On the whole, analysis of CH ₄ emission in two years showed FWS CW systems were
207	a CH ₄ source, but the mean CH ₄ emission rate measured in this study were lower
208	than the values (0.2-36 mg m ⁻² h ⁻¹) in FWS CW treatment systems and the values
209	$(0.064-23~\text{mg m}^{-2}~\text{h}^{-1})$ in SSF CWs reported in the literature (Mander et al., 2014), and
210	as high as the values (0.003-6.2 mg m ⁻² h ⁻¹) in enhancing CW systems such as
211	aerated CWs (Maltais-Landry et al., 2009). CH ₄ emission was also found to be
212	significantly lower than the results (0.61-9.7 mg m ⁻² h ⁻¹) from natural wetlands (Chen
213	et al., 2013). Moreover, when compared with common WWTPs, the emission rate was
214	greatly lower than the values (0.06-978 g m ⁻² d ⁻¹) obtained among different processing
215	units in the typical conventional WWTPs (Ren et al. 2015).

3.3 Variation of N₂O emission

The variation of N ₂ O emission from FWS CW systems during the experimental period
is illustrated in Figure 4a. It is shown that $N_2\text{O}$ emission from the planted and
unplanted wetlands varied annually and seasonally. The N ₂ O emission rate ranged
from -58.8 $\mu g~m^{\text{-}2}~h^{\text{-}1}$ to 1251.8 $\mu g~m^{\text{-}2}~h^{\text{-}1},$ and specially, the average N_2O flux in CW
systems in 2013 (381.8 $\mu g \ m^{-2} \ h^{-1}$) was significantly higher than that (311.1 $\mu g \ m^{-2} \ h^{-1}$)
recorded in 2012. This result indicated that the wetlands plants were well developed in
the second year, and flourishing plants and active microbial population beneficial to
nitrification and denitrification would promote production and transport of N ₂ O as
compared with in the initial first year. The N ₂ O emission was also observed to be
higher in summer than in spring and fall, and there had a lower N ₂ O emission rate in
winter due to the plants withering and microorganism activity decreasing. This result
indicated that temperature had an important effect on the $\ensuremath{\text{N}}_2\ensuremath{\text{O}}$ emission in CWs, which
is in agreement with other reports (Zhang et al. 2005). Figure 4b presents the
polynomial regression relationship between N ₂ O emission rates and air temperature
during the 2-year operating period. It can be illustrated that the $N_2\text{O}$ emission rate in
CWs was increased with the rising temperatures, but the statistics was not significant.
Mander et al. (2014) reported that the higher temperature of the environment slightly
increased CH ₄ emission in CWs, whereas in terms of N ₂ O emission the relationship
was insignificant and unclear.
The average N ₂ O emission rate from the different wetlands in the whole experiment
also varied each other. The planted wetlands had higher N ₂ O emission rate than the
unplanted wetlands, which indicated that the plant species had an impact on $N_2\mbox{O}$
emission, N ₂ O emission rates from the planted wetlands also varied among plant

species because of the relative differences in intrinsic species, possible ecotype and growth characteristics. On the whole, wetlands (W1) vegetated with *Phragmites* australis had highest N₂O fluxes (mean value 514.2 µg m⁻² h⁻¹) following by wetlands (W3) planted with Zizania caduciflora (mean value 334.9 µg m⁻² h⁻¹) and the wetlands (W2) with *Cyperus rotundus* (mean value 328.2 µg m⁻² h⁻¹). The results suggested that FWS CW systems were a source of N₂O throughout experimental period when treating sewage treatment plant effluent. The maximum N₂O emission rates measured in this study were higher than the values (50 µg m⁻² h⁻¹) in natural ecosystems reported by Saggar et al. (2007), but lower than the values measured in CWs treating wastewater (2145 µg m⁻² h⁻¹) reported by Wu et al. (2009), and still greatly lower than the values (2×10³ mg m⁻² h⁻¹) measured in sewage treatment plants (Benckiser et al. 1996).

4. Conclusions

In this study, CH_4 and N_2O emissions from FWS CWs treating sewage treatment plant effluent, ranging from -30.2 μ g m⁻² h⁻¹ to 450.9 μ g m⁻² h⁻¹, and -58.8 μ g m⁻² h⁻¹ to 1251.8 μ g m⁻² h⁻¹, respectively, were significantly low compared with traditional WWTPs, but obvious temporal variations were found in different FWS CWs. The average emission rates of CH_4 and N_2O in CWs in the second year (138.6 μ g m⁻² h⁻¹ and 381.8 μ g m⁻² h⁻¹) were significantly higher than that in the first year (88.6 μ g m⁻² h⁻¹ and 311.1 μ g m⁻² h⁻¹), and the higher fluxes of CH_4 and N_2O were observed in summer compared to spring and fall. The rate of CH_4 and N_2O emission was increased as the temperature rising, and the planted wetlands had higher CH_4 and N_2O emission than the unplanted wetlands. These results showed that FWS CWs, achieving the better treatment performance and lower GHG emission, could be an

- alternative method for sewage treatment plant effluent, which would be beneficial for
- the sustainable operation and successful application of CW systems.

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374	

375	Figure Captions:
376	Figure 1 Figure 1 Profile of the laboratory-scale constructed wetland (a) and
377	photograph of the experimental constructed wetland systems (b)
378	Figure 2 The variation of air temperature and relative humidity during the experimenta
379	period.
380	Figure 3 The variation of CH₄ emissions from different wetland systems (W1:
381	Phragmites australis, W2: Cyperus rotundus, W3: Zizania caduciflora, W4:
382	unplanted) during the experimental period (a), and linear regression between
383	air temperature and CH₄ emission rates (b).
384	Figure 4 The variation of N₂O emissions from different wetland systems (W1:
385	Phragmites australis, W2: Cyperus rotundus, W3: Zizania caduciflora, W4:
386	unplanted) during the experimental period (a), and polynomial regression
387	between air temperature and $N_2\text{O}$ emission rates (b).

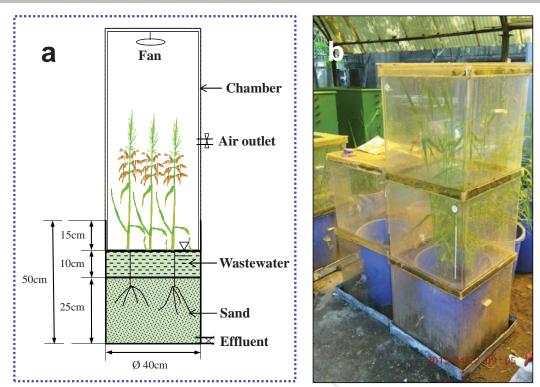


Figure 1 Profile of the laboratory-scale constructed wetland (a) and photograph of the experimental constructed wetland systems (b)

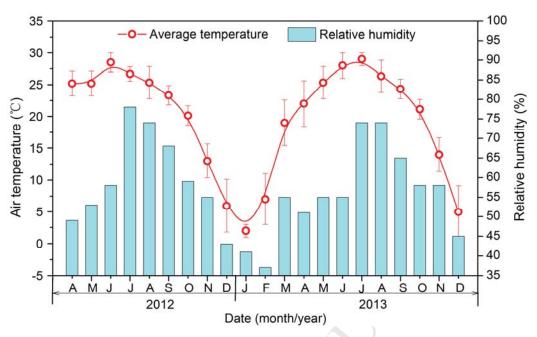


Figure 2 The variation of air temperature and relative humidity during the experimental period.

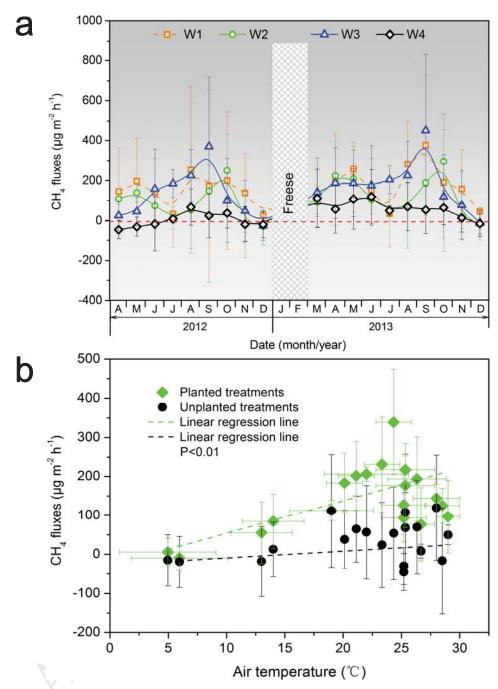


Figure 3 The variation of CH₄ emissions from different wetland systems (W1: *Phragmites australis*, W2: *Cyperus rotundus*, W3: *Zizania caduciflora*, W4: unplanted) during the experimental period (a), and linear regression between air temperature and CH₄ emission rates (b).

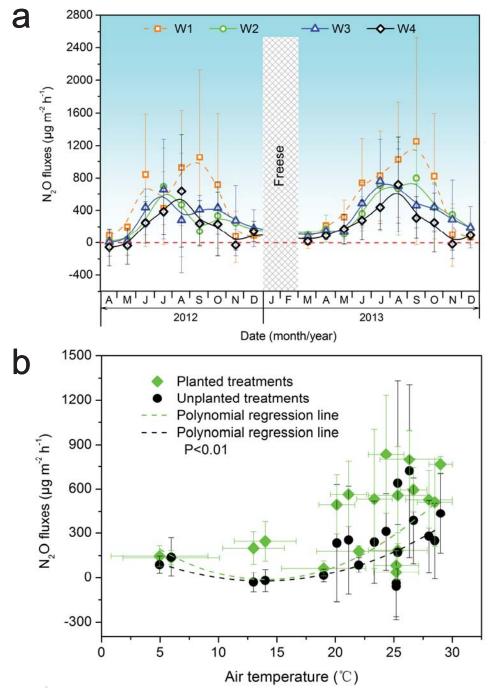


Figure 4 The variation of N_2O emissions from different wetland systems (W1: *Phragmites australis*, W2: *Cyperus rotundus*, W3: *Zizania caduciflora*, W4: unplanted) during the experimental period (a), and polynomial regression between air temperature and N_2O emission rates (b).

Research Highlights

- FWS CWs were used to treat sewage treatment plant effluent for about two years.
- 2) Annual and seasonal variations of CH₄ and N₂O fluxes from FWS CWs were observed.
- 3) FWS CWs might be the significant source of CH₄ and N₂O.
- 4) Temperatures and plant species had an impact on CH₄ and N₂O emission.